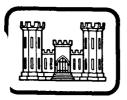
ARMY ENGINEER WATERWAYS EXPERIMENT STATION VICKSBURG--ETC F/G 6/6
FIELD SURVEY OF HEAVY METAL UPTAKE BY NATURALLY OCCURRING SALTM--ETC(U)
JUN 81 J W SIMMERS. B L FOLSOM, C R LEE
MES/TRZEL-01-5 NL AD-A101 662 UNCLASSIFIED JOF 2 11 111 ø ينه



CV 

Ċ

-

B



**TECHNICAL REPORT EL-81-5** 

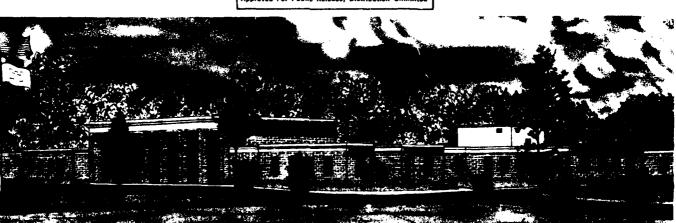
## FIELD SURVEY OF HEAVY METAL UPTAKE BY NATURALLY OCCURRING SALTWATER AND FRESHWATER MARSH PLANTS

John W. Simmers, Bobby L. Folsom, Jr., Charles R. Lee, and Derrick J. Bates

Environmental Laboratory U. S. Army Engineer Waterways Experiment Station P. O. Box 631, Vicksburg, Miss. 39180

> June 1981 Final Report

Approved For Public Release; Distribution Unlimited



Prepared for Office, Chief of Engineers, U. S. Army Washington, D. C. 20314

Under Dredging Operations Technical Support Program

81 7 21 039

Destroy this report when no longer needed. Do not return it to the originator.

The findings in this report are not to be construed as an official Department of the Army position unless so designated.

by other authorized documents.

The contents of this report are not to be used for advertising, publication, or promotional purposes. Citation of trade names does not constitute an official endorsement or approval of the use of such commercial products.

REPORT NUMBER	TATION PAGE		READ INSTRUCTIONS  JEFORE COMPLETING FORM
Technical Report EL-81-5	AD-A1C	1 60	3. RECIPIENTIS CATALOG NUMBER
TITLE (and Substitut) FIELD SURVEY OF HEAVY METAL DOCCURRING SALTWATER AND FREE			Final report. 1/7/
John W./Simmers, Bobby L/Fo Charles R., Lee Derrick			8. CONTRACT OR GRANT NUMBER(*)
PERFORMING ORGANIZATION NAME AND U.S. Army Engineer Waterway Environmental Laboratory P.O. Box 631, Vicksburg, Mi	ys Experiment Sta	ation	10. PROGRAM ELEMENT PROJECT, TAS AREA & WORK UNIT NUMBERS Dredging Operations Technical Support Program
. CONTROLLING OFFICE NAME AND ADD Office, Chief of Engineers, Washington, D. C. 20314		(11	12. REPORT DATE, June 1981 15. NUMBER OF PAGES
			161
MONITORING AGENCY NAME & ADDRESS	166	Ma Office)	15. SECURITY CLASS. (of this report) Unclassified
(12	1)166	ļ	15a. DECLASSIFICATION/DOWNGRADING SCHEDULE
. DISTRIBUTION STATEMENT (of the abetr	act entered in Block 20, if	different from	Report)
. supplementary notes Available from National Tech	hnical Informatio	on Servi	ce, Springfield, Va. 2216
. KEY WORDS (Continue on reverse side if n	Marshes Plants (Botany) Sediment	ock number)	
Field investigations Great Lakes Gulf Coast Heavy metals			
Great Lakes			

DD 1 1473 1473 EDITION OF 1 NOV 65 IS OBSOLETE

SECURITY CLASSIFICATION OF THIS PAGE (When Data Brite

411388

1

### SECURITY CLASSIFICATION OF THIS PAGE(When Date Entered

20. ABSTRACT (Continued).

species were collected from the freshwater marsh areas.

Results indicated that, in the saltwater marsh, manganese and zinc concentrations were slightly lower than those previously determined in greenhouse and disposal site studies with contaminated sediments, while the concentrations of the remainder of heavy metals were similar. Calculation of total uptake values indicated that only cadmium may be of concern.

Cadmium levels in *Cyperus* species in naturally occurring marshes were similar to those of a greenhouse flooded (reduced) environment. Iron and manganese were generally present in lower concentration in the natural marsh than in the greenhouse plants grown on contaminated sediment, while the remainder of the heavy metals examined were present in higher concentrations.

This investigation forms the basis for the conclusion that marsh plants grown on contaminated dredged material in a flooded environment do not bioconcentrate excessive or even significantly higher levels of toxic metals than those same plant species in naturally occurring marshes.

ACCESSION FOR NTIS GRA&I	4
Unannounced Justification  By Distribution/ Availability Code:	
Dist Epocial	

#### PREFACE

This investigation was conducted by the Environmental Laboratory (EL), U. S. Army Engineer Waterways Experiment Station (WES), Vicksburg, Miss., during 1978-1980 by Drs. John W. Simmers, Bobby L. Folsom, Jr., and Charles R. Lee and Messrs. Derrick J. Bates, Thomas C. Sturgis, Dewey S. Sandberg, Roger D. Brock, and R. Glenn Rhett of the Contaminant Mobility Research Team, under the general supervision of Dr. R. M. Engler, Chief, Ecological Effects and Regulatory Criteria Group, Dr. R. L. Eley, Chief, Ecosystem Research and Simulation Division (ERSD), and Dr. John Harrison, Chief, EL. Assistance was also received from several other members of the ERSD. Funding for the study was provided by the Dredging Operations Technical Support (DOTS) Program, Mr. C. C. Calhoun, Jr., Program Manager.

Dr. C. B. Loadholt, Professor of Biometrics, Medical College of South Carolina, assisted the authors on statistical matters.

This report is the fourth in a group of reports addressing bioconcentration of heavy metal contaminants by marsh plants.

Helicopter landings for plant collections at selected locations within the respective refuge areas were arranged by the following: Mr. Russ Ernest, Manager, Area 3, Region 4, U. S. Fish and Wildlife Service (USFWS); Mr. Frank Johnson, USFWS, Aransas National Wildlife Refuge; Mr. Russ Clapper, USFWS, Anahuac National Wildlife Refuge; Mr. Ron Bisby, USFWS, San Bernard and Brazoria National Wildlife Refuges; Mr. Donald Hankla, Manager, Area 1, Region 4, USFWS; Mr. William Hickling, Manager, Area 2, Region 4, USFWS; Dr. John Bozeman, Georgia Department of Natural Resources; Mr. Martin D. Perry, Acting Project Leader, Savannah Complex of Refuges; Mr. Larry Nygren, USFWS, Morton, Amagansett, Conscience Point, Target Rock, Brigantine, and Barneget National Wildlife Refuges; Messrs. Phil Feiger and Charlie Blair, USFWS, Eastern Neck National Wildlife Refuge; Messrs. Bill Julian and Matt Kershbaum, USFWS, Martin National Wildlife Refuge; Mr. Dell Kidde, UCFWS Area 5; and Mr. William Sherman, Superintendent, Metro beach, Metropark, Mt. Clemens, Mich.

The Commanders and Directors of the WES during the study and the preparation and publication of the report were COL John L. Cannon, CE, and COL Nelson P. Conover, CE. Technical Director was Mr. F. R. Brown.

This report should be cited as follows:

Simmers, J. W. et al. 1981. "Field Survey of Heavy Metal Uptake by Naturally Occurring Saltwater and Freshwater Marsh Plants," Technical Report EL-81-5, U. S. Army Engineer Waterways Experiment Station, CE, Vicksburg, Miss.

## CONTENTS

													Page
PREFACE									•	•		•	1
LIST OF FIGURES										•			4
PART I: INTRODU	CTION					•							6
Background Purpose an Approach	d Scope												6 6 <b>8</b>
PART II: FIELD	COLLECTION .												9
Spartina a Cyperus Co Field Plan Helicopter Laboratory	nd Documentat **Iterniflora C **Ilection **t Identificat ***Utilization **Procedures **I Analysis .	ollecti  ion . in the	ion Field	· · · · · · · · · · · · · · · · · · ·		• •	•		• •	•	:		9 11 17 19 19 20 20
PART III: RESUL	TS AND DISCUS	SION .							•	•			21.
	Natural Marsh Natural Mars		• • •			•							21 45
PART IV: CONCLU	SIONS AND REC	OMMENDA	ATIONS										62
REFERENCES													63
TABLES 1-17													
	TOGRAPHS OF S. LECTION SITES		ER AND			TER					•		Al
APPENDIX B: SAL	TWATER COLLEC	TION SI	TE PHY	SICA	AL D	ATA							Bl
APPENDIX C: FRE	SHWATER COLLE	CTION S	SITE PH	iysi	CAL	DATA	٠.						Cl
APPENDIX D: CYF	ERUS SPECIES	COLLECT	ED BY	SITI	Ξ.								Dl
OF	F TISSUE HEAV SPARTINA ALTE	RNIFLOR	RA SAME	PLES	•••	•							El
APPENDIX F: MEA	N HEAVY METAL ERNIFLORA IN	UPTAKE NATURAL	E (μg n L MARSH	n <sup>-2</sup> ) HES	OF.	SPAF	RTII	7A • •	•			•	Fl
APPENDIX G: HEA	VY METAL CONT	ENT (µg	$(g^{-1})$	OF C	CYPE	RUS	SPE	CIE	S				G

## LIST OF FIGURES

No.		Page
1	Locations of sediment and marsh plant collections	1.0
2	Saltwater natural marsh collection sites, Corpus Christi, CCl-CCl5	. 12
3	Saltwater natural marsh collection sites, New Orleans, NO1-NO12	. 13
4	Saltwater natural marsh collection sites, Jacksonville, JV1-JV12	. 14
5	Saltwater natural marsh collection sites, New York, NY1-NY12	. 15
6	Saltwater natural marsh collection sites, Baltimore, BM1-BM12	. 16
7	Freshwater natural marsh collection sites, Detroit, DEl- DE9; Menominee, MN1-MN6; Milwaukee, MW1-MW2; Indiana Harbor, IN1; and Michigan City, MC1-MC13	. 18
8	Sites A through L for site-specific comparisons of leaf tissue concentrations and uptake data from natural marsh and disposal areas	. 23
9	Distribution of arsenic concentration, S. alterniflora	. 25
10	Distribution of cadmium concentration, S. alterniflora	. 26
11	Total cadmium uptake by S. alterniflora in natural marshes and disposal sites	. 27
12	Distribution of chromium concentrations, S. alterniflora	. 29
13	Total chromium uptake by S. alterniflora in natural marshes and disposal sites	. 30
14	Distribution of copper concentrations, S. alterniflora	31
15	Total copper uptake by S. alterniflora in natural marshes and disposal sites	. 33
16	Distribution of iron concentrations, S. alterniflora	. 34
17	Distribution of lead concentrations, S. alterniflora	. 35
18	Total lead uptake by S. altermiflora in natural marshes and disposal sites	. 37
19	Distribution of manganese concentrations, S. alterniflora	. 38
20	Distribution of mercury concentrations, S. alterniflora	. 39
21	Total mercury uptake by S. alterniflora in natural marshes and disposal sites	. 41

No.			Page
22	Distribution of nickel concentrations, S. altermiflora	•	42
23	Total nickel uptake by S. alterniflora in natural marshes and disposal sites	•	43
24	Distribution of zinc concentrations, S. alterniflora		44
25	Total zinc uptake by S. alterniflora in natural marshes and disposal sites		46
26	Distribution of arsenic concentrations, Cyperus species		48
27	Distribution of cadmium concentrations, Cyperus species		50
28	Distribution of chromium concentrations, Cyperus species		51
29	Distribution of copper concentrations, Cyperus species		53
30	Distribution of iron concentrations, Cyperus species	•	54
31	Distribution of lead concentrations, Cyperus species		55
32	Distribution of manganese concentrations, Cyperus species		57
33	Distribution of mercury concentrations, Cyperus species		58
34	Distribution of nickel concentrations, Cyperus species		59
35	Distribution of zinc concentrations, Cyperus species		60

The state of the s

# FIELD SURVEY OF HEAVY METAL UPTAKE BY NATURALLY OCCURRING SALTWATER AND FRESHWATER MARSH PLANTS

PART I: INTRODUCTION

### Background

- 1. The U. S. Army Corps of Engineers is required to dredge more than 205 million cubic metres of sediment from the waterways of the United States each year to maintain navigation channels (Water Resources Support Center 1979). Disposal of this amount of sediment is of immediate concern. Marsh creation from dredged material has received increased public support in recent years. However, contaminants such as toxic metals may move from the substrate through plant uptake and seriously affect biological components of the ecosystem. The heavy metals that are considered contaminants when present in high concentrations in dredged material are also present naturally in soils and sediments, presumably in lower concentrations.
- 2. Research under the Dredged Material Research Program (DMRP) and the Dredging Operations Technical Support Program (DOTS) has documented the uptake and bioconcentration of certain heavy metals in the common marsh plants Spartina alterniflora and Cyperus esculentus when these species are grown on contaminated dredged material. However, no complementary data base exists for the same species grown on naturally occurring sediments. A need exists to develop a data base that will serve as a baseline for evaluation of contaminant uptake by plants grown on contaminated dredged material being used for marsh or upland habitat development. The establishment of this natural marsh heavy metal baseline is necessary to relate plant uptake data from dredged material to naturally occurring levels of heavy metal uptake.

#### Purpose and Scope

3. This study is the fourth in a group of studies addressing the

bioconcentration of heavy metal contaminants via marsh plants. In a previous greenhouse hydroponic study, Lee, Sturgis, and Landin (1976) found that Cyperus esculentus, Spurtina patens, S. alterniflora, and Distichlis spicata accumulated heavy metals from the hydroponic solutions. In development of prediction techniques for contaminant mobility, Lee, Sturgis, and Landin (1978) related heavy metal accumulations in S. alterniflora, S. patens, and D. spicata collected from dredge! material disposal sites to the heavy metal content of various chemical extractions of dredged material. The greenhouse study of Folsom, Lee, and Bates (1980) focused on the distinctions between reduced (flooded) and oxidized (upland) contaminated sediments with respect to heavy metal uptake by C. esculentus and S. alterniflora. Throughout the remainder of this report, these studies will be referenced as the hydroponic study, the disposal site study, and the greenhouse study, respectively. Accordingly, a field survey and sampling study was designed with the following objectives:

- a. To determine concentrations of heavy metals in plants from natural marshes along the Atlantic and gulf coasts and around the Great Lakes.
- <u>b</u>. To compare the heavy metal concentrations in plants from the above-mentioned natural marshes to those previously found in marsh plants grown on dredged material.
- 4. The field sampling was based on the plant species used in the greenhouse study. The estuarine marsh plant Spartina alterniflora (Loisel.) Merr. was collected along the Atlantic and gulf coasts. This widespread, abundant species took up heavy metals readily during the hydroponic study. In the greenhouse study, S. alterniflora accumulated some heavy metals. Locations sampled were limited to the Atlantic and gulf coasts of the United States. Sampling sites included natural marshes located near industrial and urban areas as well as wildlife refuges and National, State, and local parks or natural areas. Sampling was limited to those plants colonizing naturally occurring sediment.
- 5. Cyperus species, C. esculentus L., C. strigosus L., C. odoratus L., C. erythrorhyzos Muhl., and C. Englemanni Steud., were collected from naturally occurring marsh areas along the United States' shore of

the Great Lakes. Cyperus esculentus took up metals very rapidly from hydroponic solutions and was subsequently utilized in the freshwater portion of the greenhouse study. While C. esculentus is not the dominant plant in the natural marshes of the Great Lakes, one or a combination of the five similar species mentioned above is present.

## Approach

6. Helicopters were used to collect plant leaf samples from naturally occurring colonies of S. alterniflers and Cuperus species (C. esculentus where possible). At each site, four to five composite samples were taken. The samples were collected at maximum veretative growth and at least half were collected in proximity to previous sediment, dredged material, or plant collection sites for direct communiscs. This study was concerned only with plant leaf heavy metal concentrations; therefore, sediment samples were not collected. The plant comples were analyzed for zinc (Zn), cadmium (Cd), copper (Cu), nickel 'Ni', load (Pb), mercury (Hg), chromium (Cr), iron (Fe), man, tanego (Mn), and arsenic (As). Plant leaf processing procedured developed in the previous disposal site study were utilized. Heavy metal composition of plants in the natural marsh communities near the data lock of previous studies was also included in the data base in order to compare the contaminant levels in naturally occurring S. alterniflora and Cyperus species to those determined in the previous disposal site and greenhouse studies.

#### PART II: FIELD COLLECTION

## Location and Documentation of Sites

- 7. The selection of the natural marsh sites for the field study collections was based on the sources of the plants collected in the disposal site study and the dredged material and sediment utilized in the greenhouse study. Approximately one half of the samples were collected from natural marshes in proximity to the sediment collection areas used in the greenhouse study; the remaining plants were collected from natural marsh areas within 320 to 480 km along the coast overlapping the disposal area sites (Figure 1). Natural marsh areas selected included locations mentioned in the above studies and areas within Federal, State, and local wildlife refuges. Spartina alterniflora was collected from the saltwater marsh areas and Cyperus species were collected from the freshwater marsh areas.
- 8. Within each general area (Figure 1), possible natural marsh collection sites were located on U. S. Geological Survey (USGS) topographic sheets, 7.5-min series where possible, and the information transferred to 1:500,000 aeronautical sectional charts. The sectional chart usage improved communication with the charter helicopter pilots, facilitated advanced planning with contractors, and reduced redundant motions in the field. Permissions and clearances were requested for all potential collection areas from the appropriate Federal, State, and local jurisdictional agencies prior to departure.
- 9. Once the field party reached a possible collection site as shown on the sectional chart, a specific area of collection was chosen and plotted on the corresponding topographic sheet. Photographs were taken of each sample within the collection site. After collection, an aerial photograph was made of each site, looking north at 30- to 45-m altitude (Appendix A). A field sketch was made of each collection site with the locations of the plant samples denoted and the wet and dry areas of the collection site shown. The accompanying field notes included additional information on the features of the collection site. The photographs and

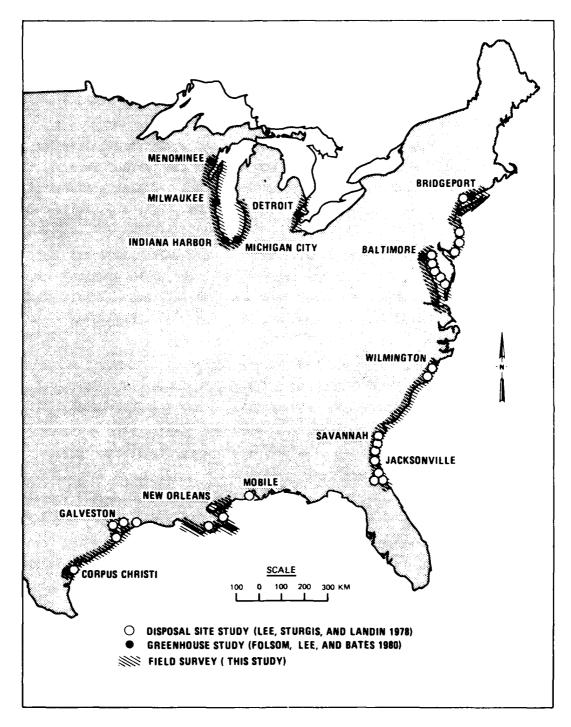


Figure 1. Locations of sediment and marsh plant collections

field sketches were compared to the topographic sheets in order to verify the map location of each site and the locations of the samples within each site. The latitude, longitude, and appropriate USGS topographic sheet for each verified saltwater and freshwater site were recorded (Appendices B and C, respectively).

10. The predicted tidal parameters for each of the salt marsh collection sites were tabulated from <u>Tide Tables 1978</u> (U. S. Department of Commerce 1977). Information on the prevailing wind direction, by month, at both saltwater and freshwater marsh collection sites (Appendices B and C, respectively) was extrapolated from weather station data from the nearest cities. Weather data were obtained from climatological references of Court (1974) and the Water Information Center, Inc. (1974).

## Spartina alterniflora Collection

- 11. Spartina alterniflora was collected from natural saltmarshes. Four samples of Spartina were taken from each collection site. Generally, two of the samples, labeled C and D, were taken from flooded conditions and two, A and B, were taken from drier, relatively upland but not necessarily farther inland, areas. Collection times were planned to permit collections to be made at low tide when tidal fluctuation was significant. Each sample consisted of the amount of Spartina that could be encompassed by a 28.7-cm square made from a folding carpenter's ruler or 823.7 cm<sup>2</sup>. The plants were clipped 5 cm above the ground with Wiss 68 or 607 clippers. The plant material from each sample was placed in a 115-& trash can liner with an acetate label, secured with a twist tie, and placed on ice in an ice chest for shipment within 24 hr to the U. S. Army Engineer Waterways Experiment Station (WES), Vicksburg, Miss.
- 12. Collections of Spartina alterniflora were made from 63 natural saltwater marsh areas along the U. S. gulf and Atlantic coasts in July and August 1978. Fifteen natural marshes were sampled along the Texas gulf coast on 18 and 19 July 1978 (Figure 2). Corpus Christi sites CCl-CC6 were taken in the vicinity of Nueces Bay, Tex., a site of the previous disposal site studies and the sediment source for the greenhouse

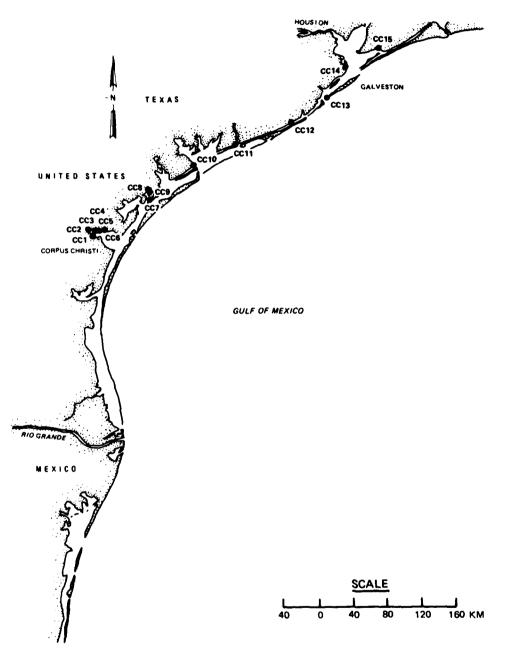


Figure 2. Saltwater natural marsh collection sites, Corpus Christi, CC1-CC15

study. Additional samples were collected north along the coast to the Houston area corresponding to the disposal site study collection areas. Collections of *Spartina* along the Louisiana coast (Figure 3) on

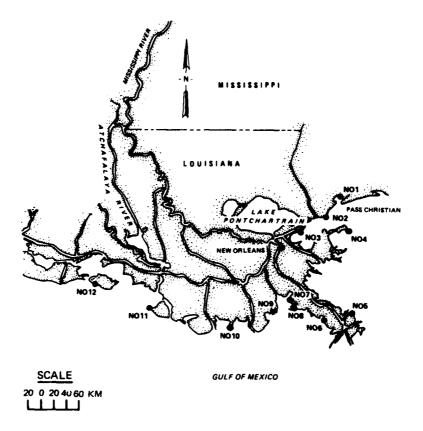


Figure 3. Saltwater natural saltmarsh collection sites, New Orleans, NO1-NO12

20 and 21 July 1978 were made in the same general area used in the disposal site study and in the study of Gosselink, Hopkinson, and Parrondo (1977). The 12 sampled areas in Louisiana were designated NO1-NO12. Twelve sampling areas were also utilized along the Atlantic Coast (JV1-JV12) from Jacksonville, Fla., north to Wilmington, N. C. (Figure 4), on 1 and 2 August 1978. These collection sites were based on the locations of the disposal site study and studies by Broome, Woodhouse, and Seneca (1973) and Dunstan and Windom (1975). In the New York area (Figure 5), collections NY1-NY6 were made in natural marshes adjacent to the Bridgeport, Conn., sediment collection sites of the greeenhouse study and near

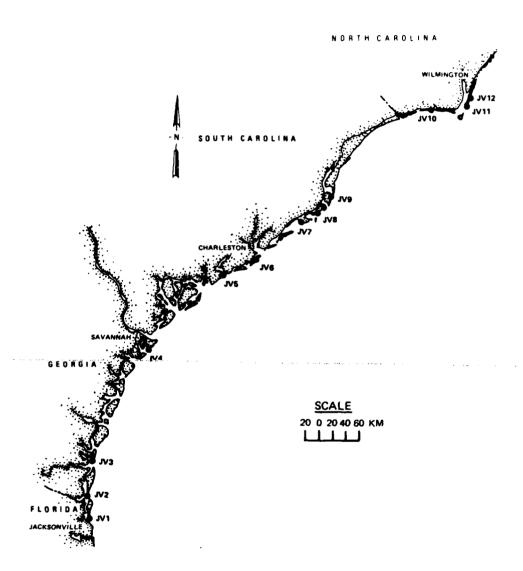


Figure 4. Saltwater natural marsh collection sites, Jacksonville, JV1-JV12

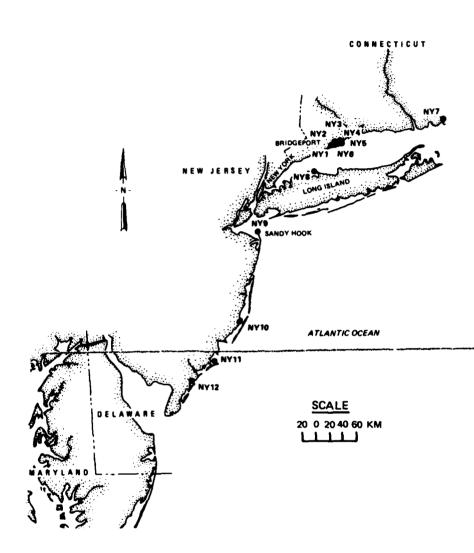


Figure 5. Saltwater natural marsh collection sites, New York, NY1-NY12

a disposal site study area. The remaining marsh collection sites were located along Long Island Sound and south along the New Jersey coast where disposal sites had been sampled. Collections were made 8 and 9 August 1978. The final saltwater natural marsh collections were taken in the Baltimore, Md., area on 10 and 11 August 1978 (Figure 6). Sites

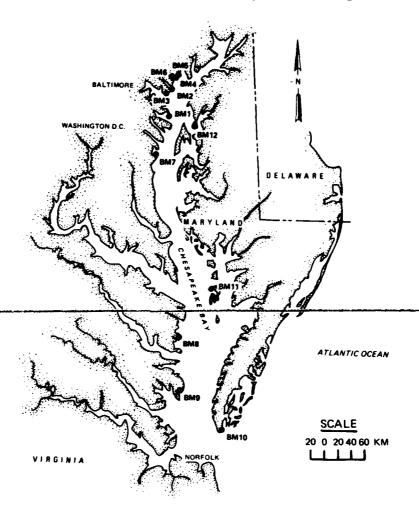


Figure 6. Saltwater natural marsh collection sites, Baltimore, BM1-BM12

BM1-BM6 were located in natural marsh areas near Baltimore Harbor with several near the collection sites of the disposal site study and the sediment collecton sites of the greenhouse study. Collection sites BM7-BM12 were located along the Chesapeake Bay marshes south to the

Norfolk, Va., area near locations of previously sampled disposal sites utilized in the disposal site study and the study by Drifmeyer and Odum (1975).

## Cyperus Collection

- marshes associated with the harbors and adjacent coastal areas of Menominee, Mich.; Milwaukee, Wis.; Indiana Harbor, Ind.; Michigan City, Ind.; and Detroit, Mich. (Figure 7). Cyperus is not an abundant genus in these freshwater marshes, and, therefore, the sampling system was adjusted. A folding carpenter's ruler was arranged as a 59.2- by 28.7-cm rectangle (1669 cm²) and placed around each of the four most abundant stands of Cyperus. The samples were designated A, B, C, and D. The plants were harvested in the same manner as the Spartina with the exception that 3.8-l storage bags were used to contain the plants. A fifth composite grab sample of Cyperus, labeled E, was collected by hand picking the plants dispersed around the site.
- 14. At each collection site an attempt was made to locate any tubers formed by the species present. Tubers were not found.
- Lakes area at 31 sites (Figure 7) during August 1978. Natural marshes were selected near the sediment collection sites, namely Detroit,
  Menominee, Milwaukee, Indiana Harbor, and Michigan City. Collections in the vicinity of Detroit (coded DE) were made along the west shore of Lake Erie, from Toledo, Ohio, north along the Detroit River to Mt.
  Clemens, Mich., on 16 and 17 August. Six collections, MEl-ME6, were made in the Lake Michigan shoreline marshes at Menominee, Mich., on 22 August. The emergent lake shore in the Milwaukee (MW) area is not conducive to marsh formation. Only two suitable collection areas were found; plants were collected on 23 August. One natural marsh containing Cyperus was found near Indiana Harbor (IN) on 24 August. Thirteen Cyperus collections were made along the east shore of Lake Michigan northeast of Michigan City (MC), during 24 and 25 August. Michigan

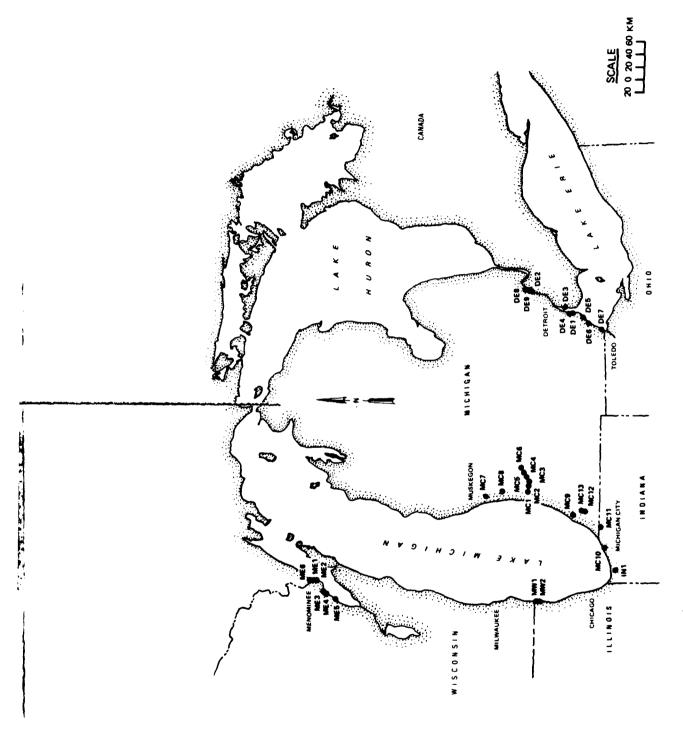


Figure 7. Freshwater natural marsh collection sites; Detroit, DE1-DE9; Menominee, ME1-ME6; Milwaukee, MW1-MW2; Indiana Harbor, IN1; and Michigan City, MC1-MC13

City collections (MC1-MC13) were made along the Kalamazoo River and smaller streams just east of Lake Michigan.

### Field Plant Identification

- 16. Spartina alterniflora was identified in the field from vegetative material by comparison with material from the greenhouse study and use of the key to vegetative material in <u>Grey's Manual of Botany</u> (Fernald 1950). No attempt was made to characterize the variety.
- 17. Cyperus is a highly technical genus. The many similarappearing species may only be accurately distinguished by examination of mature floral parts. Field identifications were made to genus only; species identifications were made later from voucher specimens stored as a permanent record. Representative inflorescences and seeds taken from the plants analyzed were mounted on herbarium sheets and placed in the WES Research Herbarium as sheets numbered 0001FS78 through 0031FS78. The Cyperus species collected were identified by utilization of the keys of Voss (1972), Gleason (1952), Fernald (1950), Swink and Wilhelm (1979), Mohlenbrock (1960), and Marcks (1974). Collected specimens were compared with materials in the herbaria of the U. S. National Museum, the Morton Arboretum, and the Department of Botany, Southern Illinois University-Carbondale. Identified as the field collections were C. esculentus, C. strigosus, C. odoratus (= C. feruginescens), C. erythrorhyzos, and C. Englemanni. The Cyperus species are listed by site in Appendix D.

#### Helicopter Utilization in the Field

18. The availability of helicopters made access to remote areas less time-consuming and allowed site selection to be enhanced by aerial survey. With a helicopter it was possible to select and visit six or more collection sites per day over more than 500 km of coastline. Aerial photographs made of the site, from the helicopter, significantly improved the accuracy of site documentation.

19. The most common helicopter in the size class utilized in this study is the shaft turbine-powered helicopter. The turbine engines have a higher safety factor and use jet fuel A, which contains no lead or compounds that could leave deposits on the plant leaves. When the helicopter was deployed downwind from the collection site, there was little possibility of fuel residue contaminating the plant samples. Additionally, surface contamination was addressed in the treatment of the plant material in the laboratory.

### Laboratory Procedures

20. Plant samples were shipped and stored at 4°C until processed. The plant leaf samples were cleaned using the procedure of Elias and Patterson (1975) as modified during the previously described disposal site and greenhouse studies. Several duplicate samples were processed without the cleaning procedure for evaluation of the extent of leaf surface contamination. The plant tissue was oven dried at 70°C, ground into a coarse powder using a Wiley mill, and digested by nitric acid. The plant digestates were analyzed for Zn, Cd, Cu, Ni, Pb, Hg, Cr, Fe, Mn, and As. The instrumentation and detection limits of the chemical parameters are given in Table 1.

### Statistical Analysis

21. Descriptive statistics were calculated for all chemical concentration and uptake data for the plants, by species and collectively. A paired test was used for guidance in (a) evaluation of differences in contaminant content of S. alterniflora collected from flooded and upland conditions, (b) comparison of contaminant data from washed and unwashed plants addressing the significance of surface contamination, and (c) comparison of quadrat and grab samples of Cyperus species.

#### PART III: RESULTS AND DISCUSSION

## Saltwater Natural Marshes

Comparison of flooded and upland plant collections

22. No statistically significant differences in heavy metal content of S. alterniflora were found when the means of samples A and B, collected upland at each site, and samples C and D, collected from flooded areas at each site, were compared (Table 2). Studies of rice grown in oxidized and reduced environments suggest that less heavy metal uptake occurs under reduced (flooded) conditions (Bingham et al. 1976; Jugsujinda and Patrick 1977; and Reddy and Patrick 1977). The greenhouse study established the relationship of sediment contamination to plant contaminant uptake for S. alterniflora and D. spicata under flooded conditions. In addition, C. esculentus grown on reduced (flooded) contaminated sediments was shown to take up less Cd, Zn, and Mn than when grown under oxidized (upland) conditions. Comparison of the data of this field study with that of the greenhouse study (Figures 9-17) often shows a generally higher heavy metal concentration in the plants from the natural marsh. The differences between the natural marsh and greenhouse data may be explained by the nature of the tidal fluctuation in the natural saltwater marsh. Generally, during a lunar month the tidal cycles result in flooding of the whole collecting site; additional complete or partial submergence may be due to storm events. Likewise, during the same time span the site may be partially drained and dry or oxidized on the surface. The natural marsh substrate may have been oxidized to a greater extent than the continuously flooded sediment in the greenhouse study. An illustration of tidal range at sites NY1-NY5 (Figure 5) in the Bridgeport, Conn., area was 0 to 2.16 m above mean sea level (msl) for the collection date of 8 August 1978, but the range for the year, excluding storm events, was -0.46 to 2.41 m above msl (U. S. Department of Commerce 1977). As a consequence, the marsh sediment has an opportunity to drain and become oxidized when the tide is -0.46 m above msl. Increased

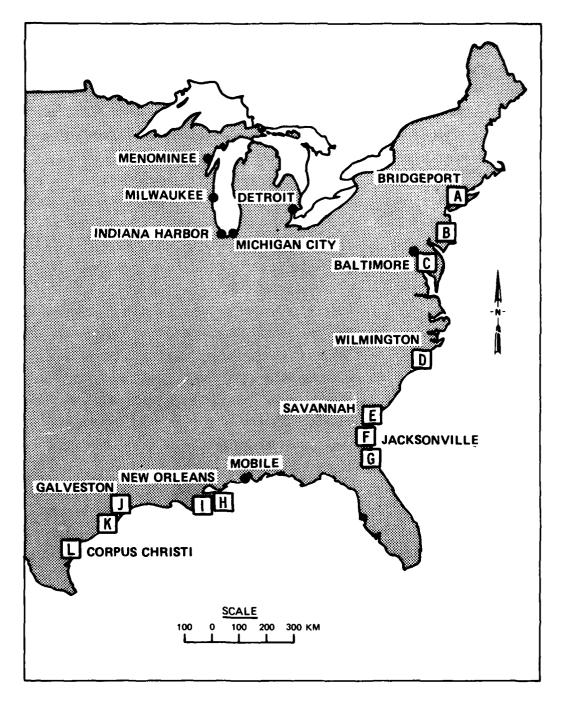
oxidation of the marsh substrate may increase plant available heavy metals with a resultant increase in plant contents.

## Comparison of the heavy metal content of washed and unwashed samples

23. Statistically significant differences in heavy metal concentrations were only found between Cd concentrations of washed and unwashed S. alterniflora when leaf samples were compared (Table 3). These data are in contrast to the significantly higher leaf adsorbed Ni and Zn of the disposal site survey attributed to airborne contaminants. Apparently the natural marsh plants collected were not noticeably influenced by airborne contaminants. It appears, from the sporatic results of washing, that washing the leaves is only necessary when airborne contamination is a reality at the collection site.

## Heavy metal content of S. alterniflora leaf tissue

- 24. Leaf tissue heavy metal concentrations (Appendix E) were compared with those determined in the previously described greenhouse and disposal site studies and with existing literature values.
- 25. Both leaf tissue concentration (Appendix E) and total plant uptake values, calculated by multiplying leaf tissue concentration by the total yield of the sample collected and then dividing by the area sampled (square metres) (Appendix F), were considered for each metal.
- 26. The data presented in Appendices E and F indicate that a pronounced geographic variation occurs in the baseline heavy metal levels in the natural marsh. To address this variation, site-specific comparisons were made utilizing existing literature values and portions of the data collected during the disposal site study. Leaf tissue concentration and uptake data from natural marsh and disposal sites were compared for sites located within approximately 0.5 deg of latitude and longitude (Table 4). These are arbitrarily designated areas in which the compared sites are no more than 60 km apart within an area of approximately 11,840 km<sup>2</sup>. Each area is small enough to allow regional grouping of data loci for comparison. These resultant areas were designated areas A through L (Figure 8).



- 27. The comparison of total heavy metal uptake values of plants from natural marsh and disposal site at maximum vegetative growth is intended to reflect the generally greater biomass found on the disposal sites (Table 5) and the concomitant increased potential for contaminant movement into the biota. The comparisons of disposal site uptake, based on collections made in 1975 and the natural marsh survey of 1978, may be less indicative of real differences than comparisons of plants collected during the same year.
- 28. Arsenic. The concentration of As in S. alterniflora was generally very low, usually less than 0.03 μg g<sup>-1</sup> or below detection limits (Figure 9). This is similar to the level observed in the greenhouse study. Due to the hot acid digestion of the plant samples, these values may be low due to As loss via volitization. In the natural marsh a few relatively high levels were observed at JV9 C, 0.83 μg g<sup>-1</sup>, and JV1 C, 0.48 μg g<sup>-1</sup>. Arsenic uptake for JV1 through JV4, which are in Areas E, F, and G in Figure 8, ranged from 25.5 to 63.7 μg m<sup>-2</sup>. Jacksonville 9 was the location with the greatest plant uptake value, 164.6 μg m<sup>-2</sup>, without explanation. Other areas with high total plant As uptake values were NY2, NY3, and NY4 in Area A, Long Island Sound; BM1, near Baltimore Harbor; CC10 and CC11 in Area K; CC1, Nueces Bay in Area L; and NO11 and NO12 in Area I. There are no disposal site data available with which to make a total As uptake comparison.
- 29. Cadmium. The Cd content of S. alterniflora leaves was usually less than  $0.02~\mu g~s^{-1}$ , a level slightly less than that of the disposal site study and significantly less than the levels of the greenhouse study (Figure 10). A few samples from NY1, NY3, and NY4 (Figure 5) ranged in concentration upward to  $0.72~\mu g~s^{-1}$ , although these sites were not uniformly high in Cd. The uptake in the natural marsh in Area A, an industrial area of long standing that includes these sites, was notably higher than that of the disposal site of the same area (Figure 11). In the remaining coastal areas, the Cd uptake values were lower in the natural marshes sampled or equal to those of the disposal sites (Figure 11). Dunstan and Windom (1975) found an average concentration of  $0.61~\mu g~s^{-1}$  Cd in the natural marshes along six southeastern United States river

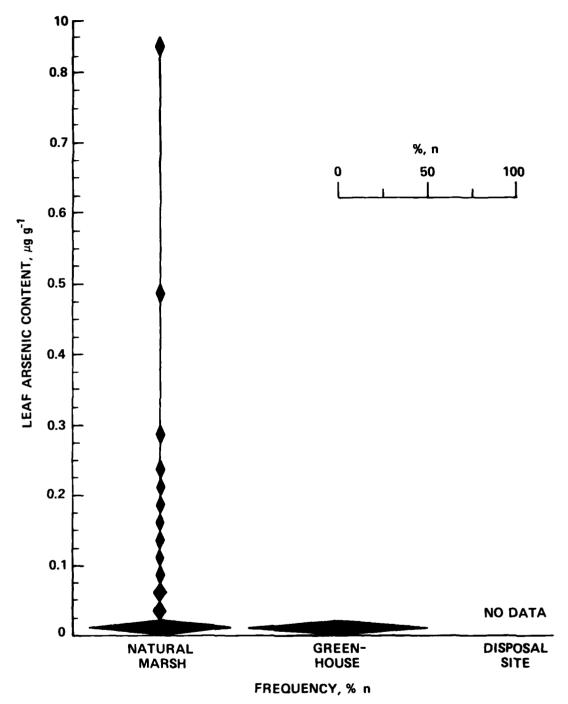


Figure 9. Distribution of arsenic concentration, S. alterniflora

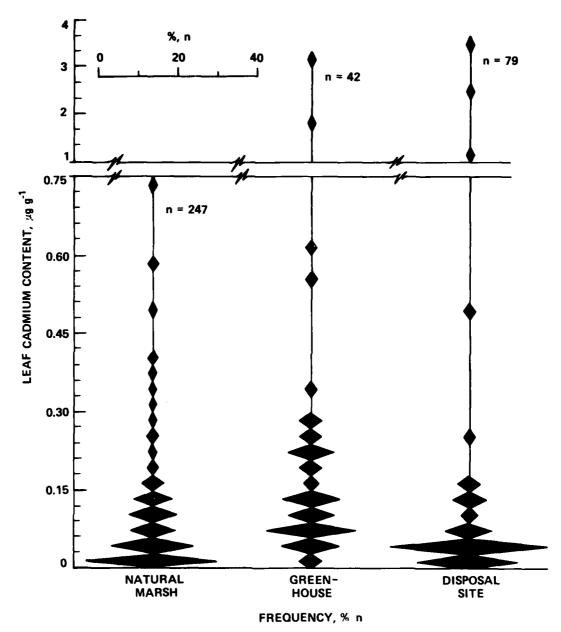


Figure 10. Distribution of cadmium concentrations, S. alterniflora

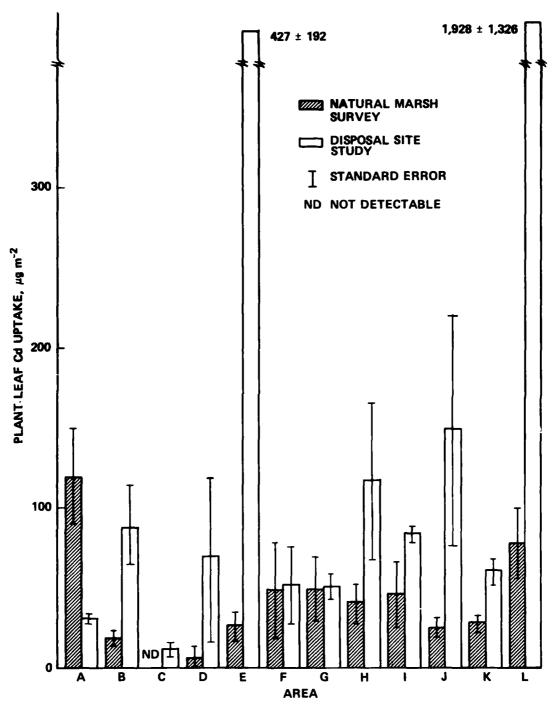


Figure 11. Total cadmium uptake by S. alterniflora in natural marshes and disposal sites

- systems, a value higher than the mean of 0.02  $\mu g g^{-1}$  for the Cd in the same species reported in this study. The plant Cd mean of the JV sites that generally corresponds with Dunstan and Windom's (1975) collection area was 0.07  $\mu g g^{-1}$ . With the exception of Area A (Figure 8), Cd content and uptake in natural marsh S. alterniflora was less than or equivalent to that of the same species grown on disposal sites or flooded contaminated sediments (Figure 11). Disposal sites appear to be contributing more Cd into the environment via plants than the natural marsh.
- 30. Chromium. The mean Cr content of natural marsh S. alterniflora was 1.17 µg g<sup>-1</sup> (Figure 12). This is a significantly higher level than found in either the greenhouse or the disposal site studies. The levels of Cr uptake in the natural marsh of the comparison areas (Figure 13) were higher or within one standard error of the uptake of the disposal sites. Area D (Figure 8) would perhaps be expected to have a higher level due to the Wilmington, N. C., fabric mills and the utilization of Cr in dyes, but this difference was not statistically significant. There are no data available in the literature for further comparison.
- 31. Copper. Leaf samples of S. alterniflora grown in natural marshes contained 1.2 to 5.5 µg g<sup>-1</sup> Cu. This is similar to the Cu content of those plants from the disposal site and greenhouse studies (Figure 14). Broome, Woodhouse, and Seneca (1973) reported 2.0 to 4.0 ug g-1 Cu from S. alterniflora collected from natural marshes along the North Carolina coast. Collections made from the same general area, JV11 and JV12 (Figure 4) during this field survey yielded values of 0.2 to 3.0 µg g<sup>-1</sup>, values similar to those of the previous authors. Analysis of S. alterniflora from the U. S. gulf coast by Gosselink, Hopkinson, and Parrondc (1977) indicated a mean Cu level of 4.0 µg g<sup>-1</sup>. The analyses field survey collection in the New Orleans (NO) area indicated the leaf samples contained 0.2 to 11.2 µg g<sup>-1</sup> Cu (Appendix E6) with the majority of the samples falling in the 2.0- to 4.0-µg g<sup>-1</sup> range. These Cu content levels were similar to those found in S. patens and D. spicata during the disposal site study. The highest concentrations observed were in Areas A and B (Figure 8) and NO4 and NO5 (Figure 3). The former locations (Areas A and B) were associated with intensive

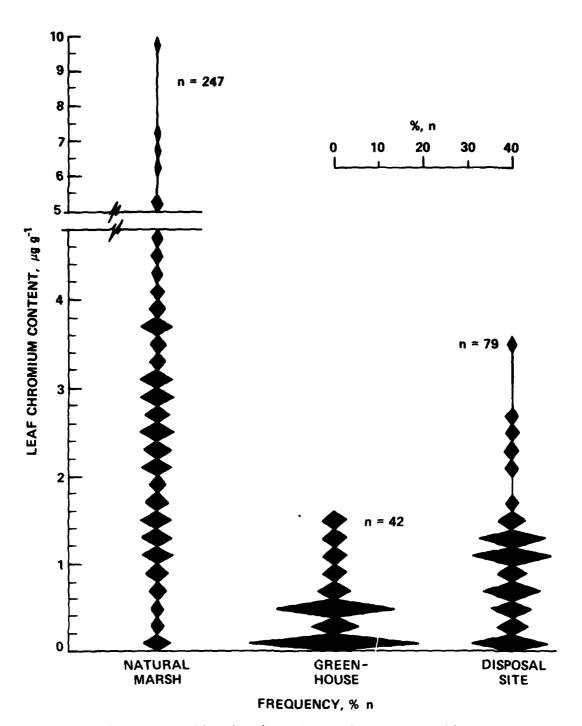


Figure 12. Distribution of chromium concentrations,  $S.\ alterniflora$ 

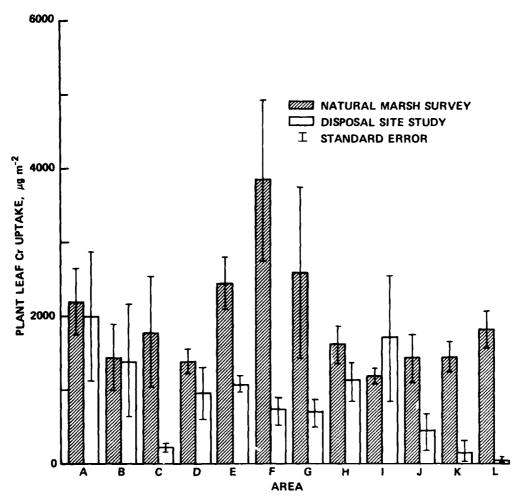


Figure 13. Total chromium uptake by  $S.\ alterniflora$  in natural marshes and disposal sites

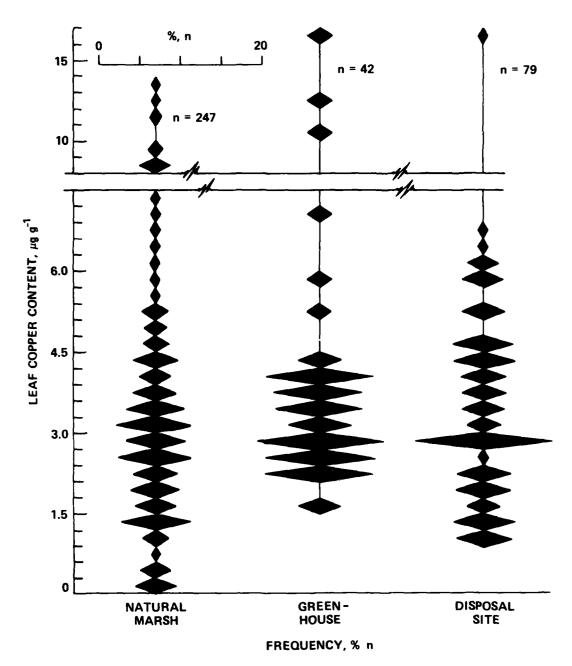


Figure 14. Distribution of copper concentrations, S. alterniflora

industrialization, but the latter (NO4 and NO5) were located in isolated marsh areas. In terms of leaf content data, it appears that plants grown on flooded contaminated sediment and plants grown on dredged material disposal sites did not contain any more Cu than natural marsh plants.

- 32. Total uptake of Cu in the natural marsh plants was generally less than or equal to that of the plants collected during the disposal site study (Figure 15). This observation reflects the generally lower biomass production in the natural marsh as compared to that in the disposal site. As expected from the previous discussion, natural marsh Cu uptake was greatest in Area A. In Area C, the natural marsh Cu uptake was considerably higher than that of the disposal sites. This is mainly the result of a higher biomass production in the natural marsh in that area.
- 33. Iron. The concentration of Fe in natural marsh S. alterniflora was marginally greater than that recorded in the greenhouse study (Figure 16). This was contrary to the more completely reduced conditions maintained in the greenhouse study under which Fe availability was thought to be increased. Consistently high values for Fe were associated with more than one sample at JV9, NY1, NY4, NY6, and NO12. NY1, NY4, and NY6 were sites located in and around Bridgeport, Conn., an area of long standing industrial activity (Figure 5). In addition, NY6 was adjacent to a gun club on Stratford Point, Conn., which is also a site with high plant leaf Pb content. The Fe uptake at these sites is also high, ranging from 297,482 to 792,682  $\mu$ g m<sup>-2</sup>. Site JV9 was located at Santee Point, S. C., and appeared to be essentially free from any source of contamination (Figure 4), although the Fe concentration ranged from 80 to  $4738 \, \mu g \, g^{-1}$ , with a mean of 1328  $\mu g \, g^{-1}$ . Dunstan and Windom (1975) found 1000 µg g<sup>-1</sup> Fe in Spartina, leaves and rhizomes combined, at this same location. Site NO12 samples were collected on Marsh Island, a Louisiana State Wildlife Refuge (Figure 3). There were no Fe data collected during the disposal site study for comparison.

 $3^{4}$ . Lead. The mean Pb concentration of natural marsh S. alterniflora was 1.3  $\mu g g^{-1}$ , a value greater than that of the greenhouse study, 0.6  $\mu g g^{-1}$ , and the disposal site study, 0.9  $\mu g g^{-1}$  (Figure 17). Plant

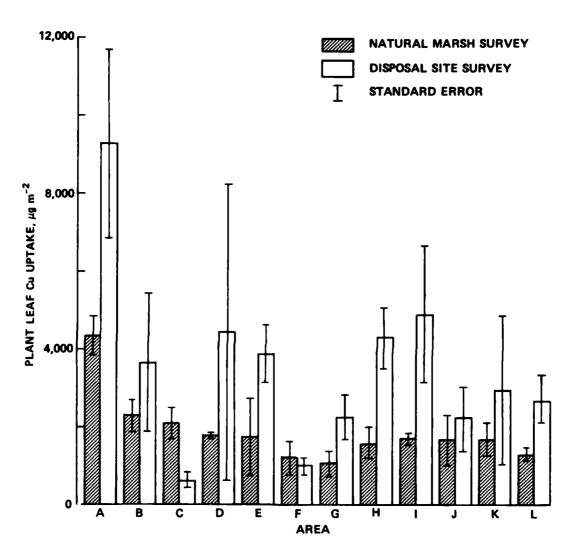


Figure 15. Total copper uptake by S. alterniflora in natural marshes and disposal sites

The state of the s

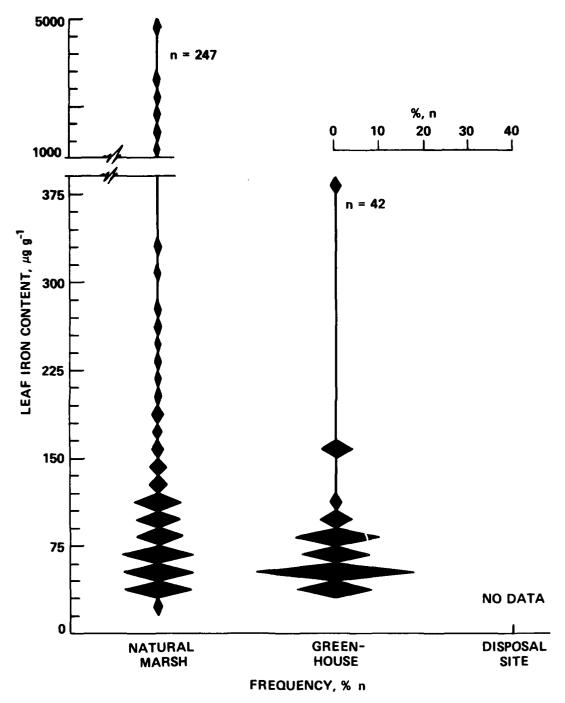


Figure 16. Distribution of iron concentrations, S. alterniflora

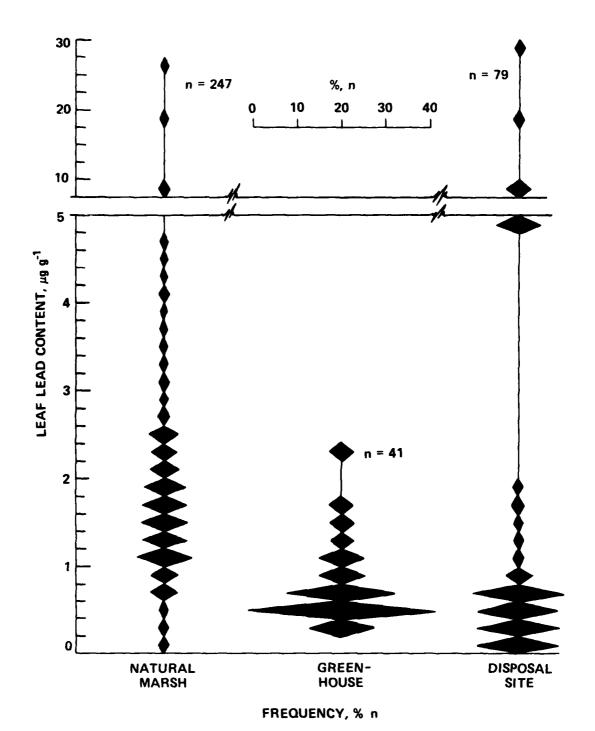


Figure 17. Distribution of lead concentrations, S. alterniflora

leaf Pb content was greater than 5.0  $\mu g e^{-1}$  at only one site, NY6. This natural marsh was adjacent to the gun club on Stratford Point, where the high Pb content may be due to Pb shotgun pellets. This site was also high in Fe. Drifmeyer and Odum (1975) reported a concentration of 9.1  $\mu g^{-1}$  Pb in S. alterniflora collected from a confined dredged material disposal area in Virginia and 1.9  $\mu g g^{-1}$  in a natural marsh nearby. The mean of Pb concentrations in S. alterniflora in the natural marsh plants sampled during this study, 1.3 µg g<sup>-1</sup>, is slightly lower than the value of the previous authors but represents a range of 0.0 to 26.7 µg g<sup>-1</sup> (Appendix E). The mean of the Pb concentrations from sites BM8 to BM11 (Figure 6), near the collection area of Drifmeyer and Odum (1975), was 1.8 µg g<sup>-1</sup>. Lead uptake was found to be highest in Arca A that includes NY6 (Figure 18). With the exception of Areas A, B, and L, Fb uptake in the natural marsh was less than or equal that in the disposal sites. Disposal site uptake levels were higher than 1000 µg m<sup>-2</sup> only in Areas A, B, and L.

- 35. Manganese. Levels of Mn in the natural marsh were found to be significantly lower, with a mean of 51.6 µg g<sup>-1</sup>, than those of the greenhouse study (Figure 19). No data from the disposal site study were available for comparison. As demonstrated in the greenhouse study with Cyperus esculentus, Mn uptake would be greater under reduced conditions. As tidal fluctuations may allow some oxidation, the actual difference between the greenhouse and natural marsh plant leaf Mn is more pronounced. In contrast to the greenhouse study, examination of the Fe-Mn ratio did not indicate that Mn levels were generally higher than those of Fe. Manganese concentrations were uniformly high in Areas C, H, and J (Figure 8) and also in the NO6 area (Figure 3). Total plant Mn uptake (Appendix F) was commensurately higher in these areas also. There appeared to be no relationship between Fe and Mn concentration data on visual comparison.
- 36. Mercury. The concentration of Hg in S. alterniflora I aves was relatively low and similar to the disposal site findings, approximately one third the level found in the greenhouse plants (Figure 20). There were no areas of uniformly high Hg concentrations. While Dunstan and Windom (1975) found up to 0.44  $\mu_{\rm Hg}$  g<sup>-1</sup> of Hg in S. alterniflora in the

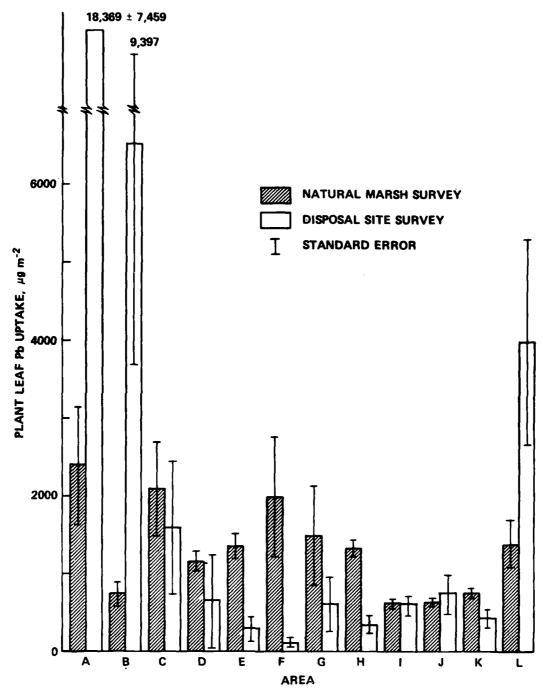


Figure 18. Total lead uptake by S. altermiflora in natural marshes and disposal sites

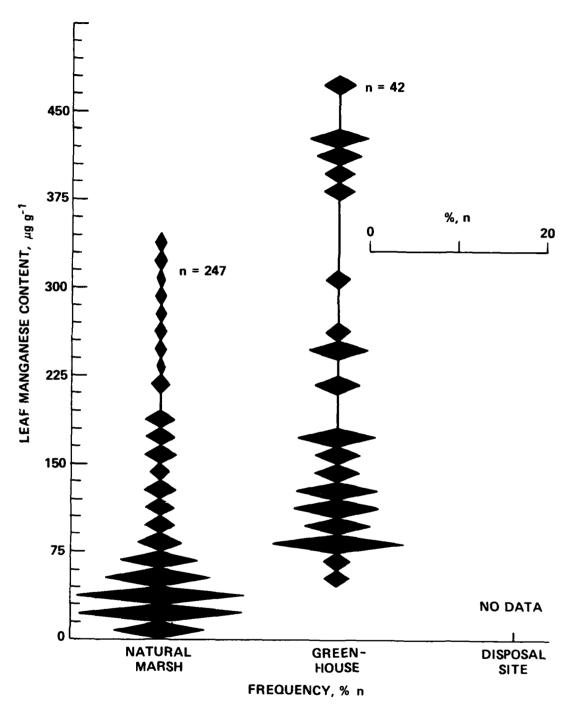


Figure 19. Distribution of manganese concentrations,  $S.\ altermiflora$ 

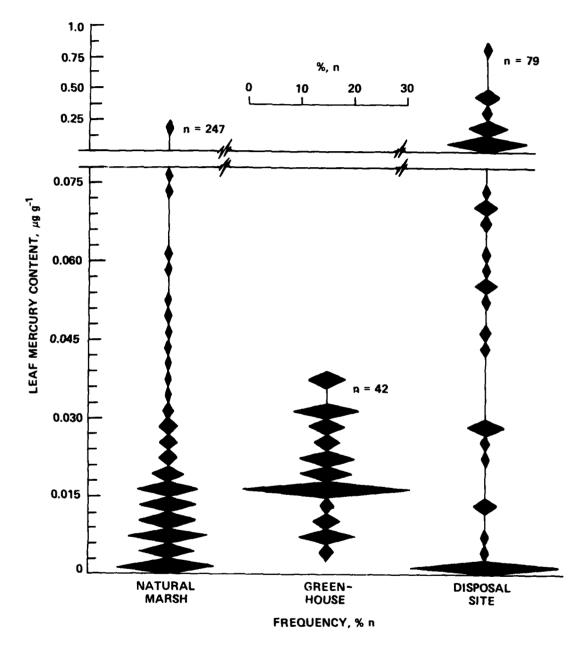


Figure 20. Distribution of mercury concentrations,  $S.\ alterniflora$ 

area referred to here as JV, no concentration of Hg determined during the natural marsh survey exceeded 0.07  $\mu g$  g<sup>-1</sup>. The levels of Hg in saltwater natural marsh plants approximate those found in plants grown on dredged material. Due to the low levels of Hg in both natural marsh and disposal site plants, extensive variation appears in the uptake comparison (Figure 21). Natural marsh Hg uptake values never exceeded 43  $\mu g$  m<sup>-2</sup> (Appendix F). These low values may be a result of Hg loss through volitization during the hot acid digestion of the plant material.

- 37. Nickel. The Ni concentration of S. alterniflora leaf tissue was generally less than 6.0  $\mu g \, g^{-1}$  (Figure 22). The distribution of leaf Ni concentrations generally paralleled that of the disposal site data in contrast to the very low levels, 0.3  $\mu g \, g^{-1}$ , found in the greenhouse study. Nickel uptake in the natural marsh plants (Figure 23) varied from higher (Areas E, F, K, and L) to within one standard error of the disposal site plants. Area A showed less Ni uptake in natural marsh plants than disposal site plants (Figure 23). The concentration of Ni at Areas E and F were, however, less than 3.5  $\mu g \, g^{-1}$  Ni, but due to more biomass production in the natural marsh compared to the disposal site, these areas showed the highest uptake values.
- 38. Zinc. The mean concentration of Zn in natural marsh S. alterniflora was 15 μg g<sup>-1</sup> (Figure 24). This value is slightly lower than those from the disposal site study and significantly lower than the 43 μg g<sup>-1</sup> mean of the greenhouse study plants. Gosselink, Hopkinson, and Parrondo (1977) found 11 μg g<sup>-1</sup> Zn in S. alterniflora of natural marshes in Louisiana. Collections from the same general vicinity, NO (Figure 3), yielded a higher mean of 14 μg g<sup>-1</sup>. Spartina alterniflora collections from JV11 and JV12 (Figure 4), near the locations from which Broome, Woodhouse, and Seneca (1973) reported 17 μg g<sup>-1</sup> Zn, contained a mean of 23 μg g<sup>-1</sup>. These values are higher than the 7 to 10 μg g<sup>-1</sup> Zn reported by Williams and Murdock (1969) from studies made further north at Beaufort, N. C. In the vicinity of BM8 to BM11 (Figure 6), in Virginia, Drifmeyer, and Odum (1975) reported Zn concentrations of 39 and 20 μg g<sup>-1</sup>, respectively, from a dredged material disposal area and an adjacent natural marsh. In comparison, the concentration of Zn

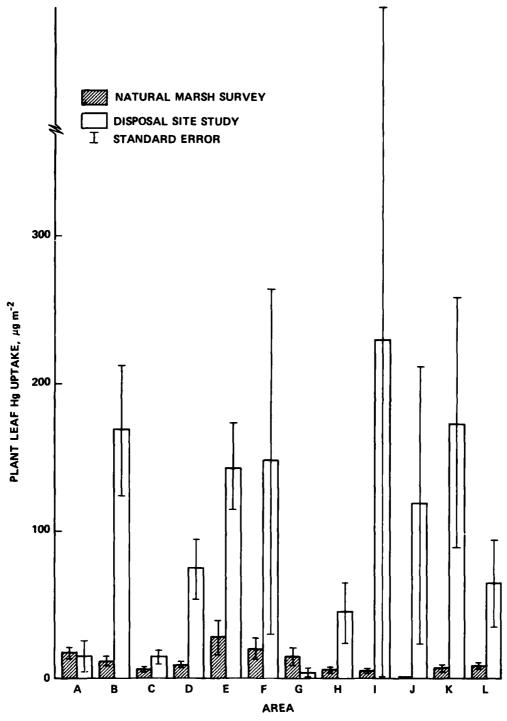


Figure 21. Total mercury uptake by  $S.\ alterniflora$  in natural marshes and disposal sites

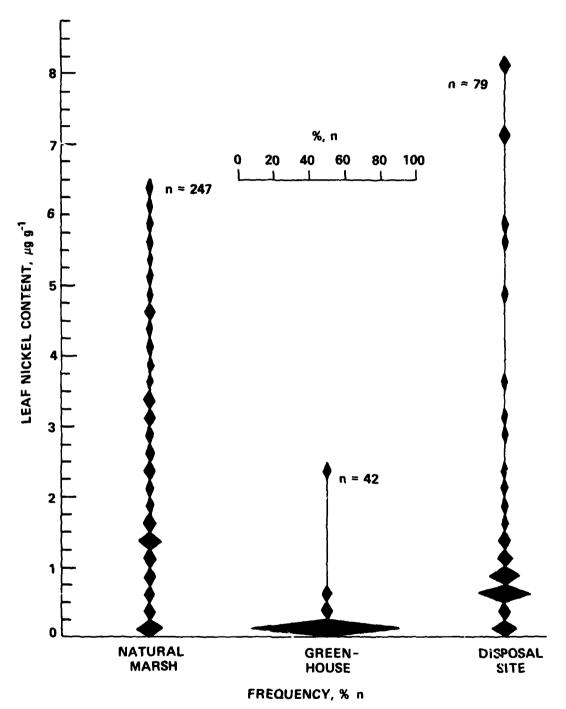


Figure 22. Distribution of nickel concentrations, S. alterniflora

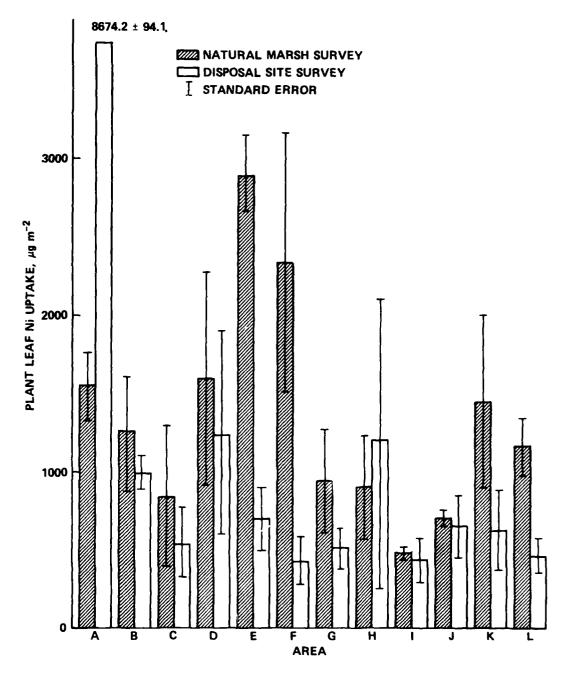


Figure 23. Total nickel uptake by S. altermiflora in natural marshes and disposal sites

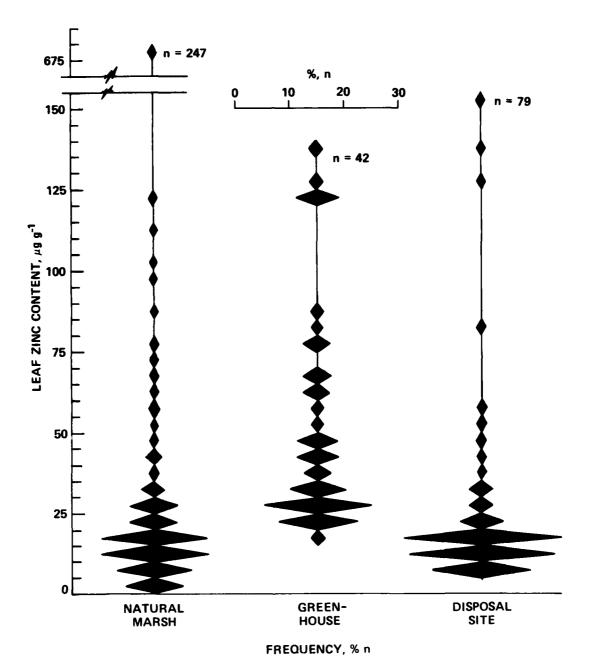


Figure 24. Distribution of zinc concentrations, S. alterniflora

found at BM8 to BM11 was 15.81 µg g<sup>-1</sup>. Zinc concentrations were not uniformly greater than 45 µg g<sup>-1</sup> in Area L (Figure 8) where, until 1975, a Zn smelter had been in operation. Zinc uptake in the natural marsh was generally equivalent to that of the disposal sites (Figure 25). Notable exceptions were the disposal sites in Area L which yielded plants with much higher Zn uptake than the natural marsh. Higher biomass production on the disposal site may have accentuated this difference in Zn uptake.

## Freshwater Natural Marshes

Comparison of heavy metal contents of quadrat and grab samples of *Cyperus* 

39. Statistical comparisons of the means of heavy metal contents of plants collected by both quadrat (samples A, B, C, and D) and hand picking or grab methods (sample E) indicated no significant difference (Table 6). The variability of the data does not allow a conclusion to be drawn in respect to the value of either sampling technique.

Correlation of heavy metal content and uptake with the percent of *Cyperus* in the quadrat

40. The percentage of *Cyperus* species by weight collected at each field site is shown in Table 7. Competition effects of other plant species within a site were hypothesized as potentially influencing the heavy metal content and uptake of *Cyperus*. Accordingly, a statistical examination was made to see if the lower biomass of *Cyperus* might have higher metal contents and lower metal uptake reflecting a dilution effect. A statistical examination did not evidence any correlation between the percent of *Cyperus* in the collected biomass and either concentration or uptake of heavy metals. The percent of *Cyperus* in the collected biomass was not correlated to either concentration or uptake of heavy metals.

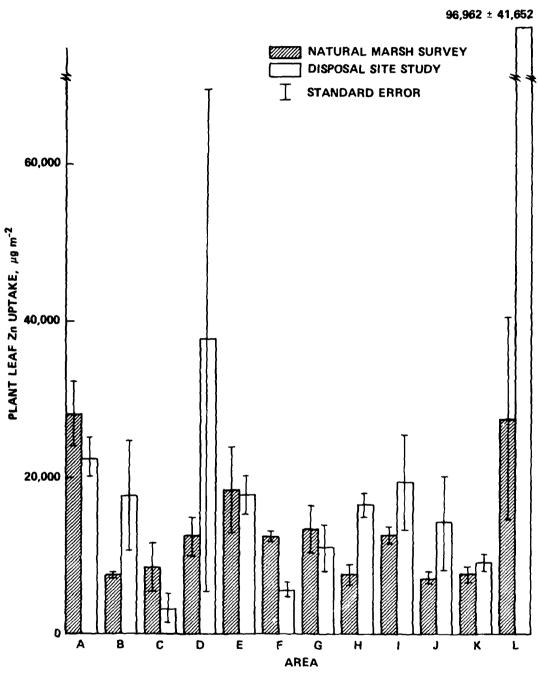
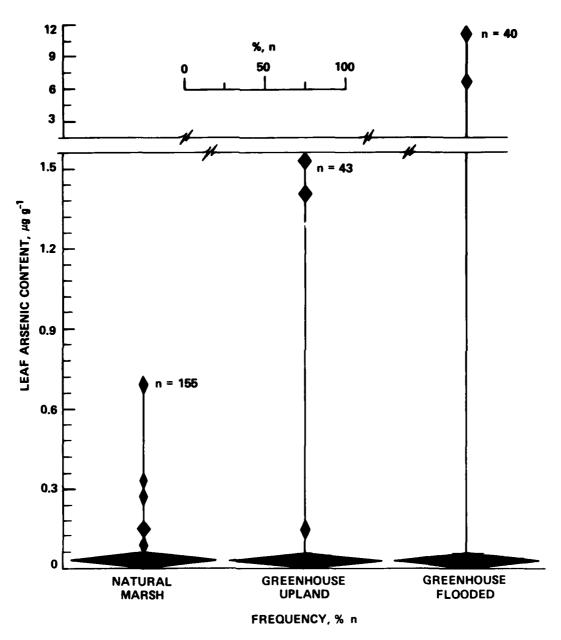


Figure 25. Total zinc uptake by S. altermiflora in natural marshes and disposal sites

## Heavy metal content of Cyperus tissue

- 41. The leaf heavy metal concentrations of the morphologically similar species of Cyperus, C. esculentus, C. odoratus (= feruginescens), C. strigosus, and C. Englemanni (Appendix G) collected from natural freshwater marsh areas were compared with those of C. esculentus var. sativus grown in the greenhouse study. The results of the previous greenhouse study indicated that, under upland (oxidized) conditions, C. esculentus had higher concentrations of Zn, Cd, Cu, Mn, and Pb, while Fe content was highest under flooded or reduced conditions. Concentrations of Ni, Cr, and Hg were the same under both disposal conditions. The watering protocol utilized in the upland portion of the greenhouse study approximated the conditions prevalent in Cyperus habitats. Most of the samples collected in the field were collected from areas that flooded during rains and subsequently dried out. Additionally, the Cyperaceae are among the first species present in disturbed areas; therefore they are collected only from the edges of the natural marsh, often near man-made structures. For this reason, the samples collected in the field might be expected to have the highest of the natural marsh Zn, Cd, Cu, and Pb levels. There are no literature values other than those of the greenhouse study available for comparison with the heavy metal content data discussed herein.
- 42. The total uptake values of *Cyperus* aboveground tissue were calculated for comparison to those determined during the greenhouse study. These data, converted to milligrams per square metre, appear in the following discussion as Tables 8-17. The uptake of heavy metals by *Cyperus* during the greenhouse study generally appears greater than that noted in the natural marsh. This phenomenon is directly related to the greater biomass developed under the greenhouse conditions, the nutrient contents of the sediment, and the absence of the competition from other early successional plants.
- 43. Arsenic. The plant leaf tissue content of As was never greater than 0.35  $\mu g g^{-1}$  and usually was below detection levels (Figure 26). This is essentially the same distribution found in the



A SEA COMMISSION OF THE PARTY O

Figure 26. Distribution of arsenic concentrations, Cyperus species

greenhouse study. Arsenic contents of plants grown under flooded conditions in the greenhouse were as high as 14  $\mu g \, e^{-1}$  on sediments highly contaminated with 316  $\mu g \, e^{-1}$  total As. Arsenic uptake by *Cyperus* was negligible (Table 8). This may have been the result of volitization of As during the hot acid digestion procedures prior to analysis.

144. <u>Cadmium.</u> Cadmium concentrations were found to be considerably lower than those of the greenhouse upland plants and slightly less than those of the greenhouse flooded group. As the field collections were made principally from areas presumed to be oxidized substrate similar to the greenhouse upland, the contrast is greater (Figure 27). No collection sites were found to contain consistently high Cd levels. Most samples contained less than 0.75 μg g<sup>-1</sup> and none exceeded 3.50 μg g<sup>-1</sup> Cd. The total uptake of Cd by the natural marsh *Cyperus* was usually less than 0.10 mg m<sup>-1</sup> and invariably less than that of the greenhouse study (Table 9). Flant uptake of Cd under greenhouse conditions could be expected to be greater since the growth environment was free from competitive plants and other adverse growing conditions that could occur in the field. These data indicate that plant uptake of Cd from dredged material should be of concern and should be monitored.

from 0.0 to 33.23 µg g<sup>-1</sup> (Figure 28). These levels were higher than those determined in *Cyperus* from either greenhouse disposal condition. The highest consistent levels were found in the MC area and at ME3 (Figure 7). There appeared to be no pattern in the location of the plants with higher Cr levels in the field. *Cyperus* leaves were rinsed with deionized reverse osmosis (RO) water before processing. Leaf surface adsorbed Cr could account for the higher contents of leaf Cr in marsh samples versus the greenhouse-grown plants. Washing *S. alterniflora* leaves appeared to result in lower Cr content, but this difference was not statistically significant in this study. Chromium uptake was less than 1 mg m<sup>-2</sup> (Table 10). Greater plant biomass production in the greenhouse study would explain the larger Cr uptake in the greenhouse plants.

46. Copper. The mean Cu concentration in the natural marsh was

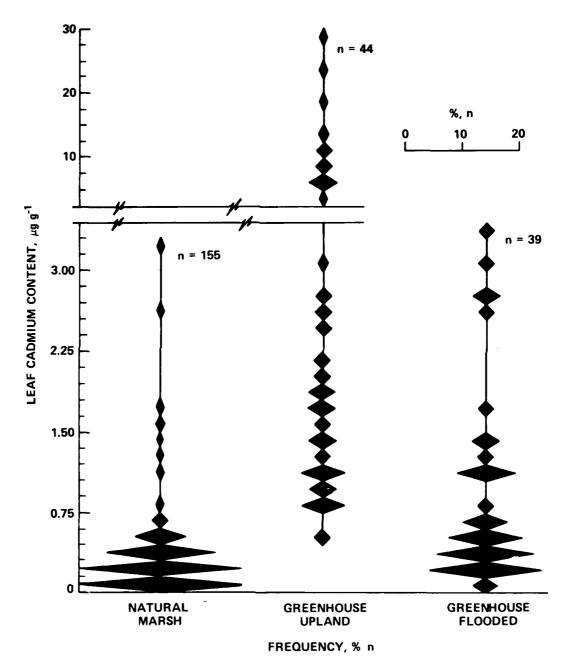


Figure 27. Distribution of cadmium concentrations, Cyperus species

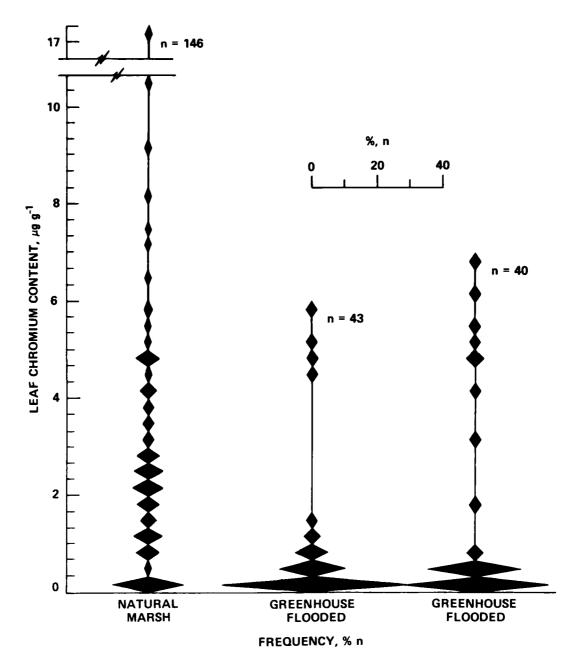


Figure 28. Distribution of chromium concentrations,  $\it Cyperus$  species

8.2 µg g<sup>-1</sup>, a value significantly higher than the means of 3.7 and 2.7 µg g<sup>-1</sup> for the upland and flooded greenhouse plants, respectively (Figure 29). While no area yielded plants with uniformly high Cu concentrations, the higher levels were usually associated with the MC collections (Figure 7). Collections from this area generally contained from 5.0 to 23 µg g<sup>-1</sup> Cu. The overall range of Cu for all Cyperus samples was 0.0 to 26.8 ½ g<sup>-1</sup> with no variation attributable to species differences. Total uptake of Cu was usually less than that measured in the greenhouse study (Table 11). The larger biomass production of greenhouse plants would explain the larger plant Cu uptake in the greenhouse study.

47. Iron. Iron concentrations ranged from 28 to 1873  $\mu g g^{-1}$  with a mean of 166  $\mu g g^{-1}$ . This concentration was higher than those of the greenhouse flooded and upland disposal environments (Figure 30). The concentrations of the samples at MCll (Figure 7) were the highest, ranging from 498 to 1893  $\mu g g^{-1}$ . There were no obvious sources of Fe at the site such as refuse. Iron total uptake values ranged from 1 to 36 mg m<sup>-2</sup>. This range is similar to that of the greenhouse upland condition, 1 to 48 mg m<sup>-2</sup>, and the greenhouse flooded, 0.5 to 35 mg m<sup>-2</sup> of Table 12.

48. Lead. The mean of the Pb concentrations in the natural marsh Cyperus was 6.07  $\mu g$  g<sup>-1</sup>, representing a range of 0.0 to 85.20  $\mu g$  g<sup>-1</sup>. These levels were significantly greater than either the greenhouse flooded or upland means of 1.05 and 1.20  $\mu g$  g<sup>-1</sup>, respectively (Figure 31). No field sites were associated with either consistently high levels, over 5  $\mu g$  g<sup>-1</sup>, or low levels, less that 5  $\mu g$  g<sup>-1</sup>. While the helicopter with nonleaded fuel should have eliminated the possibility of additional airborne Pb contamination of plant leaves, the deionized RO water rinse may not have removed previous airborne Pb contamination of collected leaves. This could explain the higher Pb content in field-collected leaves when compared to the greenhouse plants. Total uptake of Pb was generally less in natural marsh Cyperus than either upland or flooded greenhouse values (Table 13). The greater biomass production of greenhouse-grown plants would explain the larger total uptake of Pb in the greenhouse study plants.

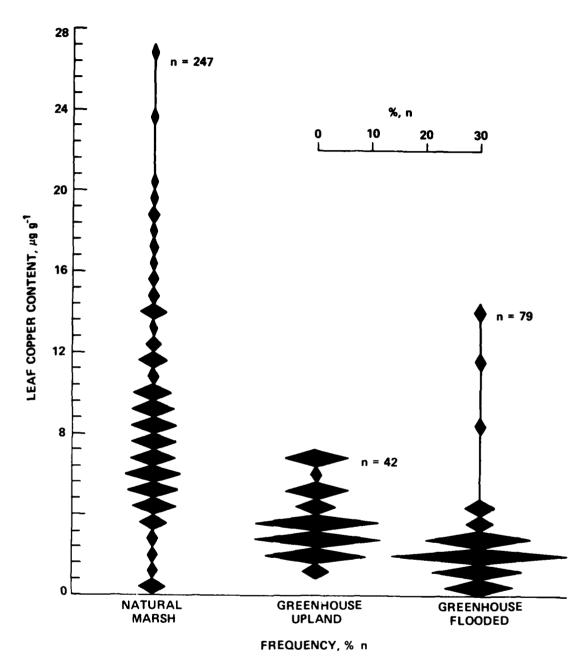


Figure 29. Distribution of copper concentrations, *Cyperus* species

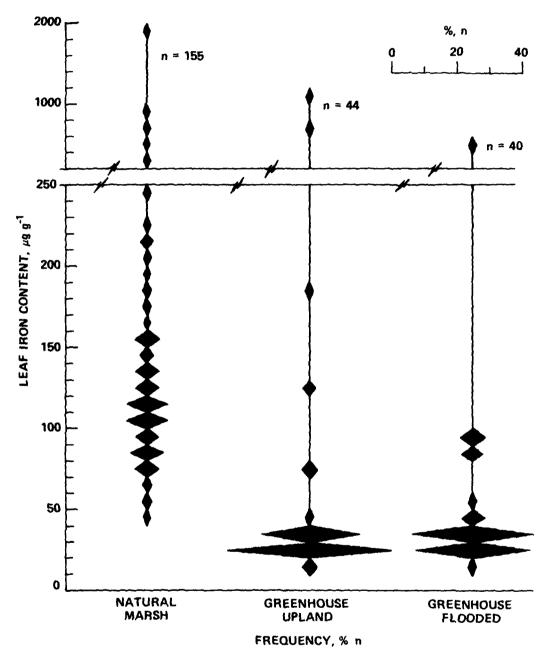


Figure 30. Distribution of iron concentrations, Cyperus species

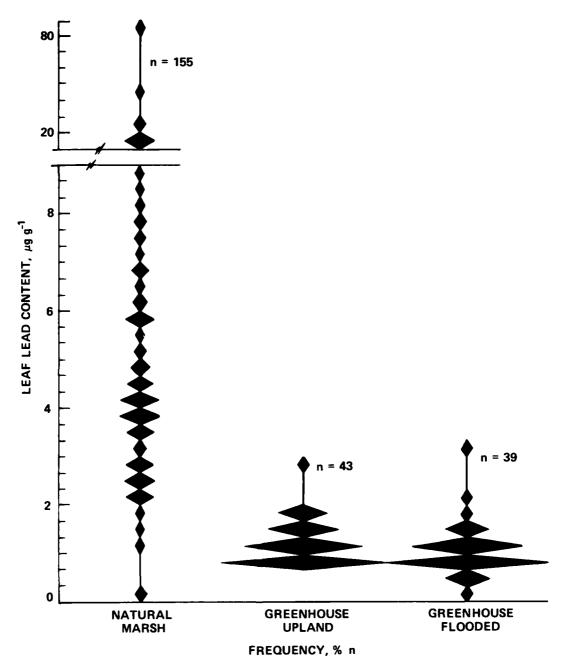


Figure 31. Distribution of lead concentrations, Cyperus species

- 49. Manganese. In contrast to Cr, Cu, Fe, and Pb, Mn leaf tissue concentration was lower in the natural marsh than in either treatment in the greenhouse study (Figure 32). The range of Mn concentrations in natural marsh Cyperus was 19 to 551 μg g<sup>-1</sup>, with a mean of 158 μg g<sup>-1</sup>. Manganese concentrations were not uniformly high or low at any location. In the natural marsh, there was a reversal of the Fe-Mn ratio of the greenhouse study, or the Mn concentrations were lower than those of Fe. The total uptake values for Mn in natural marsh Cyperus ranged from 0.6 to 50 mg m<sup>-2</sup>. These total uptake values are generally lower than those of either upland or flooded conditions in the greenhouse study treatments (Table 14).
- 50. Mercury. Concentration of Hg in natural marsh *Cyperus* seldom exceeded 0.02  $\mu$ g g<sup>-1</sup>, a value slightly lower than either of the green-house treatments (Figure 33). Sites DE1, DE2, and DE3 (Figure 7) appeared to have the highest levels of Hg, up to 2.0  $\mu$ g g<sup>-1</sup>. These low values for Hg may be a result of gaseous loss of Hg during the hot acid digestion of the plant material. Total uptake of Hg by *Cyperus* from the natural marsh was negligible (Table 15).
- 51. Nickel. The Ni concentrations in the leaf tissue of Cyperus from the natural marsh ranged from 0 to 14.1 µg g<sup>-1</sup> (Figure 34). The distribution of values was relatively uniform and most values were significantly higher than the values reported in the greenhouse study. No single area yielded plants with uniformly high or low concentrations. The deionized RO water rinse may not have washed airborne particulate Ni contamination from the plant leaves and therefore could explain the higher leaf Ni contents from the field collections. Biomass production and Ni uptake of the greenhouse plants was usually greater than that of the field-grown plants (Table 16).
- 52. Zinc. Zinc concentrations in the natural marsh *Cyperus* ranged from 0 to 317  $\mu g \, g^{-1}$  (Figure 35). The mean of the natural marsh plant concentrations, 79  $\mu g \, g^{-1}$ , was less than that of the greenhouse upland plants, 122  $\mu g \, g^{-1}$ , and slightly higher than that of the greenhouse flooded plants, 67  $\mu g \, g^{-1}$ . Since the field substrate conditions were presumed to be similar to the conditions in the greenhouse upland

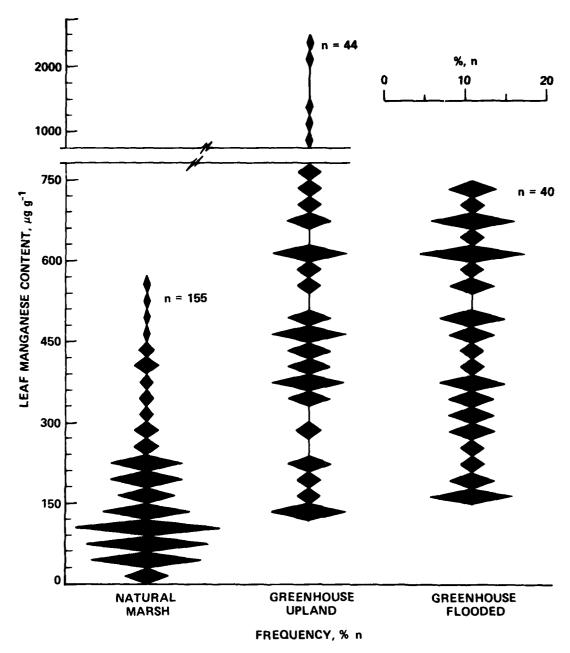


Figure 32. Distribution of manganese concentrations, *Cyperus* species

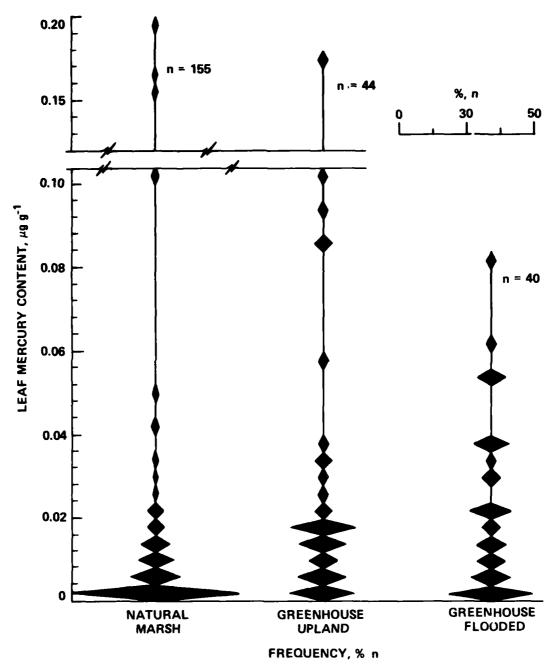


Figure 33. Distribution of mercury concentrations, *Cyperus* species

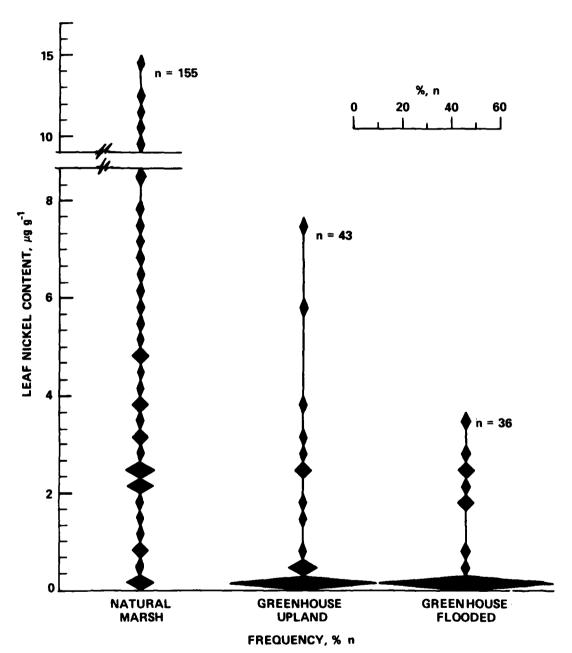


Figure 34. Distribution of nickel concentrations, Cyperus species

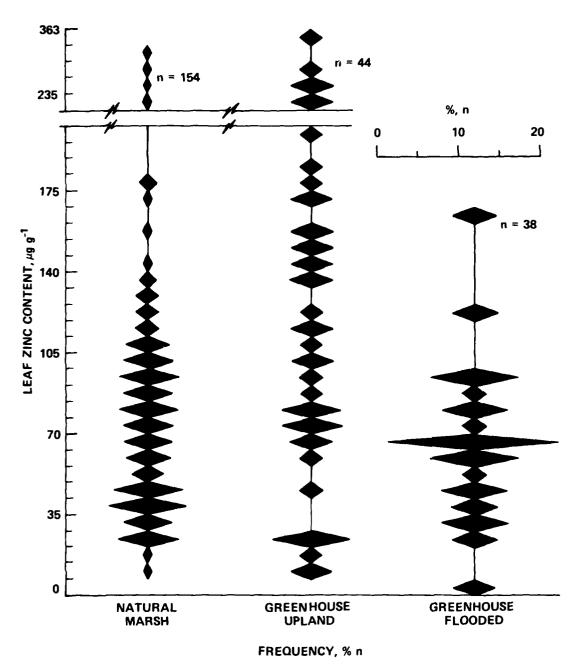


Figure 35. Distribution of zinc concentrations, Cyperus species

environment where Zn uptake was found to be greater, it appears that Zn concentrations may be lower in the natural marsh. Uniformly high levels in the vicinity of 100 to 200 µg g<sup>-1</sup> were found in plants collected at MC11 and MC12 (Figure 7). There were no obvious factors influencing the elevated Zn levels in these natural marsh areas. The total uptake of Zn by Cyperus aboveground tissue was generally lower in the natural marsh (Table 17). Sites MC11 and MC12, two field sites exhibiting high Zn concentration, did not show correspondingly high plant uptake values. These data indicate that the high Zn contents were related to small plants as a concentration effect due to plant size. Since the greenhouse plants produced more biomass than field-grown plants, it follows that Zn uptake was greater in the greenhouse plants.

## PART IV: CONCLUSIONS AND RECOMMENDATIONS

- 53. Heavy metal concentrations in marsh plants sampled in natural stands were quite similar in range to those concentrations observed in plants collected from CE disposal sites and those grown in contaminated sediments in the greenhouse. Exceptions were the lower Mn and Zn contents of S. alterniflora sampled in the natural saltwater marsh and lower Mn and Fe contents of Cyperus species sampled in the natural freshwater marsh. These data indicate that, if a contaminated dredged material is placed in a flooded disposal environment, the marsh plants colonizing that area should contain heavy metals in levels similar to those of natural marsh plants in the vicinity of the dredged material disposal.
- 54. A comparison of total plant heavy metal uptake values of natural saltwater marsh plants with those of disposal site plants of the same geographical area indicated a generally higher Cd total uptake in the disposal site plants. This is related to the larger biomass production on the disposal sites versus the natural marsh. The reader must remember that Cd contents and uptakes have been compared for different years and the conclusions may not be true if plants were collected from disposal sites and their associated natural marshes during the same growing season.
- 55. Additional research is needed to more accurately establish the relationship among heavy metal uptake by marsh plants growing in disposal sites and in adjacent natural marshes during the same growing season of the same year. Both saltwater and freshwater environments need to be studied.
- 56. This report conclusively indicates that dredged material containing elevated Cd concentrations should not be disposed in an upland environment, but rather in a flooded (reduced) environment. Placing a Cd-enriched dredged material in a flooded environment should result in marsh plants containing Cd levels equal to or less than those corresponding values of adjacent, naturally occurring marsh plants. Consequently, the environmental impact of disposing of a Cd-enriched dredged material should be minimized under such a flooded disposal environment.

## REFERENCES

- American Public Health Association. 1976. Standard methods for the examination of water and wastewater. 14th ed., Washington, D. C.
- Bingham, F. T. et al. 1976. Cadmium availability to rice in sludge-amended soil under "flood" and "non flood" culture. Soil Sci. Soc. Am. J. 40:715-719.
- Broome, S. W., W. W. Woodhouse, and E. D. Seneca. 1973. An investigation of propagation and the mineral nutrition of *Spartina alterniflora*. Sea Grant Publication UNC-SC-73-14.1, North Carolina State University, Raleigh, N. C.
- Court, A. 1974. The climate of the counterminous United States. Pages 193-343 in H. E. Landsberg, ed. World survey of climatology, Vol II, Climates of North America. Elsevier, New York, N. Y.
- Drifmeyer, J. E., and W. E. Odum. 1975. Lead, zinc, and manganese in dredged material pond ecosystems in Virginia. Environmental Conservation 1(2):1-7.
- Dunstan, W. M., and H. L. Windom. 1975. The influence of environmental changes in heavy metal concentrations of *Spartina alterniflora*. Pages 393-404 in L. E. Cronin ed. Estuarine research, Vol II, Geology and engineering. Academic Press, N. Y.
- Elias, R. W., and C. C. Patterson. 1975. Lead aerosol deposition on plant surfaces. Abstracts 26th annual meeting of Biological Societies, Corvallis, Oreg.
- Fernald, M. L. 1950. Gray's manual of botany. 8th ed., corrected printing 1970. Van Nostrand Co., N. Y.
- Folsom, B. L., Jr., C. R. Lee, and D. J. Bates. 1980. Influence of disposal environment on availability and plant uptake of heavy metals in dredged material. Technical Report (in press). U. S. Army Engineer Waterways Experiment Station, CE, Vicksburg, Miss.
- Gleason, H. A. 1952. The new Britton and Brown illustrated flora of the Northeastern United States and adjacent Canada. The New York Botanical Garden, N. Y.
- Gosselink, J. G., C. S. Hopkinson, and R. T. Parrondo. 1977. Common marsh plant species of the gulf coast area. Vols I and II. Technical Report D-77-44. U. S. Army Engineer Waterways Experiment Station, CE, Vicksburg, Miss.
- Jugsujinda, A., and W. H. Patrick, Jr. 1977. Growth and nutrient uptake by rise in a flooded soil under controlled aerobic-anaerobic and pH conditions. Agron. J. 69:705-710.
- Lee, C. R., T. C. Sturgis, and M. C. Landin. 1976. A hydroponic study of heavy metal uptake by selected marsh plant species. Technical Report D-76-5. U. S. Army Engineer Waterways Experiment Station, CE, Vicksburg, Miss.

Lee, C. R., T. C. Sturgis, and M. C. Landin. 1978. Prediction of heavy metal uptake by marsh plants based on chemical extraction of heavy metals from dredged material. Technical Report D-78-6. U. S. Army Engineer Waterways Experiment Station, CE, Vicksburg, Miss.

Marcks, B. G. 1974. Preliminary reports on the flora of Wisconsin. No. 66. Cyperaceae II - Sedge family II. The genus Cyperus - the umbrella sedges. Trans. Wis. Acad. Sci. Arts Lett. 62:261-284.

Mohlenbrock, R. H. 1960. The Cyperaceae of Illinois. I. Cyperus. Am. Mid. Nat. 63:270-306.

Reddy, C. N., and W. H. Patrick, Jr. 1977. Effect of redox potential and pH on the uptake of cadmium and lead by rice plants. J. Environ. Qual. 6:259-262.

Swink, F. A., and G. S. Wilhelm. 1979. Plants of the Chicago region. Morton Arboretum, Lisle, Ill.

U. S. Department of Commerce. 1977. Tide tables 1978. National Ocean Survey, Rockville, Md.

Voss, E. G. 1972. Michigan flora, Part 1, Gymnosperms and monocots. Cranbrook Institute of Science, Bloomfield Hills, Mich.

Water Information Center, Inc. 1974. Climates of the states, Vol I. Port Washington, N. Y.

Water Resources Support Center. 1979. Summary of activities - Corps and industry - dollars and yardage (millions). Dredging Division, U. S. Army Engineer Water Resources Support Center, CE, Ft. Belvoir, Va.

Williams, B., and M. B. Murdoch. 1969. The potential importance of Spartina alterniflora in conveying zinc, manganese, and iron into estuarine food chains. Pages 431-439 in D. J. Nelson and J. C. Evans eds. Proc. 2nd Nat. Sym. Radioecology. NBS, Springfield, Va.

Table 1
Techniques Used in the Analysis of Nitric Acid Digests

Chemical Species	Procedures and/or Instrumentation	Lowest Reporting Concentration µg g <sup>-1</sup>
Zn <b>*</b>	Determined with a Spectrametrics Argon Plasma Emission Spectrophotometer Model II	0.1000
Cd*		0.1000
Cu*		0.1000
Fe*		0.1000
Mn*		0.1000
Ni*		0.1000
Cr*		0.1000
Pb*		0.1000
As	Determined with a Nisseisangyo Isotope Shift Zeeman Effect Atomic Absorption Spectrophotometer	0.0010
Hg	Cold Vapor Flameless Atomic Absorption Standard Methods**	0.0002

\*\* American Public Health Association (1976).

<sup>\*</sup> Determined with a Perkin-Elmer Heated Graphite Atomizer Absorption Unit to reach (all in µg g<sup>-1</sup>) 0.0001 Cd, 0.004 Zn, 0.001 Cu, 0.0005 Fe, Mn, and Pb, 0.003 Ni, 0.001 Cr.

Table 2

Comparison of Leaf Heavy Metal Contents of Flooded

and Upland Collections of S. alterniflora

	Mean Leaf Concentration, µg g _		Level of
Metal	Flooded	Upland	Significance
Arsenic	0.208	0.0134	ns*
Cadmium	0.08	0.07	NS
Chromium	2.49	2.38	NS
Copper	3.1	3.5	ns
Iron	128	140	NS
Lead	2.1	2.6	NS
Manganese	61	72	NS
Mercury	0.015	0.016	NS
Nickel	1.7	2.2	NS
Zinc	22	34	NS

<sup>\*</sup> NS = Not significant.

Table 3

Comparison of Leaf Heavy Metal Contents of Washed and
Unwashed Portions of Samples of S. alterniflora

Arsenic 0.009 Cadmium 0.10 Chromium 2.21 Copper 4.1 Iron 191	0.021 0.23 2.53	Significance NS* 0.05
Cadmium 0.10 Chromium 2.21 Copper 4.1 Iron 191	0.23	
Chromium 2.21 Copper 4.1 Iron 191	_	0.05
Copper 4.1 Iron 191	2.53	
Iron 191 1	2.73	ns
•	3.9	NS
	.43	NS
Lead 2.4	2.1	ns
Manganese 72	75	ns
Mercury 0.016	0.019	ns
Nickel 1.9	1.7	NS
Zinc 25	32	ns

<sup>\*</sup> NS = Not significant.

Table  $^{\mu}$  Areas of Comparison of Natural Marsh Survey and Disposal Site Study of

Heavy Metal Concentration and Uptake for S. alterniflora

Area*	North Latitude	West Longitude	Natural Marsh Site	Disposal Site**
Ą	41,000,-41,030,	72040'-73010'	NY1-NY6	1-1
щ	39°01'-39°44'	74040,-74042,	NY10-NY12	2-1, 2-3
ಬ	38°20'-38°50'	70°10'-76°30'	BM7	3-2-3-6
Д	33°30'-34°00'	77°56'-78°32'	JV10, JV11	4-1, 4-2
Ħ	31°50'-32°10'	80°50'-81°10'	$JV^{h}$ , $JV5$	5-1
[±4	31°01'-31°53'	81,01,-81,040	JV3, JV4	5-2
೮	30°20'-31°05'	81°25'-81°35'	JV1-JV3	5-2, 3, 4, 6-1-6-3
Ħ	29°13'-29°26'	89°33'-90°57'	NO6-NO10	8-1
н	29°10'-29°15'	,25,06-,70,06	NO8-NO10	8-2
Ь	29°01′-29°54′	93°56′-94°53′	cc13-cc15	10-1-10-3
×	28010'-28040'	.15,96-,90,96	CC7-CC11	11-1, 11-2
ы	27°40'-28°05'	97°20'-97°30'	900-100	11-3

\* See Figure 8 for locations of designated areas. \*\* Locations in code used in Lee, Sturgis, and Landin (1978).

Table 5

Mean Yield of Aboveground Tissue of S. alterniflora
on Natural Marsh and Disposal Sites

	Natural Marsh,	Disposal Site,
Area*	1978, $g m^{-2}$	1975, g m <sup>-2</sup>
A	627	1095
В	718	801
С	716	214
D	1078	1659
E	1585	1106
F	1860	691
G	1184	904
Н	649	1044
I	577	1250
J	438	658
К	613	774
L	505	699

<sup>\*</sup> See Figure 8 for locations of designated areas.

Table 6

Comparison of Heavy Metal Contents of Gratand Quadrat Samples of Cyperus

	Mean Leaf Conce	entrations, µg g-l	$\alpha = 0.05$ Level of
<u>Metal</u>	Grab	Quadrat	Significance
Arsenic	0.032	0.026	ns*
Cadmium	0.40	0.35	ns
Chromium	2.68	2.40	ns
Copper	7.6	8.4	ns
Iron	158	168	ns
Lead	4.7	6.4	ns
Manganese	145	161	ns
Mercury	0.012	0.016	ns
Nickel	3.5	3.5	ns
Zinc	83	79	ns

<sup>\*</sup> NS = Not significant.

Table 7

Percent of Cyperus by Weight
in Quadrat Biomass

Site	Percent
DE1	26.9
DE2	68.6
DE3	54.9
DE4	49.6
DE5	54.5
DE6	40.0
DE7	43.5
DE8	44.4
DE9 IN1 MC1 MC2	25.8 23.6 53.7 66.7
MC3	59.3
MC4	43.6
MC5	45.3
MC6	39.8
MC7	50.7
MC8	47.5
MC9	63.8
MC10	63.2
MC13 MC13	39.1 19.6 27.4 19.1
ME2	22.3
ME3	10.5
ME4	53.1
ME5	31.6
ME6	53.3
MW1	42.4
MW2	40.4

Table 8

Total Uptake of Arsenic by Aboveground

Tissue of Cyperus Species

	Tota	l Uptake, mg m <sup>-2</sup>	
			house
Location*	Natural Marsh	<u>Upland</u>	Flooded
DEl	<0.01	<0.01	<0.01
2	1	<0.01	<0.01
3 4		<0.01	<0.01
4	l		
5 6	j		
6			
7	ì		
8			
9			
IN1	i i	<0.01	<0.01
2	<u>*</u>	<0.01	<0.01
3		<0.01	0.05
MCl	<0.01	<0.01	<0.01
2	10.01	<0.01	<0.01
3	1	<0.01	<0.01
3 4	i		
5	Į.		
5 6	\ \		
7	j		
8			
9	1		
10	l l		
11	f		
12	)		
13	}		
MEl		<0.01	<0.01
2	İ	<0.01	<0.01
3	1	2.45	<0.01
4	V		
	0.09		
5 6	<0.01		
MWl	l l	<0.01	<0.01
2	1	<0.01	<0.01
3	<u> </u>	<0.01	<0.01

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

Table 9

Total Uptake of Cadmium by Aboveground

Tissue of Cyperus Species

	Tot	al Uptake, mg m <sup>-2</sup>	<u> </u>
			ihouse
Location*	Natural Marsh	Upland	Flooded
DEl	0.01	1.16	0.24
2	0.08	15.36	8.26
3	0.04	0.49	0.06
4	0.02		
5 6	0.02		
	0.01		
7	0.03		
8	0.03		
9	0.01		
INL	0.07	0.70	0.34
2	<b></b>	4.40	1.90
3		1.43	1.22
MCl	0.02	49.52	6.26
2	0.01	96.65	22.34
3	0.01	28.55	11.88
3 4	<0.01		
5 6	0.01	<b></b>	
	0.02		
7	0.02		
8	0.01		
9	0.04		
10	<0.01		
11	0.01		
12	0.03		
13	<0.01		
ME1	<0.01	2.47	1.41
2	<0.01	0.56	1.53
3 4	<0.01	16.28	0.05
	<0.01		
5	0.01	~~	
6	0.20		
MWl	0.01	53.22	7.39
2	0.01	3.21	0.94
3		1.92	0.70

يعا الإنتاج المتحاج المتحاجة والمعاد المالي

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

Table 10

Total Uptake of Chromium by Aboveground

Tissue of Cyperus Species

	Tot	al Uptake, mg m <sup>-2</sup>	
			house
Location*	Natural Marsh	Upland	Flooded
DEl	0.05	0.11	0.06
2	0.88	0.92	1.77
3 4	0.00	0.03	0.06
14	0.47		
5 6	0.39		
	0.16	<del></del>	
7	0.26		
8	0.03		
9	0.05	<del></del>	
INl	0.06	0.14	0.65
2	<b></b>	1.22	2.21
3		0.80	3.58
MCl	0.07	IS**	14.59
2	0.05	24.20	21.64
3	0.05	7.74	16.19
3 4	0.04	<del></del>	
5 6	0.04		
6	0.18		
7 8	0.15		
	0.13		
9	0.23		
10	0.07		
11	0.10		
12	0.04	<del></del>	
13	0.01		
MEl	0.03	0.21	0.99
2	0.09	0.58	0.57
3	0.09	0.99	0.11
4	0.22	<del></del>	
5 6	0.20	-	
6	0.74	~-	
MWl	0.17	0.62	26.72
2	0.04	0.41	0.18
3		1.12	0.53

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

<sup>\*\*</sup> IS = Insufficient sample for analysis.

Table 11

Total Uptake of Copper by Aboveground

Tissue of Cyperus Species

	То	tal Uptake, mg m <sup>-2</sup>	· ·
			house
Location*	Natural Marsh	Upland	Flooded
DE1	0.54	1.05	1.27
2	1.09	6.55	9.52
3 4	0.50	1.37	0.48
4	0.67		
5 6	0.82		
6	0.68		
7 8	1.08		
	0.83		
9	0.29		
INl	0.27	1.55	1.91
2		3.98	0.86
3		2.96	9.93
MCl	0.37	16.39	7.74
2	0.33	27.85	18.21
2 3 4	0.32	13.59	12.63
4	0.22		
5 6	0.22		
6	0.70		
7 8	0.77		
	0.51		
9	1.38		
10	0.61		
11	0.35		
12	0.29		
13	0.29		
ME1	0.14	5.52	14.00
2	0.13	1.00	5.96
3 4	0.29	7.82	0.72
4	0.24		
5 6	0.23		
6	1.80		
MWl	0.15	27.52	21.18
2	0.04	5.07	6.07
3		4.76	5.11

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

Table 12

<u>Total Uptake of Iron by Aboveground</u>

<u>Tissue of Cyperus Species</u>

	Tot	al Uptake, mg m <sup>-2</sup>	
			house
Location*	Natural Marsh	Upland	Flooded
DE1	10.30	1.80	0.90
2	35.71	5.86	32.46
3 4	12.50	0.90	0.90
4	9.98		
5 6	11.52		
	10.10		
7 8	13.51		
	11.51	<b></b>	
9	4.47		
INI	4.29	1.35	4.50
2		3.16	11.27
3		39.67	33.81
MCl	7.12	9.47	16.68
2	5.37	22.54	27.95
2 3 4	4.72	32.01	34.72
	3.87		
5 6	3.32		
	7.28		
7	6.28		
8	13.61		
9	24.38	<b></b>	
10	13.01		
11	33.53		
12	1.46		
13	1.67		
ME1	3.92	4.06	4.50
2	2.13	3.16	5.41
2 3 4	2.64	4.50	0.45
4	7.31		
5	8.31		
6	29.47		
MWl	7.85	27.95	31.11
2 3	7.95	47.79	5.86
3		6.76	5.86

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

Table 13

Total Uptake of Lead by Aboveground

Tissue of Cyperus Species

	Tot	tal Uptake, mg m <sup>-2</sup>	
T M			house
Location*	Natural Marsh	Upland	Flooded
DEl	0.57	0.50	0.44
2	3.22	3.92	24.32
2 3 կ	1.09	0.47	0.26
<u>1</u> ,	0.35		
5 6	0.66		
	0.49		
7	0.63		
8	0.65		
9	0.19		
IN1	0.14	0.35	1.19
2		1.37	2.85
3		1.27	4.13
	0.00		
MCl	0.29	3.23	3.18
2	0.25	5.65 3.16	10.90
2 3 4	0.15	3.10	5.50
	0.15 0.14	<del></del>	
5 6		<del></del>	
7	0.92		
7 8	0.30		<b></b>
9	0.21 2.00	<del></del>	
9 10	0.28		
11	0.20	<del></del>	
12	0.10		
13	0.14	<del></del>	
ME1	0.11	1.80	1.95
2	0.07	1.38	1.98
3 4	0.05	1.90	0.08
	0.18		
5 6	0.24		
6	1.01		
MWl	0.11	4.95	6.24
2	0.28	2.12	1.76
3		3.26	2.16

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

Table 14

Total Uptake of Manganese by Aboveground

Tissue of Cyperus Species

	Tota	1 Uptake, mg m <sup>-2</sup>	
			house
Location*	Natural Marsh	Upland	Flooded
DE1	6.22	6.3	5.86
2	50.22	158.70	207.39
3 4	13.60	16.23	18.94
4	8.09	~-	
5 6	10.49		
	4.31	~-	
7	3.17	~-	~-
8	10.08	~-	
9	6.12	~-	
INl	3.02	46.89	121.28
2	=-	317.85	233.54
3		50.95	131.65
	6.18		
MCl	6.80	99.19	243.91
2	2.83	271.87	296.21
3 4	4.10	154.19	130.30
4 C	4.10	<del></del>	
5 6	4.68		
7	4.53		<del></del>
7 8	13.77	<b></b>	
9	4.20		
10	11.95		
11	5.63	_ <del>_</del>	
12	0.59		
13	2.24		
ME1	14.00	44.18	121.73
2	7.63	60.41	47.79
3	6.58	185.75	6.76
3 4	12.17		
5	20.10	<del></del>	
6	41.18		
MWl	7.15	288.10	235.35
	7.49	89.27	237·37 136.61
2 3	1.47	30.21	104.15
ک	<del></del>	) ∪ • ₾.±	104.17

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

Table 15

Total Uptake of Mercury by Aboveground

Tissue of Cyperus Species

	Total	al Uptake, mg m <sup>-2</sup>	
T = 4 * <b>X</b>			ihouse
Location*	Natural Marsh	Upland	Flooded
DEl	<0.01	0.02	<0.01
2	0.03	0.02	0.41
3	0.02	<0.01	<0.01
4	<0.01		
5 6	<0.01		
6	<0.01		
7	<0.01		
8	<0.01		
9	<0.01		
INl	<0.01	<0.01	0.06
2		0.04	0.06
3	<del></del>	0.05	0.13
MCl	<0.01	0.20	0.06
2	<0.01	0.16	0.42
	<0.01	0.06	0.21
3 4	<0.01		
5	<0.01		
5 6	<0.01		
7	<0.01		
8	<0.01		
9	<0.01		
10	<0.01		
11	<0.01		
12	<0.01		
13	<0.01		
MEL	<0.01	0.01	<0.01
2	<0.01	0.01	<0.01
3 4	<0.01	0.03	<0.01
4	<0.01		
5	<0.01	<del></del>	
6	<0.01		
MWl	<0.01	0.05	0.02
2	<0.01	0.03	0.02
3		0.03	0.03

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

Table 16

Total Uptake of Nickel by Aboveground
Tissue of Cyperus Species

Location*	Total Uptake, mg m <sup>-2</sup>			
	Greenhouse			
	Natural Marsh	Upland	Flooded	
D <b>E1</b>	0.07	IS**	IS	
	0.11	0.81	4.77	
2 3 4	0.30	IS	IS	
	0.29			
5 6	0.18			
6	0.19			
7	0.30			
8	0.07	<del></del>		
9	0.07			
INl	0.20	IS	IS	
2		IS	IS	
3		3.65	0.56	
MCl	0.28	1.89	8.94	
2	0.28	11.93	20.63	
3	0.20	12.19	12.24	
3 4	0.15			
5 6	0.14			
6	0.21			
7 8	0.48			
	0.20			
9	0.20			
10	0.29			
11	0.15			
12	ರಂ.0			
13	0.11			
ME1	0.06	IS	IS	
2	0.08	IS	0.66	
3	0.13	5.71	IS	
14	0.10			
5 6	0.18			
6	0.49			
MWl	0.15	2.82	0.72	
2	0.01	IS	IS	
3		IS	0.21	

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

<sup>\*\*</sup> IS = Insufficient sample for analysis.

Table 17

Total Uptake of Zinc by Aboveground

Tissue of Cyperus Species

	Total Uptake, mg m <sup>-2</sup>			
	Greenhouse			
Location*	Natural Marsh	Upland	Flooded	
DEl	7.47	55.36	49.77	
2	18.59	591.52	986.38	
3	5.04	118.39	85.93	
14	4.31			
5 6	3.22			
	2.72			
7	8.88			
8	12.98			
9	4.33	- <del>-</del>		
IN1	4.09	40.39	62.39	
2		213.79	194.95	
3		194.95	321.32	
MCl	1.97	719.12	300.67	
2	1.31	781.78	561.31	
3	2.51	311.86	268.21	
14	1.18			
5 6	1.40			
6	5.27			
7	6.69			
8	3.35			
9	9.60			
10	3.77			
11	2.71			
12	3.88			
13	1.92			
ME1	3.11	40.44	51.67	
2	2.03	87.19	120.65	
3	1.34	27.99	4.50	
14	4.12	<del></del>		
5	5.06			
6	37.75			
MWl	3.05	1080.12	291.70	
2	0.69	147.66	204.91	
3		126.28	184.35	

<sup>\*</sup> Abbreviations are the same areas as those shown in Figure 7.

APPENDIX A: PHOTOGRAPHS OF SALTWATER AND FRESHWATER COLLECTION SITES

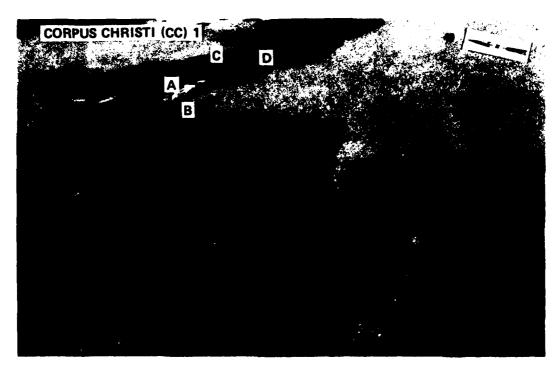


Photo 1. Site CC1

PHOTO NOT AVAILABLE ON CORPUS CHRISTI SITE:

Photo 2. Site CC2

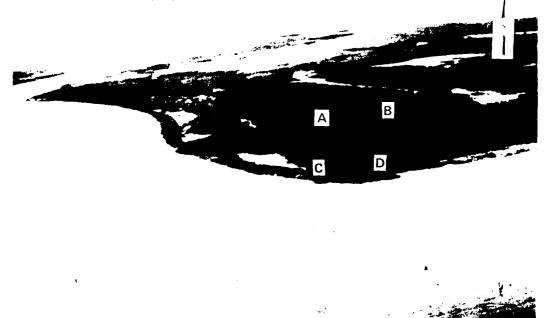


Photo 3. Site CC3



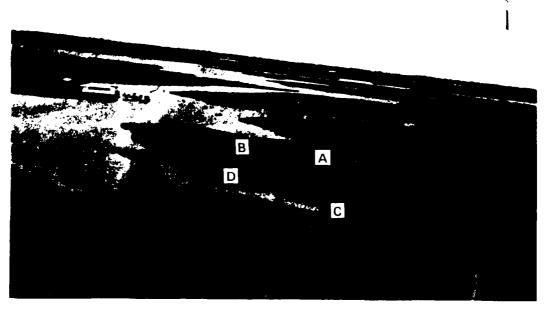


Photo 4. Site CC4

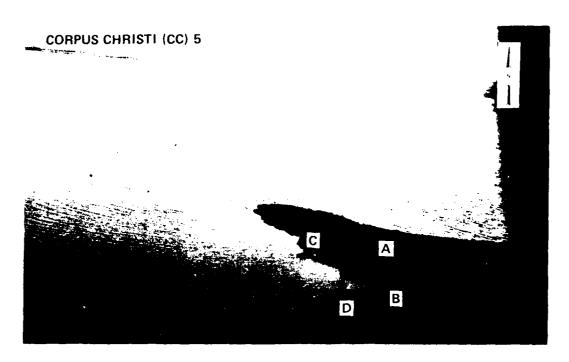


Photo 5. Site CC5

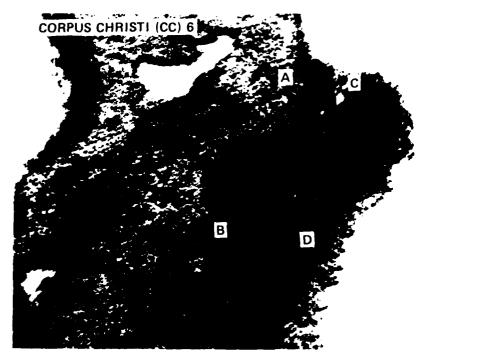


Photo F. Site 376

## CORPUS CHRISTI (CC) 7

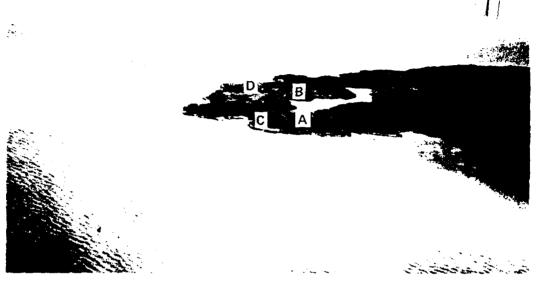


Photo 7. Site CC7

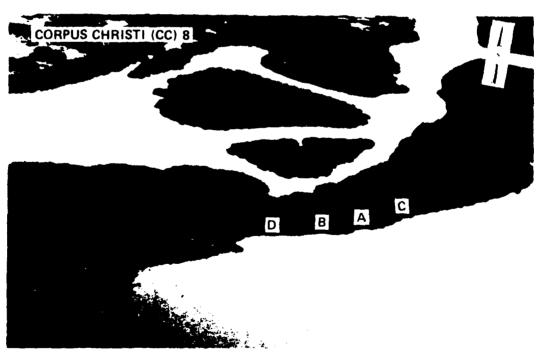


Photo 8. Site CC8

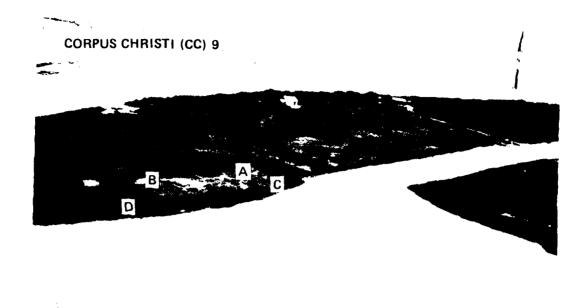


Photo 9. Site CC9

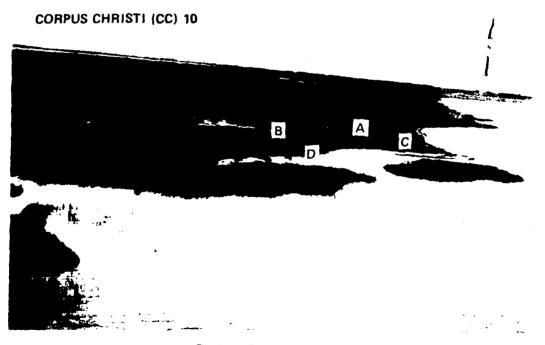
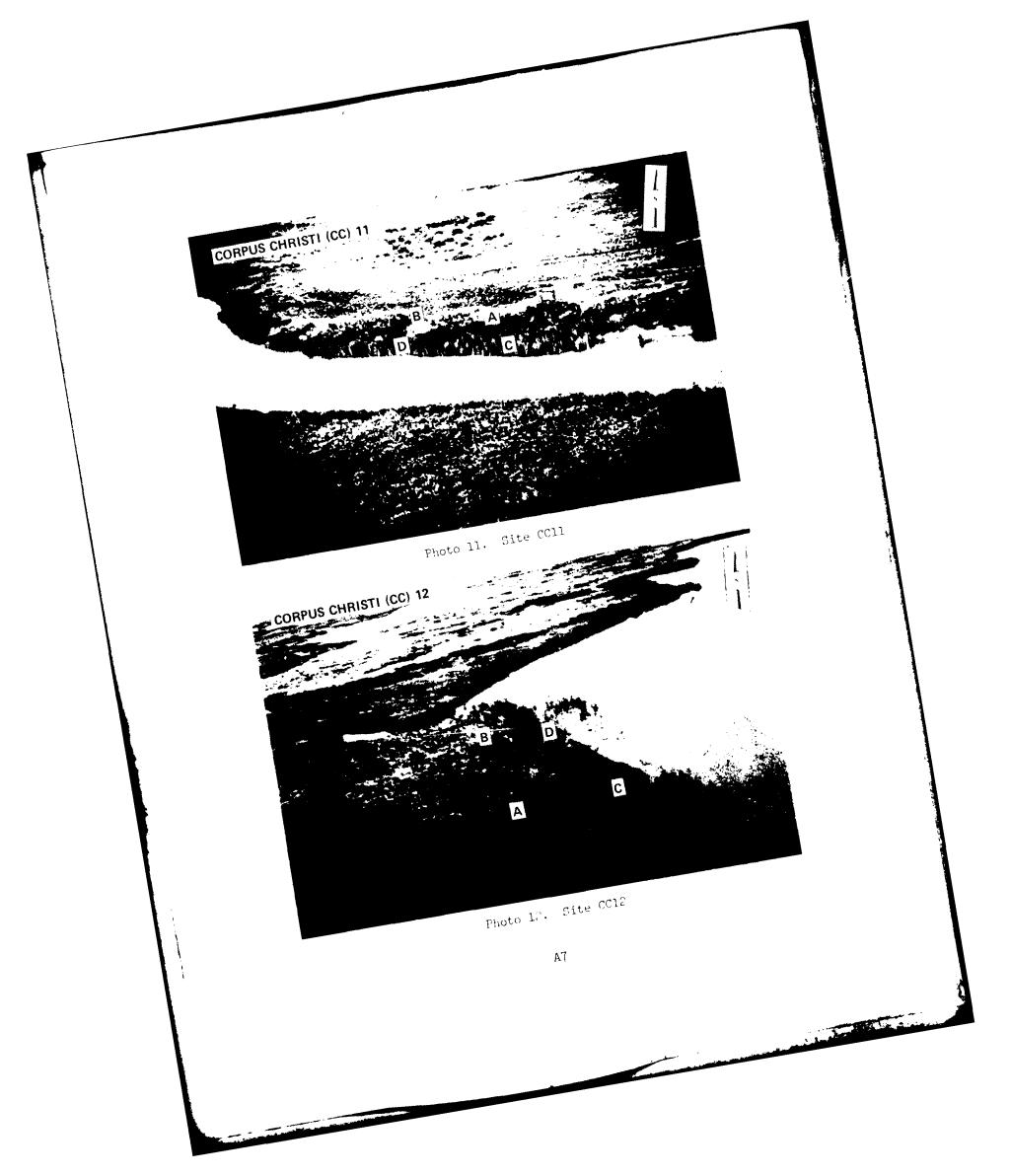


Photo 10. Site CC10



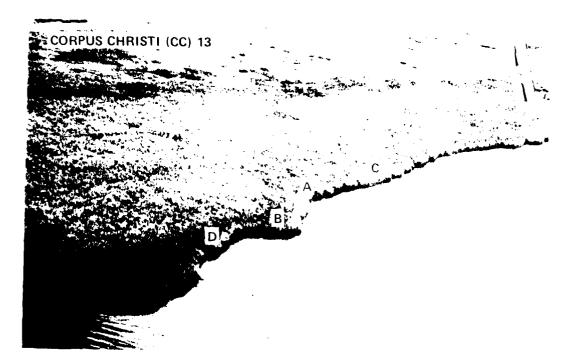
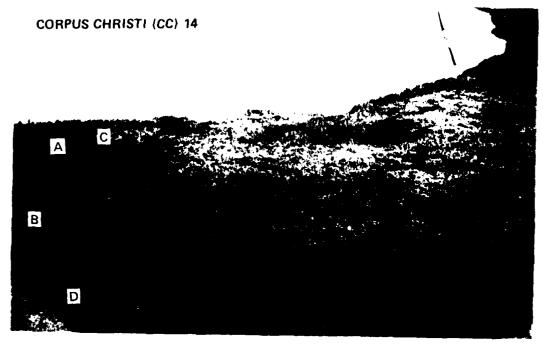


Photo 13. Cit. CCl ₹



Theta the miss out

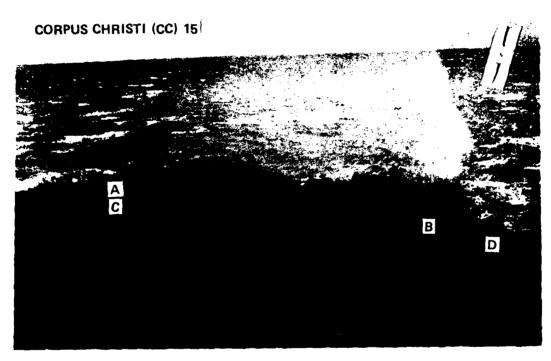


Photo 15. Site CC15

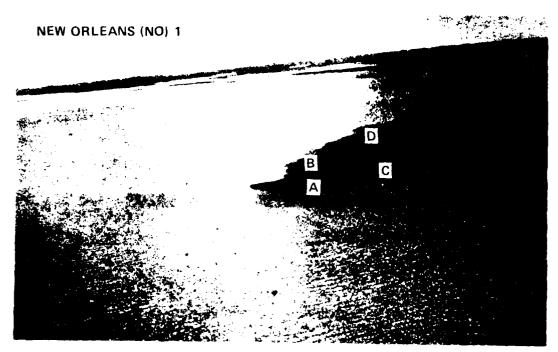
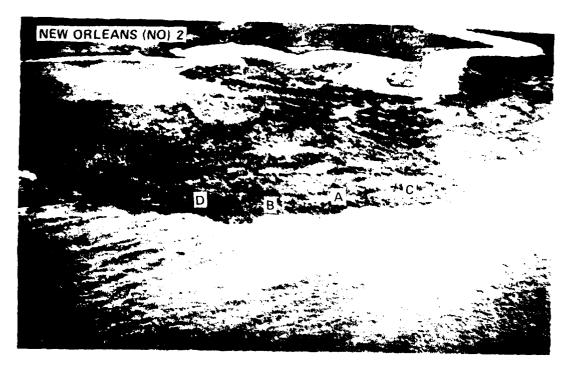


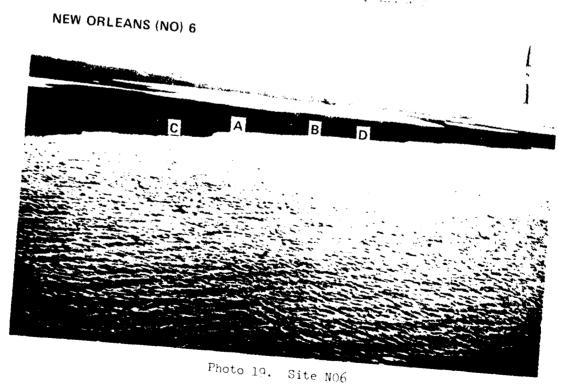
Photo 16. Site NO1



Inoto 17. Site NO:

THOUSE NOT AVAILABLE ON UNIVERSITY OF AUTOMOTION S. A. AUTO.

## that, in other ma, ma, and ma



AD-A101 662

ARMY ENGINEER WATERWAYS EXPERIMENT STATION VICKSBURG--ETC F/6 6/6
FIELD SURVEY OF HEAVY METAL UPTAKE BY NATURALLY OCCURRING SALTW--ETC(U)
JUN al J W SYMMERS, B L FOLSOW, C R LEE

WES/TR/EL-81-5

NL

END

END

END

END

BERN

BERN

END

BERN

BER

DTIC

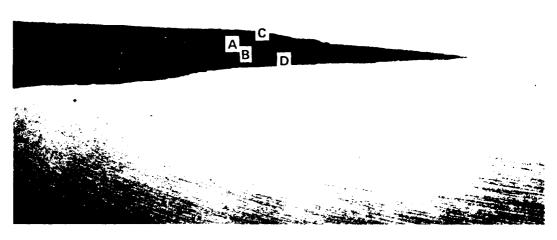


Photo 20. Site NU7

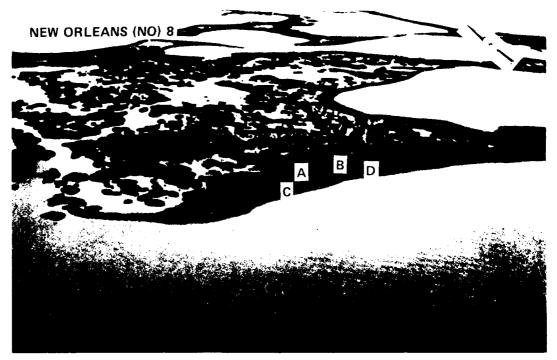


Photo 21. Site NOS

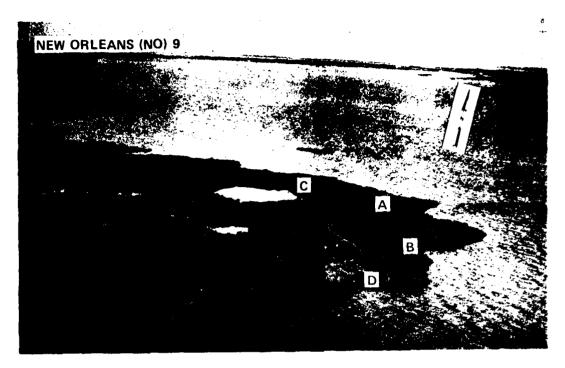


Photo 22. Site NO9

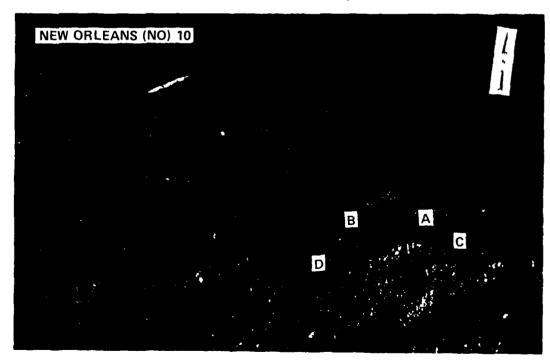


Photo 23. Site NC10

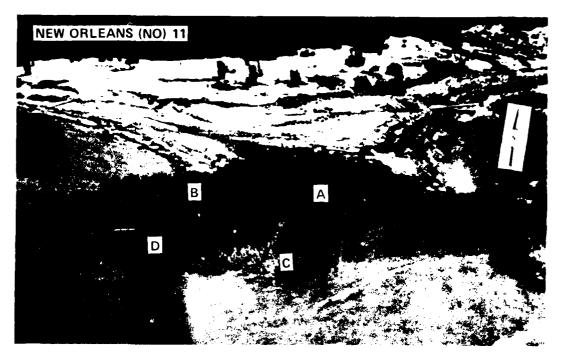


Photo 24. Site NO11

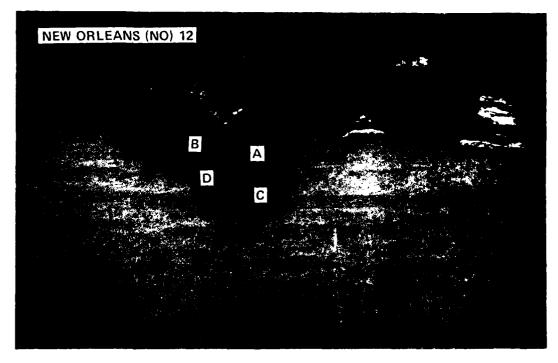


Photo 25. Site NOL2

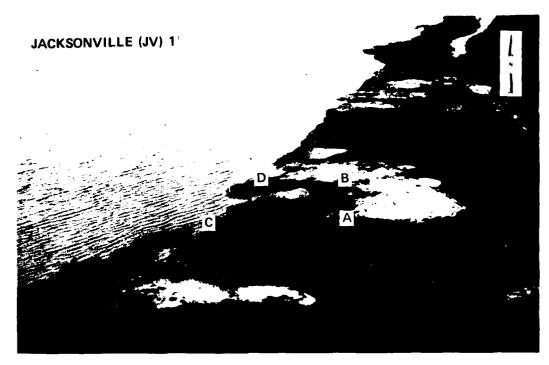


Photo 26. Site JVl

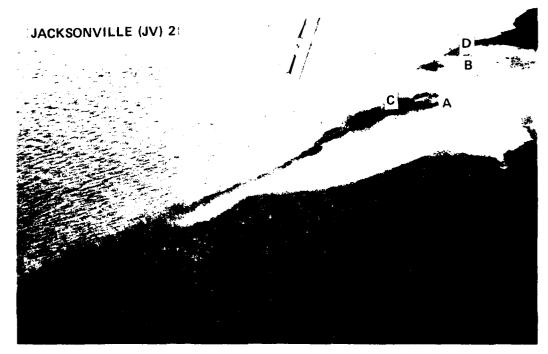


Photo 27. Site JV2

## JACKSONVILLE (JV) 3

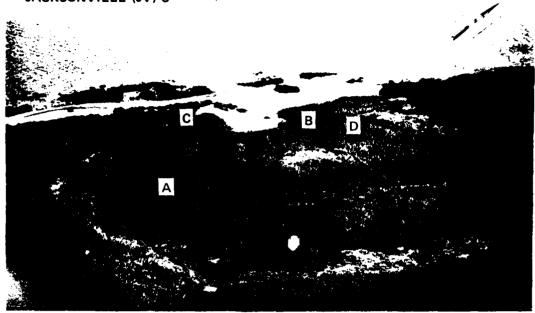


Photo 28. Site JV3

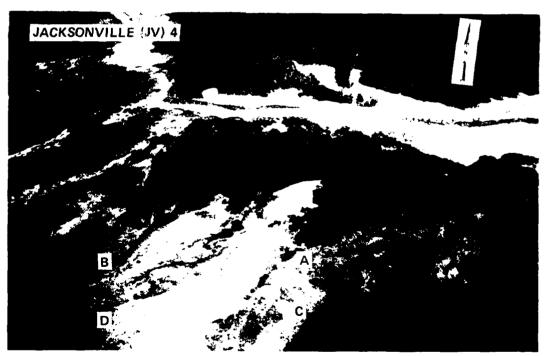


Photo 29. Site JV4

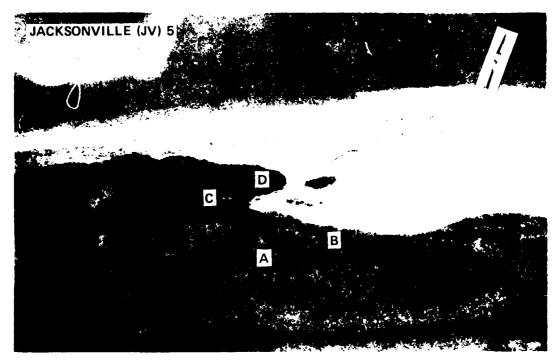


Photo 30. Site JV5

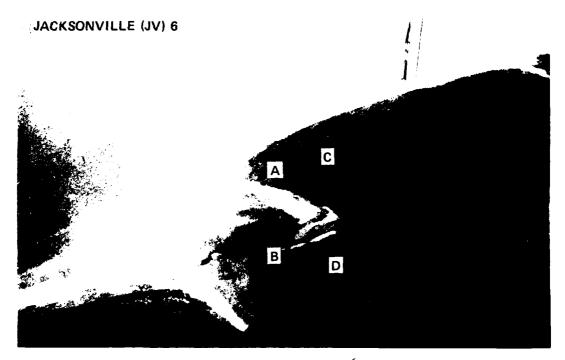


Photo 31. Site J 6

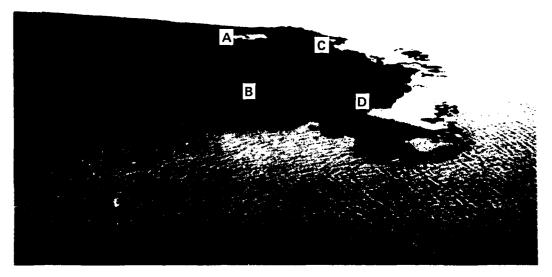


Photo 32. Site JV7

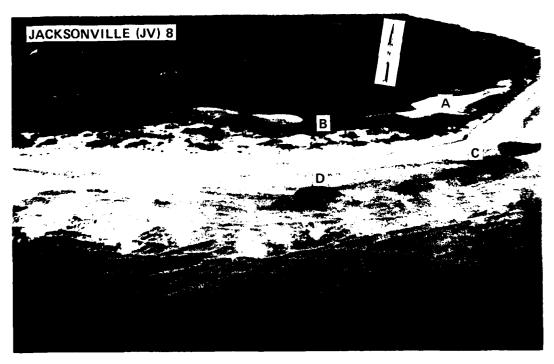


Photo 33. Site JV8

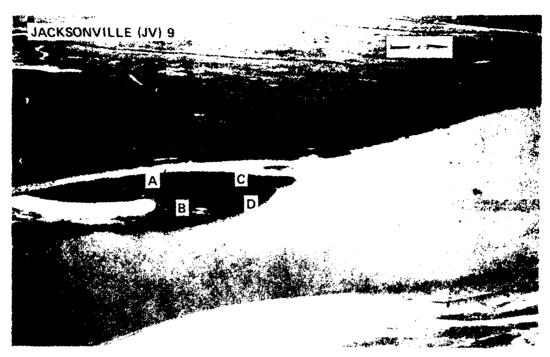


Photo 34. Site JV9



Photo 35. Site JV10

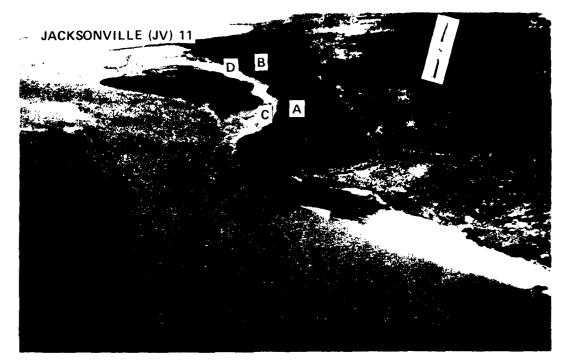


Photo 36. Cite JV11

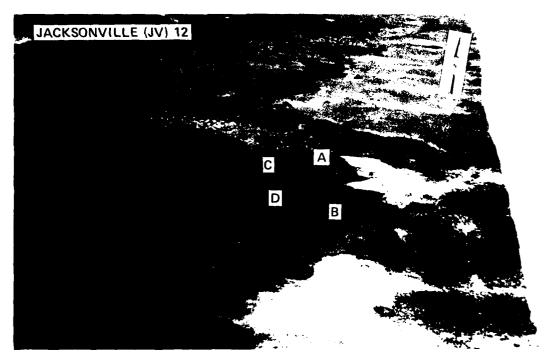


Photo 37. Site JV12

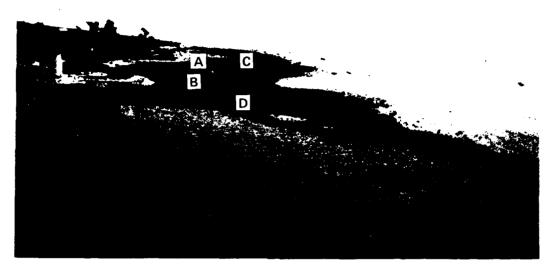


Photo 38. Site NY1

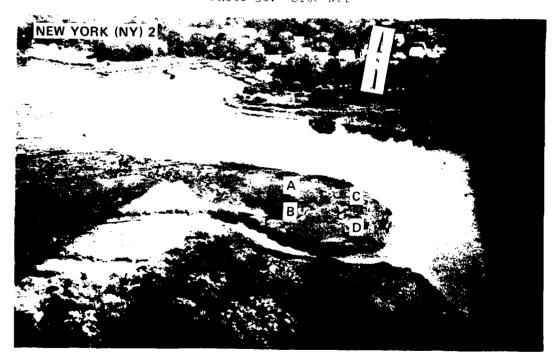


Photo 39. Site NY2

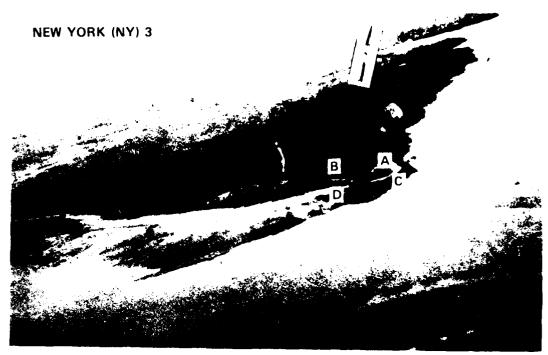


Photo 40. Site NY3

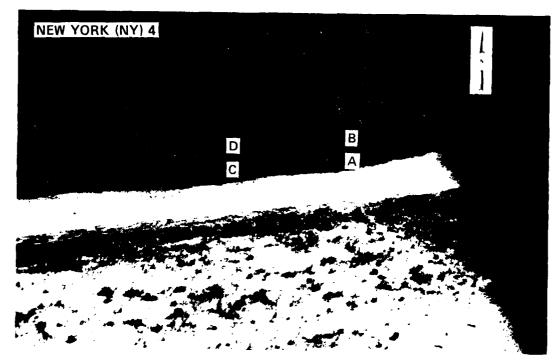


Photo 41. Site NY4

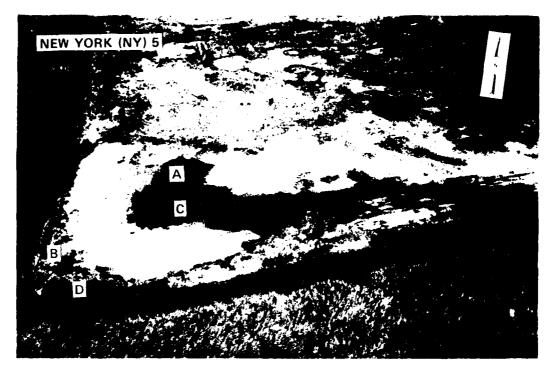


Photo 42. Site NYS

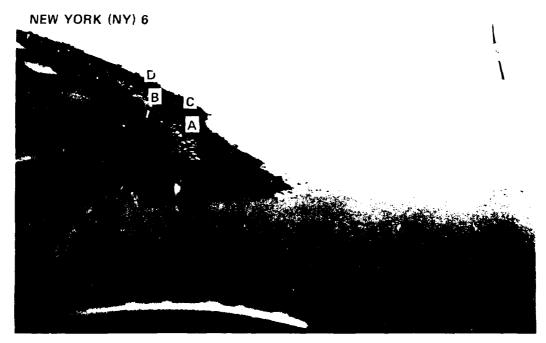


Photo 43. Site NY6

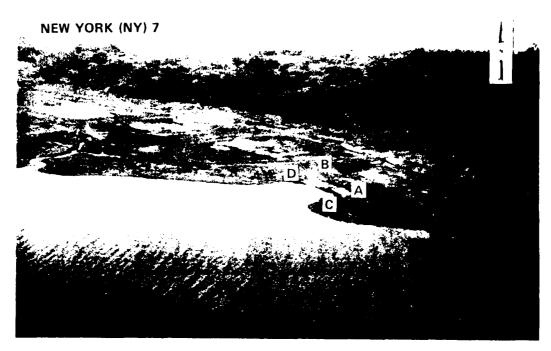
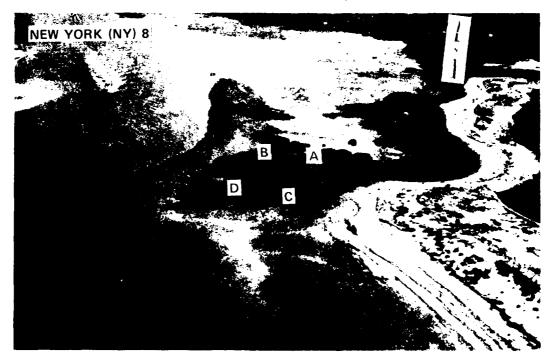


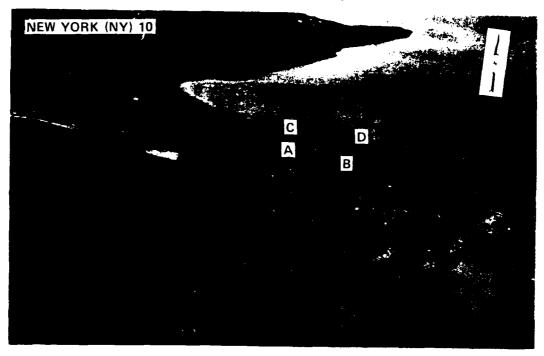
Photo 44. Site NY7



Jamo W. Olfery



Photo 46. Site NY9



Costo al. Sito NYLY

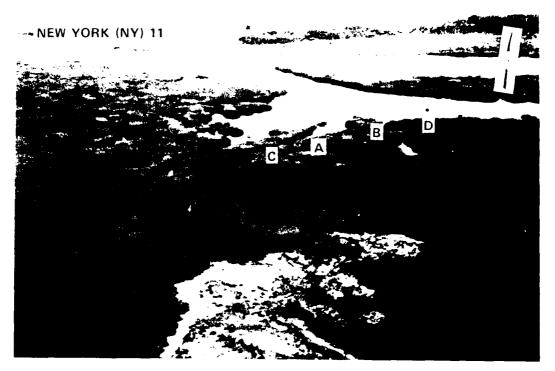


Photo 48. Site NY11

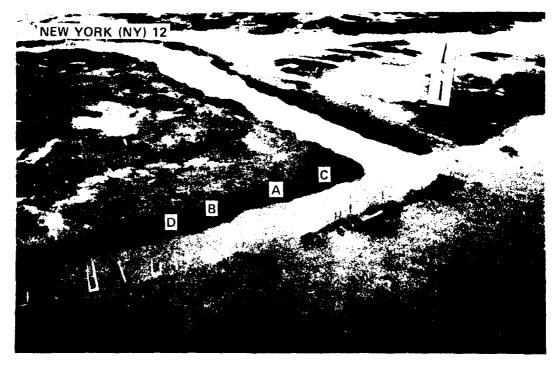


Photo 49. Site NY12

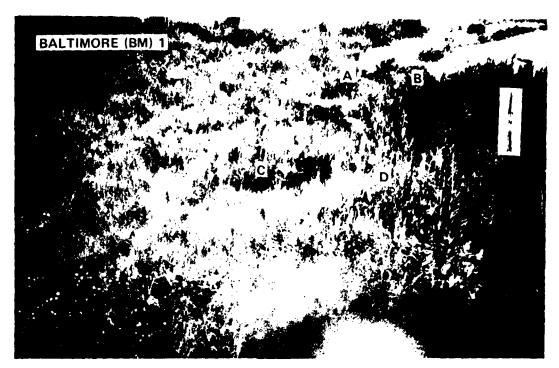


Photo 50. Site BMl



Photo 51. Site BM2

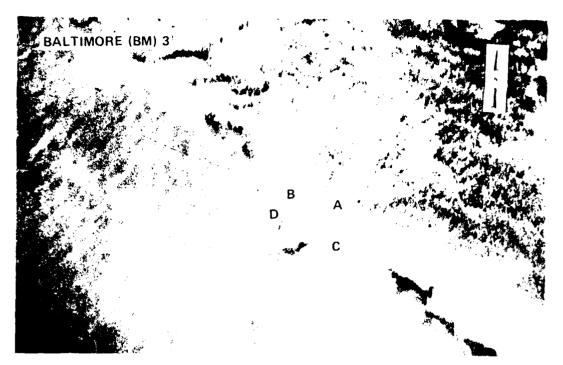


Photo 52. Site BM3

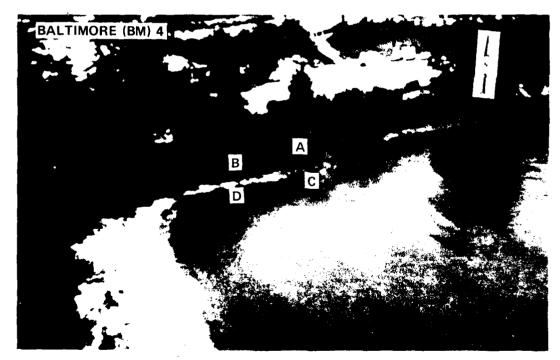


Photo 53. Site BM4

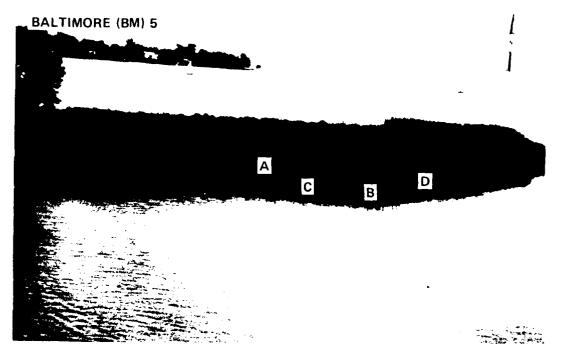


Photo 54. Site BM5

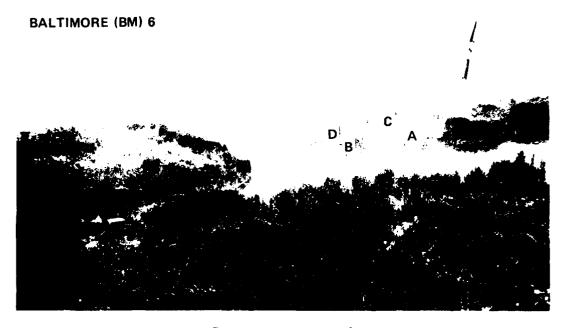


Photo 55. Site BM6

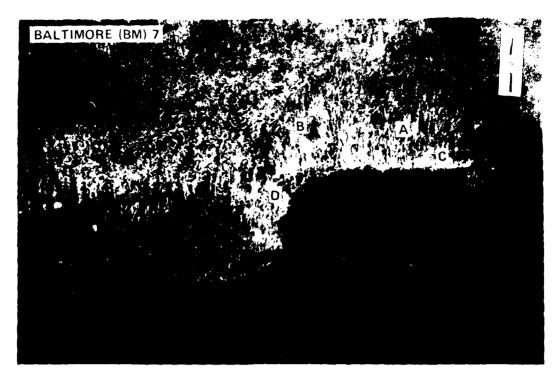


Photo 56. Photo BM7

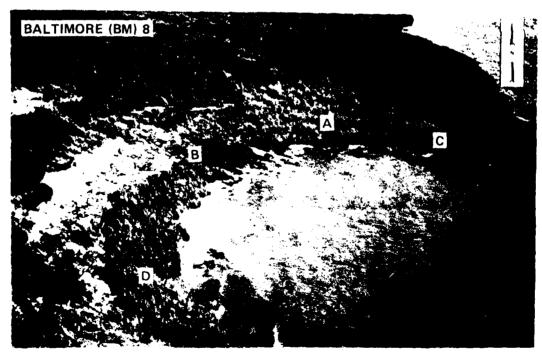


Photo 57. Site BMS

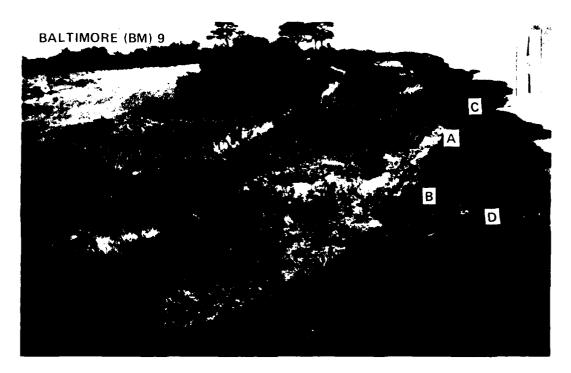
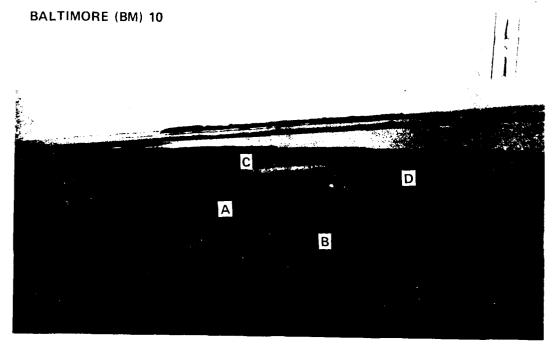


Photo 58. Site BM9



Thoto 50. Site BM10

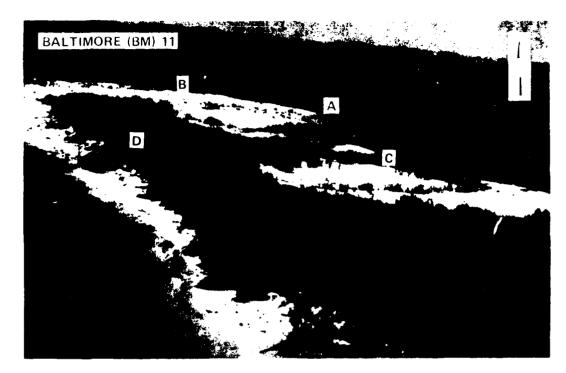


Photo . '. lite !"...

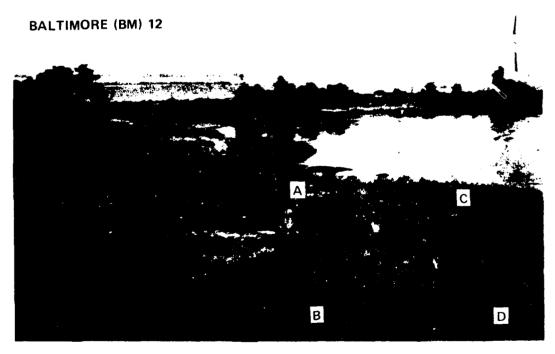


Photo GL. The PT

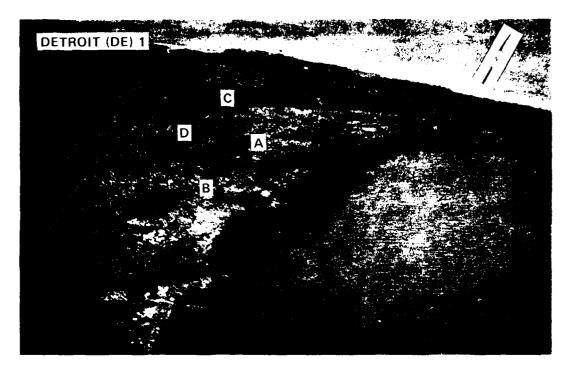


Photo 62. Site DE1

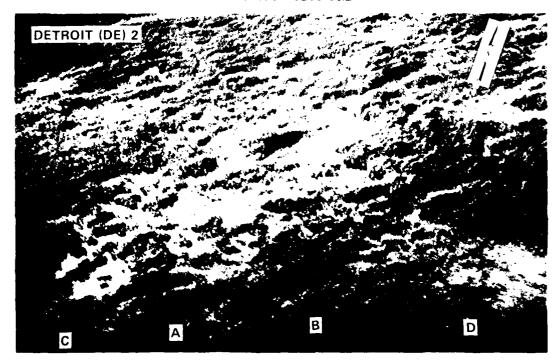


Photo 63. Site DE2

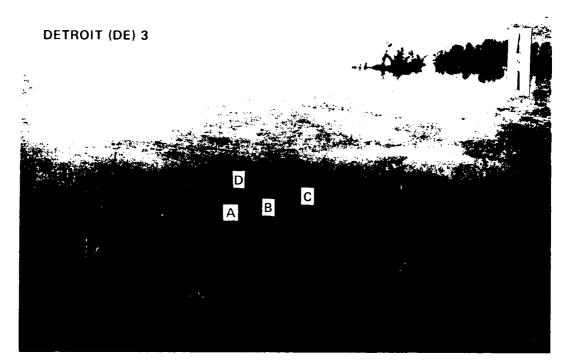
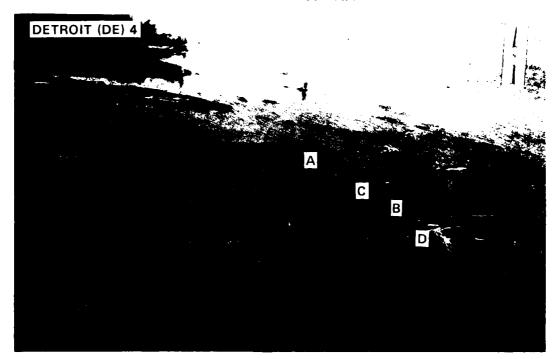


Photo M. Dita 183



Hoto 65. Site DEA



Photo 66. Site DE5

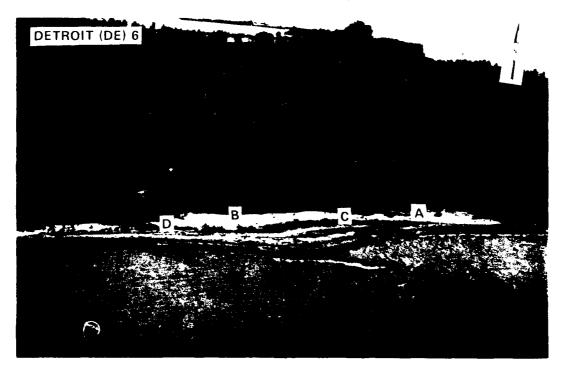


Photo 67. Site DE6

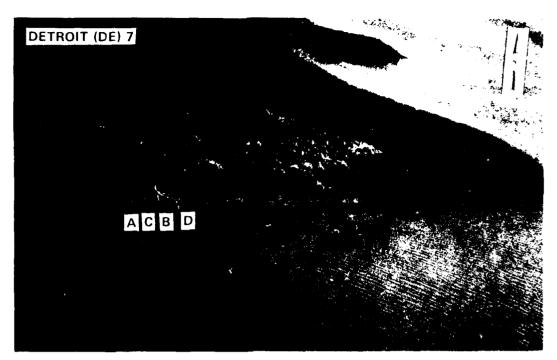


Photo 68. Site DE7

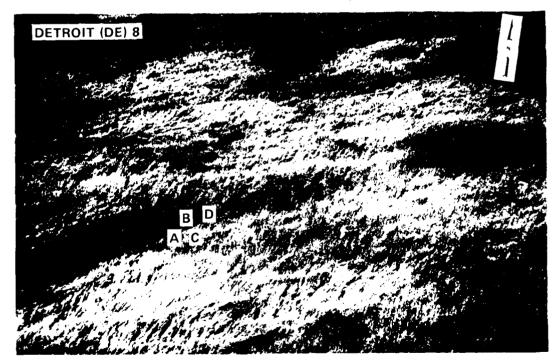


Photo 69. Site DE8

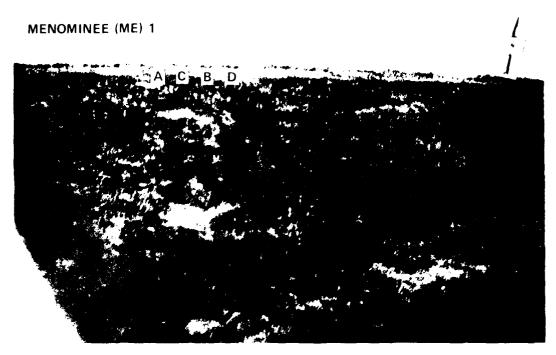
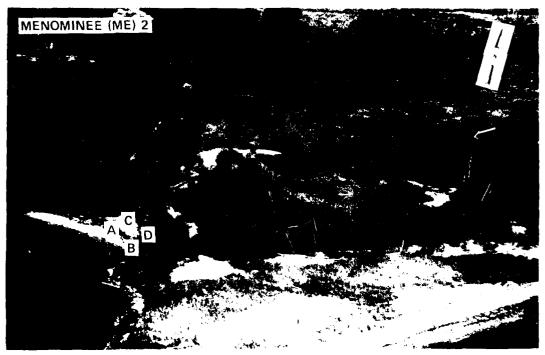


Photo 70. Site Min



shate 71. fite MES



Photo 72. Site ME3

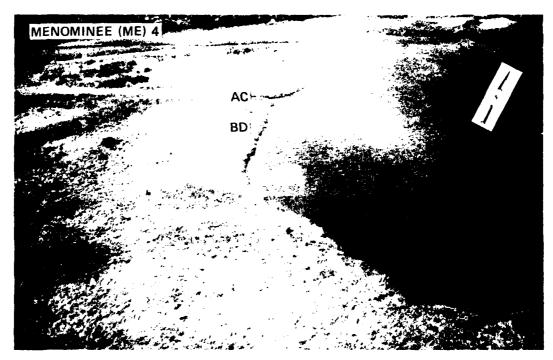


Photo 73. Site ME4

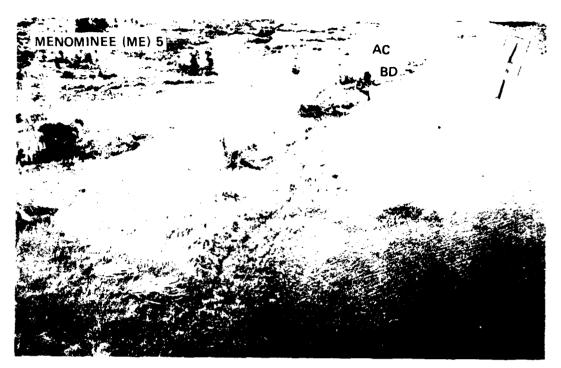


Photo 74. Site ME5

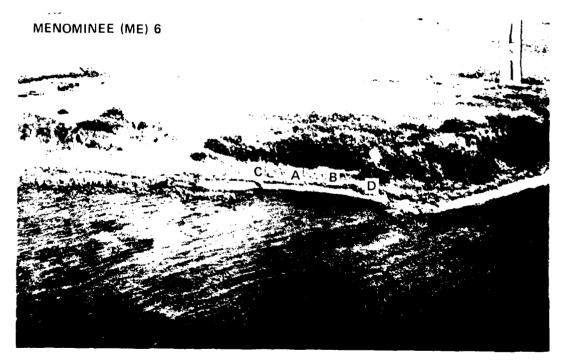


Photo T. Cita Mr



Photo 76. Site MW1

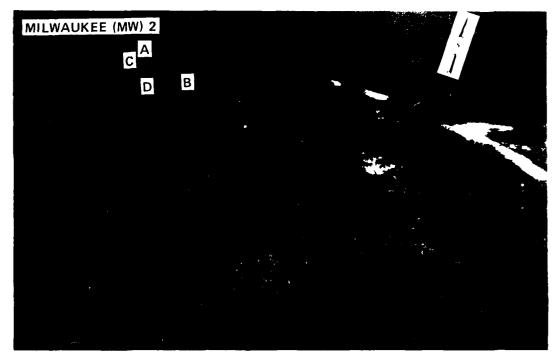


Photo 77. Cite MWS

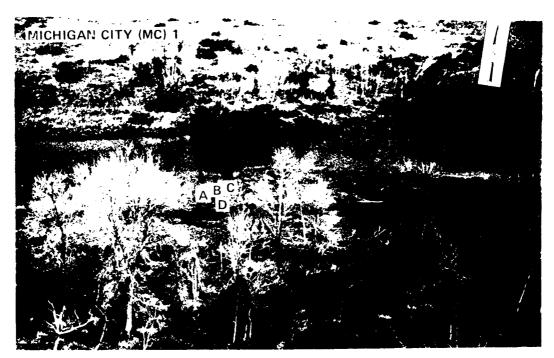
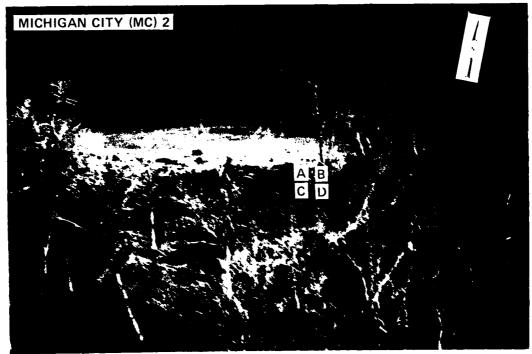


Photo /o. Site MCl



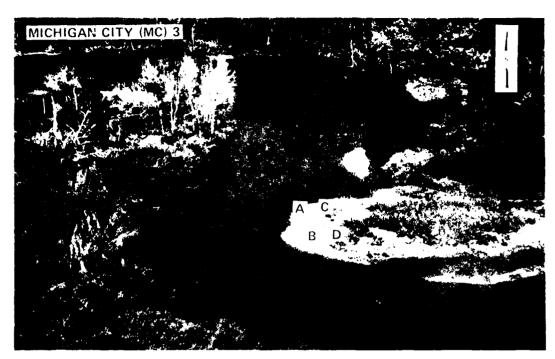


Photo 80. Site MC3

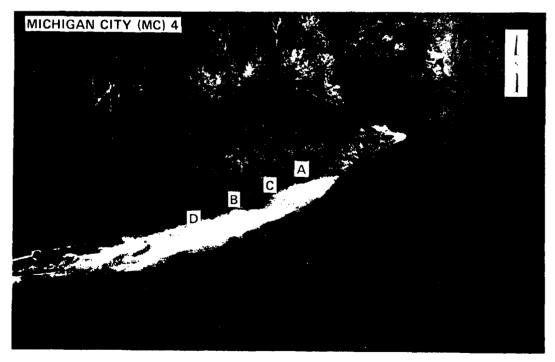


Photo Mi. Site Mch

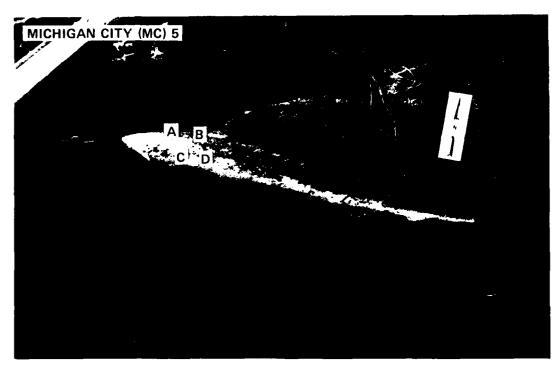


Photo 82. Site MC5

PHOTO NOT AVAILABLE ON MICHIGAN CITY SITE 6

Photo 83. Site MC6

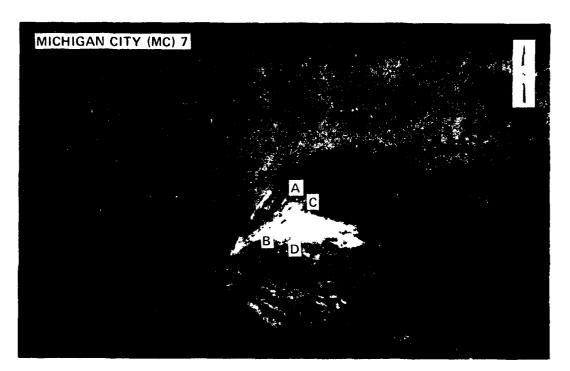


Photo 84. Site MC7

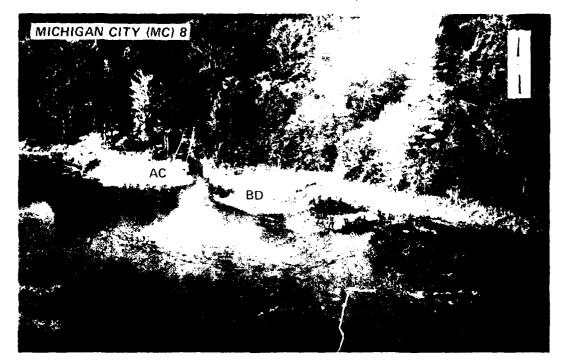


Photo 35. Site Mos

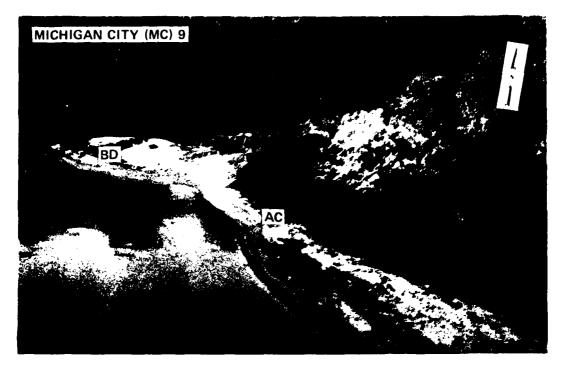


Photo 86. Site MC9

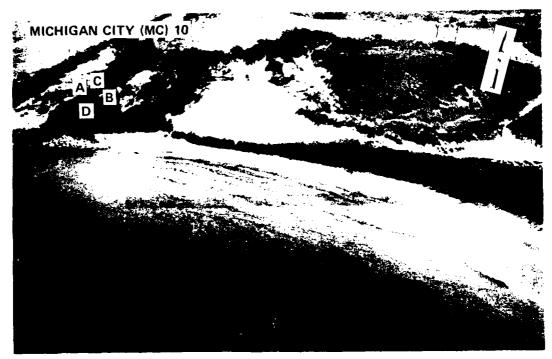


Photo 87. Site MC10

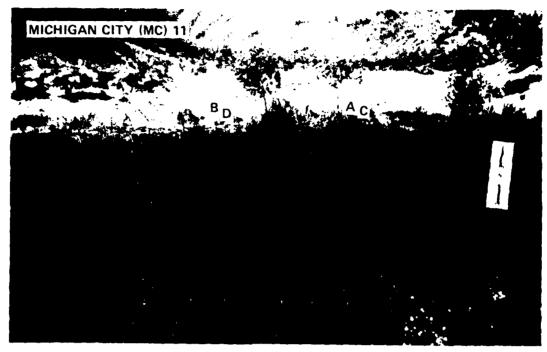


Photo 88. Site MC11

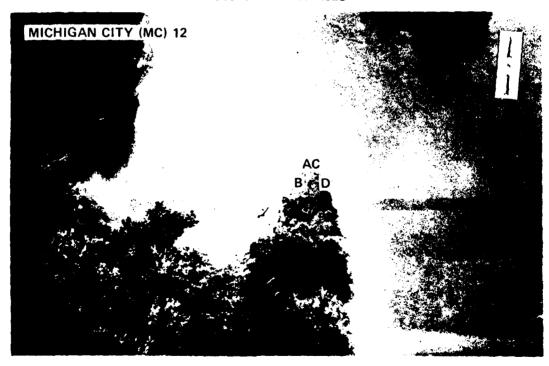


Photo 89. Site MC12

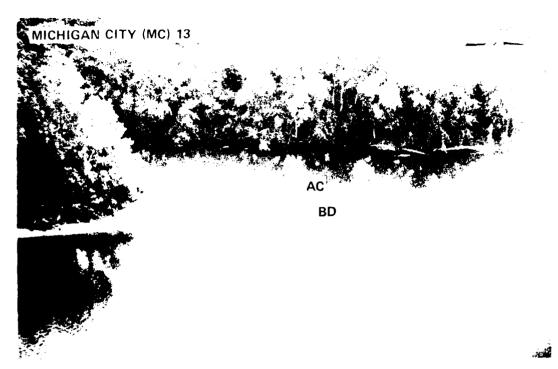


Photo 90. Site MC13



Photo 91. Site IN1

APPENDIX B: SALTWATER COLLECTION SITE PHYSICAL DATA

Saltwater Collection Site Physical Data

Jeneral Area	Collection Site No.	Collection Date	USGS Topographic Sheet 7.5'	North Latitude	West Longitude	Frevailing Wind Direction	Tidal Range, ft	Tidal Range, m
Corpus Christi, Tex.	000 000 000 000 000 000 000 0010	18 Jul 78	Corpus Christi, Tex. Corpus Christi, Tex. Taft, Tex. Corpus Christi, Tex. Corpus Christi, Tex. Portland, Tex. St. Charles Bay, Tex. Mesquite Bay, Tex. Mesquite Bay, Tex. Port Lavala, East, Tex. Palacio, NE, Tex.	27%11 27%11 27%22 27%22 27%22 28%10 28%10 28%30.51	97°30' 97°31' 97°30.5' 97°27' 97°27' 96°57' 96°53' 96°10.5'	or <del></del>	α <u>.</u> 	0.67
	0012 0013 0014 0015		Cedar Lakes, West, Tex. Christmas Point, Tex. Virginia Point, Tex. Frozen Point, Tex.	28°50.8° 29°01.5° 29°18° 29°34.5°	95°32' 94°14.5' 94°56' 94°34'			
Hew Orleans, La.	NO1 NO2 NO3 NO5 NO6	20 Jul 78	Bay St. Louis, Miss. Grand Island Pass, MissLa. Alligator Point, La. Lake Eloi, La. Main Pass, La. Burns, La.	30°21.5' 30°36' 30°31.2' 29°45.5' 29°26' 29°21'	89°27.5°89°27.5°89°23.7°89°33°89°33°89°35°8	1111111	0.	0.61
	NO8 NO9 NO10 NO11	4 21 Jul 78 21 Jul 78 21 Jul 78 21 Jul 78	Caminado Pass, La. Bay Dosgris, La. Trand Bayou Du Large, La. Point Au Per, La. Bayou Lucien, La.	29°25.5' 29°25.5' 29°28' 29°30.1'	90°06' 90°04' 90°57' 91°20' 91°53.2'		(v	· · · · · · · · · · · · · · · · · · ·
Jacksonville, Fla.	7V1 1V2 1V3 1V3 1V5 1V7 1V10 1V10	01 Aug. 78	Amelia City, Fla. Fernandina Beach, Fla. Jekyll Island, Ga. Isle of Hope, Ga. Pockville, S. C. Ames Island, S. C. Avenday, S. C. McJellanville, S. C. Santee Point, S. C. Challotte, N. C. Kure Beach, N. C.	30°31.0° 30°43.0° 31°01.0° 31°52.5° 32°01.5° 33°01.0° 33°01.0° 33°56.0°	81°25.6' 81°26.0' 81°26.0' 81°01.5' 80°50.5' 79°58.0' 79°21.0' 79°21.0' 79°21.0' 79°21.0'	దే <del>——</del> ≃——— దే దే ్	000	2.74

	Collection Collection	Collection		North	West	Frevailing		
Jeneral Area	Site No.	Date	USGS Topographic Sheet 7.5'	Latitude	Longitude	Wind Direction	Tidal Range, ft	Tidal Fange, m
New York, N. Y.	NYI	08 Aug 78	Bridgeport, Conn.	41008.5	73º12.7'	Ж.	2.5	2.19
	NY2	_	_	41,009.3	73°12.8'			
	NY3			41,010.01	73009.01		_	
	7XK			41,009.01	73°08.0			
	NYS			41°09.2'	73°07.8'	_	-	-
	NY6		Milford, Conn.	41,009.14	73,06.01	-	0.0	0.91
	NY7	-	Mystic, Conn.	41°20.0'	71°52.5'	-	(\frac{1}{2})	2.67
	NY8	09 Aug 78	Lloyd Harbor, N. YConn.	40057.7	73024.0'	- :	4.7	1.43
	6ÅN		Sandy Hook, N. J.	40°25.5	74,000.5	6.1	_	
	NY10		Ship Bottom, N. J.	39044.01	74.05.01	_		
	NY11		Brigantine Inlet	39°28.5	74,019.0	-		
	NY12		Sea Isle City, N. J.	39°12.0	74039.01	_		
		-						-
Baltimore, Md.	BMI	10 Aug 78	Gibson Island, Md.	39001.0	16024.5.	NE	1.3	0.40
	BM2		Sparrows Point, Md.	39°14.01	76°24.0"			
	BM3		Sparrows Point, Md.	39°13.0°	76°25.01			
	Bivit		Middle River, Md.	39°17.0	76°23.0'			_
	BM5		Middle River, Md.	39°19.0'	16024.01		-	
	BM6	_	Middle River, Md.	39°16.0	76°26.01			
	3M7	11 Aug 78	Deak, Md.	38°49.0	76°30.1'	-	-	-
	BM8	_	Reedville, Va.	37047.01	76,17.0	PSS.	3.1	0,93
	BW9		New Point Comfort, Va.	37,20.01	76°16.0'	AS.		0.93
	BMI 0	_	Townsend, Va.	37°08.0	75°56.2'	AS.		0.93
	BWII	_	Kedgas Straits, Md.	38°01.0'	76°03.01	:		0.95
	BM12	-	Longgood Creek, Md.	39°01.0	76°12.5'	NE	1.3	0.40

APPENDIX C: FRESHWATER COLLECTION SITE PHYSICAL DATA

Freshwater Collection Site Physical Data

General Area	Collection Site No.	Collection Date	USGS Topographic Sheet	North Latitude	West	Prevailing Wind Direction
Detroit, Mich.	DE1 DE2 DE3 DE4 DE5 DE6 DE7 DE9	16 Aug 78 16 Aug 78 17 Aug 78	Rockwood, Mich. 7.5' Mt Clemens East, Mich. 7.5' Wyandotte, Mich. 7.5' Rockwood, Mich. 7.5' Estral Beach, MichOhio 7.5' Stony Point, Mich. 7.5' Erie, MichOhio 7.5' Mt Clemens East, Mich. 7.5' Mt Clemens East, Mich. 7.5'	42°01.0' 42°32.0' 42°38.0' 42°57.5' 42°57.5' 41°52.5' 42°32.2'	83°12.0' 82°38.0' 83°07.5' 83°10.0' 83°21.0' 82°38.0'	AS .
Menominee, Mich.	ME1 ME2 ME3 ME4 ME5 ME5	22 Aug 78	Little Sturgeon, Wis. 15' Little Sturgeon, Wis. 15' Conto, Wis. 15' Conto, Wis. 15' Conto, Wis. 15' Marinette, WisMich. 7.5'	44°58.5° 44°58.2° 44°55.2° 44°55.2° 45°04.5°	87°37.5' 87°40.0' 87°48.0' 87°50.5' 87°52.0' 87°35.0'	
Milwaukee, Wis.	MW2 MW2	23 Aug 78 23 Aug 78	Zion, Ill. 7.5' Zion, Ill. 7.5'	42°25.0' 42°25.0'	87°45.0° 87°45.0°	33
Michigan City, Ind.	MC1 MC2 MC4 MC5 MC5 MC6 MC7 MC8 MC9 MC10 MC10	24 Aug 78	Fennville, Mich. 15'  Grand Haven, Mich. 7.5'  Port Sheldon, Mich. 7.5'  Benton Harbor, Mich. 7.5'  Michigan City West, Mich. 7.5'  Three Oaks, Mich. 7.5'  Sodus, Mich. 7.5'  Sodus, Mich. 7.5'	42°38.0° 42°37.7° 42°37.7° 42°54.0° 42°51.0° 41°42.6° 41°41.5° 42°58.5°	86°09.0' 86°08.5' 86°05.0' 86°04.5' 86°04.0' 86°09.3' 86°11.5' 86°23.0' 86°21.0'	ASS ASS
Indiana Harbor, Ind.	INI	-	Portage, Ind. 7.5'	41°35.2'	87°07.6'	1

APPENDIX D: CYPERUS SPECIES COLLECTED BY SITE

Cyperus Species Collected by Site

Collection Site	Location	Sample	Species	WES Herbarium Number
DEL	Wayne County, Mich.	A	C. odoratus C. erythrorhuzos	0001FS75
		Д	C. odoratus	
		ပေး	_	
		그 떠	-	
			C. strigosus	
DES	Macomb County, Mich.	А		0002FS78
		മ	C. Englemanni*	•
		బ	C. odoratus	
			C. Englemanni	•
		Д	C. Englemanni	•
		ឯ	C. Englemanni*	*
DE3	Wayne County, Mich.	А	C. strigosus	0003FS78
		മ	,	
		ن		
		D		
		ſщ		
ጋይታ	Wayne County, Mich.	A	C. odoratus	0004FS78
		Д		
		ı		
		บ	C. odoratus	
		О	C. odoratus	
			c. strigosus	
		(Continued)		

\* New county record for this species.

				WES
Collection Site	Location	Sample	Species	Herbarium Number
DE4 (Continued)		E	C. odoratus C. strigosus	
DES	Monroe County, Mich.	4 W O O H	C. odoratus	0005FS78
DE6	Monroe County, Mich.	<b>4</b> 원 0 요 표	C. odoratus C. strigosus C. strigosus C. strigosus C. strigosus C. odoratus C. strigosus	0006FS78
DET	Monroe County, Mich.	<b>4</b> 표 오건편	C. odoratus C. odoratus C. strigosus C. odoratus C. odoratus C. odoratus C. odoratus	0007FS78
DE8	Macomb County, Mich.	чшо сы	C. Englemanni* C. Englemanni* C. Englemanni* C. erythrorhyzos C. Englemanni* C. Englemanni*	0008FS78
DE9	Macomb County, Mich.	A (Continued)	C. Englemann:* C. erythrorhyzos	.0s

\* New county record for this species.

\* New county record for this species.

(Sheet 3 of 8)

\* New county record for this species.

(Sheet 4 of 8)

					WES
Collection Site	Location	Sample	Species	S	Herbarium Number
MC2	Allegan County, Mich.	A	C. erythrorhyzos	rhyzos	0019FS78
		Д	C. erythrorhyzos	rhyzos	
				nni	
		ပ	C. erythrorhyzos	rhyzos	
			-	nni	
		Ω		rhyzos	
		드리		sozhus	
			C. Englemanni	mni	
MC3	Allegan County, Mich.	A	C. erythrorhyzos	rhyzos	0020FS78
			C. odoratus	S	
		ф	C. erythrorhyzos	rhyzos	
			C. odorutus	. 81	
			C. strigosus	เนธ	
		ల	C. erythrorhyzos	sozhyso	
				81	
		Д	C. erythrorhyzos	rhyzos	
			C. odoratus	S	
			C. strigosus	เนธ	
		团	C. erythrorhyzos	rhyzos	
			C. odoratus	81	
			C. strigosus	ıns	
МС4	Allegan County, Mich.	A	C. erythrorhyzos	rhyzos	0021FS78
	1			*81	
		മ	C. erythrorhyzos	rhyzos	
				*81	
		೮	C. erythrorhyzos	rhyzos	
			C. odoratus*	*87	
		<b>A</b>	C. erythrorhyzos	rhyzos	
			C. odoratus*	*8	
		(Continued)			

The second secon

<sup>\*</sup> New county record for this species.

				WES
Collection Site	Location	Sample	Species	Herbarium Number
MC4 (Continued)		ы	C. erythrorhyzos C. odoratus*	
MC5	Allegan County, Mich.	A 0	C. erythrorhyzos C. odoratus*	0022FS78
		В		
		υ		
		Д		
		EI		
		00	C. odoratus* C. strigosus	
MC6	Allegan County, Mich.	A		0023FS78
			C. Englemanni*	
		D C.		
			C. erythrorhyzos	
		3	. odoratus*	
MC7	Ottawa County, Mich.	A C	C. odoratus	0024FS78
		0	C. strigosus	
		Э		
		0		
		o o		
			c. Englemannı	
		a 0	c. oaoratus C. Englemanni	
	0)	(Continued)		

<sup>\*</sup> New county record for this species.

				WES
Collection Site	Location	Sample	Species	Herbarium Number
MC7 (Continued)		ម	C. odoratus C. Englemanni	
мс8	Ottawa County, Mich.	∢ a o i		0025FS78
		<b>э</b> ы	c. Englemanni C. Englemanni	
мс6	Berrien County, Mich.	A		CU26FS78
		дυд	C. odoratus C. odoratus C. odoratus	
		덦	C. strigosus C. odoratus C. strigosus	
MCIO	La Porte County, Ind.	<b>ч</b> ш С С Н	C. odoratus	0027FS78
MG11	Berrien County, Mich.	чы осы	C. strigosus C. strigosus C. erythrorhyzos C. strigosus C. odoratus C. strigosus	0028FS78
MC12	Berrien County, Mich.	Ą	C. strigosus C. odoratus	0029FS78
	")	(Continued)		(Sheet 7 of 8)

					WES
Collection Site	Location	Sample	1	Species	Herbarium Number
MC12 (Continued)		М	<i>:</i> '	C. strigosus	
			<i>'</i>	odoratus	
		ಲ	ਂ	strigosus	
		А	Ċ.	strigosus	
			<i>ن</i>	odoratus	
		ы	<i>'</i>	strigosus	
			<i>'</i> :	odoratus	
MC13	Berrien County, Mich.	А	ပ	C. odoratus	0030FS78
		В			
		బ			
		А		-	
		E		-	
INI	Porter County, Ind.	A	ં	C. strigosus	0031FS78
		Д		_	
		೮ 1		_	
		<b>그</b> F			
		ů.			

APPENDIX E: LEAF TISSUE HEAVY METAL CONTENT (µg g<sup>-1</sup>)
OF SPARTINA ALTERNIFLORA SAMPLES

(Sheet 1 of 8)

	Hg	0.0138	0.0088	0.0163	0.0113	0.0113	0010.0	0.0100	0.0175	0.0125	0.0100	0.0050	0.0150	0.0125	0.0100	0.0338	0.0313	0.0125	0.0125	0.0238	0.0212	0.0138	0.0188	0.0175	0.0100	0.0075	0.0075	0.0063	0.0175	<0.0050
les	Zn	22.057	14.802	13.651	12.745	16.704	10.973	10.412	16.087	6.663	30.312	27.550	29.845	20.050	40.050	22.733	22.983	900.9	14.300	19.016	17.412	17.058	19.847	13.313	27.355	7.041	16.408	13.947	5.863	14.584
ora Samp	Ni	1.102	0.901	1.226	0.402	0.377	4.555	3.875	<0.075	4.475	<0.075	1.700	<0.075	<0.075	1.675	0.676	0.777	2.252	2.250	1.778	1.275	4.583	3.952	1.176	2.179	<0.075	1.553	0.226	3.325	2.917
of Spartina Alterniflora Samples	Mn	267.886	163.013	267.618	105.826	79.418	163.326	51.912	102.317	153.412	218.912	277.367	322.851	269.867	66.617	295.441	75.000	192.309	179.867	310.315	156.850	12.547	11.131	74.191	77.522	252.289	250.200	252.982	71.662	36.893
Spartina	Pb	1.979	1.101	1.552	0.753	1.406	1.114	2.062	2.116	1.212	1.962	2.158	2.287	1.783	2.558	2.179	2.004	2.736	7.058	3.430	4.550	1.320	0.975	3.111	1.887	4.290	2.204	1.053	2.087	2.191
(µg g <sup>-1</sup> ) of	Fe	105.060	94.194	222.573	65.394	67.871	84.347	85.512	113.933	68.762	155.512	100.617	165.615	108.367	719.17	145.140	88.277	47.915	48.617	61.693	63.350	237.415	175.938	160.027	154.426	65.545	186.072	66.015	103.762	46.157
ssue Heavy Metal Content (µg g <sup>-1</sup> )	r <sub>O</sub>	0.927	<0.025	1.602	3.089	3.188	3.854	3.350	4.206	2.025	4.875	3.950	4.882	4.925	3.200	1.428	1.403	5.931	5.250	3.731	3.375	3.228	3.052	2.578	4.208	2.226	4.108	1.905	3.100	4.432
Heavy Met	Cr	2.856	3.028	3.103	2.913	3.589	2.603	2.925	3.956	2.325	2.725	1.600	<0.025	<0.025	0.650	<0.025	<0.025	<0.025	<0.025	1.527	1.100	3.117	2.851	1.564	1.165	0.850	4.108	2.632	1.500	2.178
Leaf Tissue	Cd	0.1177	0.0275	0.1126	0.0201	0.0477	<0.0025	<0.0025	0.0075	0.0175	<0.0025	0.1783	0.0058	0.0433	0.0433	0.1854	<0.0025	0.2586	0.0958	0.1678	0.0100	0.0225	0.0050	0.0209	0.0209	0.0175	0.0100	0.0327	0.0050	0.0801
Ţ.	As	0.125	<0.025	0.150	<0.025	000.0	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.050	<0.025	<0.025	<0.025	<0.025	<0.050	<0.025	<0.025	<0.025
	Site	BM1 A	BM1 B		BM1 D	BM2 A	BM2 P												BM5 B										BM7 D	

Site	As	Cd	r)	Cu	Fe	Pb	Mn	Ni	Zn	Hg
	<0.025	0.0175	2.701	2.451	46.625	1.526	27.364	0.775	24.450	0.0163
BM8 C	<0.025	0.0450	3.752	2.651	86.393	1.851	25.863	1.276	35.955	0.0163
BM8 D	<0.025	<0.0025	4.250	2.650	58.512	0.063	21.537	<0.075	10.313	0.0075
	<0.025	0.0408	2.200	4.525	36.867	1.008	44.117	1.350	0.225	0.0100
BM9 B	<0.025	0.0075	1.027	3.255	41.149	1.740	34.139	2.717	9.326	0.0063
	<0.025	0.1134	1.638	2.826	38.636	2.610	11.397	0.450	27.064	0.0200
	<0.025	0.0000	4.100	1.875	42.512	2.287	11.687	<0.07	4.713	0.0075
0	<0.025	0.1358	1.500	3.850	70.617	2.833	31.367	1.750	1.050	0.0100
0	<0.025	0.0125	3.756	4.106	82.974	0.926	27.391	1.427	11.905	0.0038
0	<0.025	0.0800	2.926	3.477	61.293	2.864	43.684	<0.075	6.591	0.0100
BMIO D	<0.025	0.0533	2.013	4.775	103.867	1.783	48.617	0.975	10.225	0.0100
BM11 A	<0.025	0.1342	2.426	4.364	79.378	1.928	63.131	3.050	16.797	0.0269
Н	<0.025	0.1009	1.926	7,402	44.139	1.734	101.417	2.076	27.564	0.0275
$\overline{}$	<0.025	0.1883	1.800	3.775	136.867	1.283	40.117	1.875	33.050	0.0125
М	<0.025	0.0659	1.451	3.552	109.421	1.784	33.884	2.601	19.060	0.0125
S	<0.025	0.0275	1.075	4.875	103.262	1.262	162.412	4.075	7.363	0.0200
Ω.	<0.025	0.0625	009.0	2.751	71.873	3.139	109.392	3.089	12.769	0.0038
	<0.025	0.0551	0.501	4.507	92.727	2.491	230.684	5.070	29.357	0.0188
CV.	<0.025	0.0266	3.287	6.422	87.155	4.114	69.594	0.552	13.108	0.0138
	<0.025	0.0175	3.702	4.002	97.561	1.138	56.191	5.078	41.333	0.0150
	<0.025	0.3977	3.427	1.251	108.679	3.877	54.177	2.251	27.889	0.0275
CCIC	<0.025	0.2601	3.152	4.377	114.320	1.688	45.435	4.277	30.078	0.0100
	<0.225	0.2300	4.150	1.475	75.012	2.912	28.662	<0.075	26.312	0.0025
	<0.025	0.1300	3.613	3.013	98.786	1.038	36.672	2.913	17.378	0.0113
	<0.025	0.3058	2.125	3.625	120.617	1.383	76.617	1.850	45.050	0.0050
	<0.025	0.1251	3.427	0.925	94.899	2.051	38.869	1.701	41.458	0.0138
	<0.025	0.1651	2.701	3.427	101.813	1.338	159.742	1.551	25.325	0.0200
	<0.025	0.3833	9.700	6.325	153.867	8.083	120.867	3.875	686.050	0.0100
	<0.025	0.1251	2.851	1.126	962.99	2.814	44.685	1.726	26.826	0.0075
	<0.025	0.0200	3.702	1.676	113.819	5.415	26.176	2.376	27.326	0.0075
	<0.025	0.0475	3.700	<0.025	67.600	2.500	15.275	1.625	18.287	0.0013
					(Continued)					

Hg	0 6.0575																														
Zn	40.550	14.129	28.59	11.36	14.07	98.236	18.81	22.56	47.37	15.05	2.35	18.55	3.68	7.58	4.91	12.95	<0.01	12.43	20.800	15.05	18.28	22.30	24.80	4.75	28.579	7777	6.11	5.86	4.36	16.92	7.2%
Ni	1.125																														
Mn	31.617	31.900	20.908	31,412	62.225	26.627	33.662	31.883	155.369	42.138	23.704	172.203	99.712	45.100	45.435	21.559	32.650	24.200	25.117	40.137	25.375	42.367	14.242	70.117	272.639	47.186	122.535	49.437	189.912	72.923	142.805
Pb	3.383																														
Fe	108.617	209.625	154.417	56.762	66.579	1779.378	103.262	55.144	52.747	44.139	36.385	32.633	33.029	63.350	46.036	83.346	49.375	46.850	38.867	43.889	80.180	49.367	210.367	40.367	117.985	31.278	41.554	43.284	762	76.426	.77.080
Cu	2.475	1.025	1.752	3.925	2.878	3.904	3.075	2.776	2.613	2.801	2.951	4.302	1.951	<0.025	2.251	1.276	0.500	0.850	1.550	2.726	2.127	2.050	3.225	3.575	4.755	1.451	0.576	1.976	3.275	1.602	4.329
Cr	2.863	2.900	3.804	2.275	3.203	6.331	4.550	3.214	1.750	2.201	1.001	1.563	2.501	3.500	3.877	1.927	2.425	3.650	1.600	0.725	2.903	0.825	2.025	2.750	2.340	3.007	3.078	1.851	2.225	0.375	2.653
Cd	0.0958	0.1325	0.1226	0.0025	0.0200	9.1476	0.0250	0.0659	0.0904	0.1109	0.0058	0.1009	0.0975	0.0225	0.0100	<0.0025	0.0425	0.0375	0.0558	0.1159	0.0275	0.0808	0.0408	0.1433	0.0934	<0.0025	0.0100	<0.0025	<0.0025	0.0776	0.0150
As	<0.025	<0.025	<0.025	<0.025	0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	0.125	0.125	<0.025	<0.025	<0.025	<0.025	0.175	0.025	0.025	0.175	<0.025
Site	GC4 B	2 722	CC4 D	CC5 A	CC5 B	CC5 C	CC5 D	cc6 A	CCÓ B	၁ 9၁၁	c 900	CCT A	cc7 B	cc <sub>7</sub> c	CC1 D	CC8 A	CC8 B	CC8 C	CCB D	CC9 A	сс В	၁ 6၁၁	Q 600	CCIO A					CCII B		

As0	Cd 0.2932		Cr 2.481	3.659	Fe 76.541	Pb 1.479	Mn 343.208	Ni 1.228	Zn 31.266	Hg 0.0163
0	1380	1.779		5.313	69.010	2.018	325.652	4.574	18.609	0.0013
.025 0.1385 .025 0.0659	.1385 .0659	1.714		3.804	105.973	1.109	182.299	4.680 4.802	19.820	0.0275
0.025 0.1560	.1560	1.802		3.829	99,466	2.286	144.761	4.555	28,328	<0.0050
0.025 0.0275	.0275	3.052		2.051	87.056	1.463	211.768	1.751	4.090	0.0025
0.025 0.1376	.1376	3.202		2.151	54.877	1.226	87.894	1.751	19.097	0.0013
.025 0.0458	.0458	0.613		3.225	97.367	2.208	188.367	1.625	19.050	0.0075
0.025 0.0450	.0450	4.000		2.450	101.350	1.075	60.850	0.650	15.313	0.0013
.025 <0.0025	.0025	4.452		2.551	138.582	0.738	82.704	<0.075	7.616	0.0050
0.025 0.0425	.0425	3.700		1.600	60.850	1.200	46.600	1.800	16.512	0.0038
0.025 0.0175	.0175	2.301		1.426	126.163	1.576	22.286	3.552	16.496	0.0038
.025 0.1233	.1233	2.100		3.850	43.617	1.508	123.117	1.875	8.975	0.0075
0.025 0.0359	.0359	1.763		9.030	81.407	1.309	167.200	0.075	13.057	0.0025
.025 0.0577	.0577	3.813		3.889	61.064	1.530	116.006	0.477	15.341	<0.0050
0.025 0.0225	.0225	3.627		2.701	61.631	0.425	325.013	1.601	17.521	0.0038
25 0.1378 1.128	.1378 1.128	128	•	:0.025	43.960	1.554	18.797	0.376	15.752	0.0288
25 0.0608 3.188	.0608 3.188	188		2.575	61.117	1.808	22.717	0.075	8.000	0.0175
75 0.1809 2.889	.1809 2.889	389		0.350	122.678	1.059	42.638	0.275	8.429	<0.0050
25 0.1459 2.526	.1459 2.526	526		0.825	105.169	1.809	73.153	1.926	25.063	0.0050
25 0.0533 0.500	.0533 0.500	200		2.500	314.117	1.283	35.367	0.500	63.550	0.0125
25 0.0075 2.126	.0075 2.126	126		1.376	56.378	1.426	65.883	1.326	34.955	0.0188
50 <0.0025 3.305	.0025 3.305	305		0.200	61.355	1.039	29.957	<0.075	6.122	0.0100
25 0.0100 2.076	.0100 2.076	920		0.425	93.559	0.713	55.690	4.627	7.166	0.0100
50 0.0527 2.633	.0527 2.633	633		1.304	106.419	1.605	40.471	0.150	14.030	0.0088
25 0.0250 1.300	.0250 1.300	300		0.325	60.762	0.362	34.662	3.800	3.938	0.0050
25 0.0625 2.075	.0625 2.075	075		1.225	258.262	1.287	38.912	<0.075	0.388	0.0250
25 <0.0025 2.301	.0025 2.301	301	٧	:0.025	69.797	1.288	43.684	<0.075	3.614	0.0075
25 0.0100 2.501	.0100 2.501	501		0.575	49.037	0.163	71.948	3.302	7.741	0.0075
0025 2.300	.0025 2.300	300		0.150	41.012	0.512	74.162	1.425	16.062	0.0125
50 0.0450 2.177	.0450 2.177			0.125	118.631	0.413	39.455	3.478	11.649	0.0075
					(Continued)					

Hg	0.0100	0400.0	0.0500	0.0075	<0.0050	<0.0050	<0.0050	0.0038	<0.0050	0.0038	0.0038	0.0038	<0.0050	0.0013	0.0025	0.0725	<0.0050	0.0075	0.0025	0.0025	0.0050	0.0050	0.0100	0.0075	0.0050	0.0063	0.0013	0.0288	0.0025	0.0062	
Zn	4.813																														
Ni	1.400	4.3(7	0.975	<0.075	<0.075	1.250	7.600	1.076	0.301	1.276	<0.075	<0.075	1.700	3.464	2.550	0.975	1.300	4.752	5.500	<0.075	3.225	4.655	0.775	0.825	3.402	<0.075	0.600	0.450	1.675	1.075	
Mn	53.412	) 5 4 5 7	33.383	21.973	4.594	20.417	16.737	13.914	58.269	44.895	44.712	39.409	180.367	49.362	121.617	107.671	20.917	23.499	73.662	8.592	12.237	14.127	17.434	18.226	6.067	8.704	10.400	13.789	22.717	7.975	
Pp	1.537	0.(3(	1.459	<0.113	0.183	0.858	1.787	1.276	1.555	0.926	1.729	0.826	1.208	1.388	1.408	1.134	1.833	0.763	2.612	0.788	1.637	0.363	<0.013	1.559	1.763	1.276	1.175	1.351	1.933	1.300	
FF.	42.262	30.762	105,670	126.326	29.131	59.867	97.762	115.215	135.005	239.840	193.083	134.051	110.367	84.635	94.867	80.407	151.367	343.434	4738.262	54.290	84.262	138.401	43.647	69.651	70.048	52.626	36.850	51.401	3234.117	39.100	(Continued)
ηζ	0.975	1.125	2.151	1.176	0.575	1.775	2.300	1.326	2.482	1.827	1.579	1.102	0.575	2.801	4.450	3.727	2.475	1.751	1.250	1.201	1.300	1.451	0.500	2.376	3.002	2.276	1.250	1.376	2.425	0.150	
J.	1.400	<0.025 0.830	0.938	2.976	0.375	0.675	1.425	2.302	2.457	2.503	2.808	3.305	0.900	0.700	2.175	1.238	0.350	1.601	2.800	2.926	1.225	2.477	0.425	1.513	1.501	2.851	1.250	<0.025	1.250	1.475	
Cd	<0.0025	<0.00 0.0025 0.0203	0.0484	<0.0025	0.0308	0.1008	0.0600	0.0225	0.0752	0.0951	0.3183	0.5834	0.1558	0.0550	0.1258	0.1484	0.0983	0.0125	0.0375	<0.0025	0.0300	<0.0025	0.0175	<0.0025	<0.0025	<0.0025	<0.0025	<0.0025	0.3358	0.0400	
As	<0.025	<0.0≥ <0.0≥ <0.0>	<0.025 <0.025	<0.025	<0.025	<0.025	0.025	<0.025	<0.025	<0.025	<0.025	0.025	<0.025	0.025	<0.025	0.050	<0.025	0.100	0.825	<0.025	<0.025	<0.025	<0.025	<0.025	0.125	0.050	0.025	0.025	0.050	0.125	
Site	JV4 D	4 V > A	JV5 C	JV5 D	JV6 A	JV6 B	2 9Af	JV6 D	JV7 A	JV7 B	JV7 C	JV7 D	JVB A	JV8 B	JV3 C	JV8 D	JV9 A	JV9 B	JV9 C	g 6Vl	JV10 A	JV10 B	JV10 C	JV10 D	JV11 A	JV11 B	JIVII C	JV11 D	JV12 A	JV12 B	

Site	As	Cd	S.	20	۳. e	Pb	Mn	Ni	Zn	Hg
JV12 C	0.075	0.1151	2.226	0.150	78.889	1.651	12.781	1.276	15.020	0.0063
	< 0.025	0.0551	7.111	T00.0	13.764	1.26(	) (7.)	2.020	12.925	0.0038
	0.050	0.2408	<0.025	3.225	49.117	2.208	50.617	<0.075	11.300	<0.0050
	<0.025	0.1076	<0.025	1.426	77.928	2.402	749.94	1.176	18.581	0.0288
	0.275	0.1800	0.650	2.400	50.762	2.487	50.662	3.500	6.813	0.0125
	0.050	0.7158	0.913	4.950	72.867	3.258	49.367	1.000	20.300	0.0450
NO2 A	<0.025	0.0650	2.400	0.225	57.600	1.900	110.600	2.250	19.112	0.0163
	0.050	0.1333	0.125	3.500	40.867	3.733	61.117	2.150	11.550	0.0125
	<0.025	0.0476	2.579	1.227	72.459	1.152	70.956	2.278	15.711	0.0188
	<0.025	0.1552	2.454	3.405	60.441	6.234	138.558	3.130	16.738	0.0188
	<0.025	0.0376	3.482	1.829	93.387	1.353	164.429	<0.075	11.160	0.0013
	<0.025	0.0308	1.638	2.451	185.710	1.059	138.186	0.250	12,306	0.0400
	<0.025	0.1556	3.012	3.489	170.532	2.435	81.677	1.456	20.871	0.0163
	<0.025	0.0100	3.102	1.876	174.850	1.588	57.441	<0.075	7.241	0.0275
	<0.025	0.0501	1.977	3.253	51.889	1.814	75.163	3.566	4.142	0.0063
	0.100	0.0409	2.526	6.828	189.962	1.709	144.939	3.127	31.566	0.0100
	<0.025	<0.0025	3.927	3.402	72.799	1.888	43.684	4.852	4.890	0.0075
	<0.025	0.0525	1.376	11.181	105.903	1.026	88.894	0.525	22.224	0.0113
	<0.025	0.0558	2.725	5.125	48.867	1.658	64.867	3.200	21.300	0.0050
	<0.025	0.0275	3.075	3.300	91.350	3.200	176.350	1.850	19.062	0.1512
	<0.025	0.0601	3.003	3.078	53.654	0.726	92.693	0.025	16.454	0.0263
NO5 D	0.050	0.0934	006.0	3.302	199.426	2.234	340.037	2.426	28.314	<0.0050
	<0.025	<0.0025	1.551	2.976	240.69	0.838	189.507	4.002	7.091	0.0075
	<0.025	0.0358	1.475	2.650	40.117	1.433	169.117	1.675	24.050	0.0025
	<0.025	0.0708	1.100	2.375	52.867	1.908	179.117	1.825	10.125	0.0125
	<0.025	0.0559	1.176	2.628	138.255	2.361	128.745	1.802	2.177	0.0075
NO7 A	<0.025	0.0383	0.575	2.525	46.117	1.483	33.867	0.925	19.800	0.0075
	<0.025	0.0184	0.813	3.128	84.451	3.412	48.415	0.300	6.456	0.0075
	<0.025	0.0308	1.401	3.102	42.388	1.788	14.799	0.050	4.952	<0.0050
d Tok	<0.025	0.1050	3.075	2.225	41.012	2.462	24.162	4.925	6.238	0.0100
	<0.025	0.0075	0.975	2.001	84.642	1.551	107.654	0.375	19.072	0.0263
				-	(60:00:400)					

Hg	0.0100	0.0063	0.0113	0.0025	0.0075	0.0388	0.0013	0.0088	0.0050	0.0013	0500.03	0.0025	0.0775	0.0100	0.0050	0.0050	0.0050	0.0050	0.0275	0.0188	0.0150	0.0188	0.0613	0.0501	0.0413	0.0488	0.0050	0.0513	0.0238	0.0313	
Zn	6.063																													96.082	
Ni	<0.075	1.353	0.376	<0.075	<0.075	2.603	1.775	1.577	<0.075	0.800	1.777	1.825	0.900	0.775	3.225	2.026	3.416	3.125	1.275	1.725	1.525	2.525	1.051	3.806	1.251	0.826	0.701	1.477	1.280	5.629	
Mn	72.162	18.446	27.182	32.179	28.677	27.127	121.350	31.397	31.162	96.148	84.451	35.617	214.224	47.117	214.367	295.014	127.965	294.867	50.617	44.100	65.117	134.850	45.395	43.478	36.386	33.901	12.955	52.679	58.333	19.680	
Pb	2.562	1.554	2.508	2.514	2.139	3.053	2.425	2.679	1.837	1.901	2.386	2.283	2.435	1.633	1.933	2.560	2.440	2.058	0.858	1.075	1.158	1.025	1.527	3.317	1.251	1.903	1.009	3.105	3.489	4.632	
Fe	67.012	38.446	40.722	50.538	50.538	41.892	42.100	29.895	64.512	55.378	2248.866	86.117	189.211	106.117	284.117	145.940	115.453	264.117	49.867	970.100	61.867	3365.100	89.690	82.887	84.935	113.270	79.666	191.387	209.689	140.310	(Continued)
Cu	2.775	2.331	1.906	4.377	3.177	2.227	2.075	1.728	6.500	3.802	3.403	3.425	2.951	2.725	1.825	3.952	2.427	2.300	8.075	5.200	8.150	9.000	5.205	5.358	4.555	5.183	7.057	5.483	8.961	8.788	
Cr	3.350	3.509	3.786	3.402	4.177	4.304	4.025	2.904	3.950	4.727	3.166	1.425	0.713	0.813	1.200	0.600	0.926	2.425	1.363	3.700	2.125	5.425	2.853	2.379	1.852	1.252	1.226	1.127	3.941	3.630	
Cd	0.0175	0.0125	0.0777	0.0525	0.1201	0.0801	0.3650	0.2604	0.0300	0.1526	0.1710	0.1158	0.1609	0.0633	0.0183	0.0809	0.0876	0.0958	0.7283	0.0350	0.2158	0.0450	0.1076	0.0125	0.0350	0000.0	0.2586	0.2554	0.2912	0.4056	
As	<0.025	0.150	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	0.225	<0.025	0.050	<0.025	0.050	<0.025	<0.025	0.200	<0.025	0.100	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	0.075	0.200	0.151	0.050	
Site	NO8 B	NO8 D	NO9 A	NO9 B	NO9 C	0 60N	A OLON	NOIO B	NOTO C	MOIO D	MOLL A	NOIL B	NOII C	MOII D	NO12 A	NO12 B	NO12 C	NO12 D	NY1 A					NY2 B							

Site	As	Cd	t l	2	Fe	Pa	Mn	Ni	Zn	Hg
NY4 A	<0.025	0.1482	5.249	13.059	243.697		10.849	2.009	61.464	0.0138
NY4 B	<0.025	0.3675	7.350	7.975	1885.762	4.187	42.412	5.750	52.062	0.0150
NY4 C	<0.025	0.2375	3.975	9.675	173.012		19.687	5.975	69.312	0.0175
NY4 D	0.075	0.2758	6.770	8.074	479.037		35.958	1.379	27.771	0.0263
NY6 A	<0.025	0.0651	3.756	6.509	127.541		29.895	1.452	30.483	0.0163
NY6 B	<0.025	0.0800	3.575	5.850	2863.262		58.912	2.400	77.812	0.0150
NY6 C	<0.025	0.4825	4.475	6.875	258.262		62.412	6.125	33.812	0.0150
NY6 D	<0.025	0.0162	2.250	4.375	255.694		86.656	4.275	17.344	0.0237
NY7 A	<0.025	0.0434	1.351	5.528	52.143		27.630	1.726	32.316	0.0125
NY7 B	<0.025	0.1084	2.865	7.207	68.685		49.917	1.151	39.089	0.0425
NY7 C	<0.025	0.0934	2.689	8.929	57.395		50.142	1.126	49.325	0.0425
NY7 D	<0.025	<0.0025	2.528	3.253	949.64		20.145	1.351	29.967	0.0138
NY8 A	<0.025	0.0983	2.038	11.675	120.867		24.342	2.250	58.550	0.0175
NY8 B	<0.025	0.0900	3.900	12.825	134.512		23.612	6.275	71.062	0.0125
NY8 C	<0.025	0.0372	3.250	9.550	103.012		21.612	5.850	59.312	0.0125
NY8 D	<0.025	0.0651	3.529	7.207	1561.662		34.885	η·00η	99.537	0.0163
NY9 A	<0.025	0.0534	1.127	11.172	83.283		49.215	2.154	88.727	0.0125
NY9 B	<0.025	0.1101	2.578	7.758	57.407		17.743	2.302	123.061	0.0188
NY9 C	<0.025	0.1403	2.730	6.288	76.002		11.623	0.952	75.088	0.0188
ux9 d	<0.025	0.0926	4.079	8.759	101.864		15.803	0.751	114.677	0.0250
NY10 A	<0.025	0.0803	3.713	6.874	74.109		44.757	1.330	12.657	0.0213
NY10 B	<0.025	0.0125	2.951	3.802	83.054		969.19	<0.000	994.9	0.0175
NY10 C	<0.025	0.0679	3.645	3.972	29.764		969.19	<0.000	994.9	0.0138
UY10 D	<0.025	0.1525	2.850	3.900	40.262		34.412	4.025	8.813	0.0150
NY11 A	<0.025	0.0275	2.051	2.851	41.033		7.891	<0.000	10.393	0.0150
NY11 B	<0.025	<0.0025	1.775	2.425	39.262		8.812	<0.000	6.588	0.0175
NYLL C	<0.025	<0.0025	2.202	2.703	186.949		11.349	6.956	4.922	0.0150
Q TTAN	<0.025	<0.0025	2.400	2.925	53.012		25.162	<0.000	7.388	0.0200
NY12 A	<0.025	0.0333	1.100	3.150	150.117		56.867	3.125	15.300	0.0125
NY12 B	<0.025	0.0483	1.100	7.400	119.611		42.867	3.375	14.300	0.0150
NY12 C	0.050	0.0125	1.476	3.577	135.668		37.869	2.326	39.457	0.0263
NY12 D	<0.025	0.0483	1.750	4.550	120.617		92.367	1.925	23.050	0.0200

APPENDIX F: MEAN HEAVY METAL UPTAKE (µg m<sup>-2</sup>) OF SPARTINA ALTERNIFLORA IN NATURAL MARSHES

Mean Heavy Metal Uptake (µg m<sup>-2</sup>) of Spartina alterniflora in Natural Marshes

Zn	715 13,698 10.09	9,131	10,225	15,220	14,319	19,876	8,305	16,607	401,6	5,892	23,803	17,045	19,954	22,635	90,274	12,461	12,479	7,226	7,047	5,761	10,885	7,889	6,829	4,125	5,820	6,549	8,669	7,550	10,810	11,778	
Mn	1102 159,255 7	74,605	107,462	64,586	231,657	56,871	156,931	21,848	34,773	30,566	77,572	170,830	29,945	33,249	45,915	12,544	14,298	21,587	71,084	11,824	14,944	87,864	131,604	59,929	58,586	26,189	120,955	20,738	31,853	91,728	
Fe	862,78	60,465	51,146	50,611	58,179	186,599	80,523	45,720	43,560	63,756	70,161	104,249	63,680	68,425	63,409	71,445	175,063	18,563	27,975	28,179	55,759	38,586	53,564	21,828	30,384	19,867	38,656	41,562	102,058	305,044	
Cr	2513 1202	2203	708	103	733	2295	1782	2542	2032	2105	1902	1291	2316	1945	5668	1756	1442	790	1730	1207	916	1572	1961	433	758	1761	1762	1432	1460	4919	
	45.1 51.7																														
Site	BMI	BM2	BM3	BM4	BM5	BM6	BM7	BM8	BM9	BMTO	BM1.1	BM12	CCI	CCS	603	422	ccs	922	CCZ	822	622	0100	CCII	CC12	CC13	CCI	CC15	JVL	JV2	JV3	

Hg	42.02	2.18	1.61	3.76	2.02	9.18	11.62	3.16	8.71	11.70	10.77	3.78	11.35	7.09	3.59	8.97	99.6	1.95	27.17	2.47	13.33	37.71	19.69	11.53	7.66	10.23	10.64	20.46	49.9	18.74	10.20
Zn	23,782	22,164	6,454	4,579	32,819	10,560	14,972	17,168	5,349	11,388	6,851	8,020	5,843	11,416	7,668	9,681	7,099	6,177	17,756	3,987	36,452	21,093	35,468	32,212	15,310	12,954	53,903	106,151	3,480	7,858	12,274
Ni	2627	1728	199	1273	2097	2271	928	1085	7126	1703	129	1256	572	2214	893	1,71	533	7468	1011	2347	1225	1347	1131	2340	1759	777	3137	1664	535	1730	1518
Mn	39,129	16,829	23,490	64,841	707, 42	50,049	10,029	7,690	16,330	65,448	75,764	42,236	195,44	162,218	17,456	75,646	15,450	31,312	93,765	172,469	45,964	29,951	22,022	18,549	27,115	13,011	19,442	24,260	20,726	14,832	33,962
Pb	1503	1287	595	698	1281	1063	1289	1014	918	2068	814	681	240	1544	1344	1309	1423	1020	1781	1748	719	1545	2207	2311	5251	730	1026	1174	638	1047	618
Fe	150,093	101,770	89,607	53,568	966,128	92,911	49,253	436,306	22,736	41,185	81,929	51,505	20,893	629,19	30,989	44,319	24,314	22,245	361,910	148,130	792,682	71,159	97,472	525,236	287,182	19,705	453,850	85,882	25,398	90,441	72,931
Cu	2752	1683	849	1245	1448	1782	1803	485	930	1352	1265	2665	1044	2552	1521	1835	1531	1650	2447	1986	††L†	3864	4788	8909	2506	2101	7000	9875	1583	2989	2283
IJ	2104	1787	1494	009	1376	1586	1217	1165	117	1304	1625	1017	249	1283	833	2071	2115	1802	1026	1112	2350	1533	1679	3899	1516	879	2268	2846	1241	2317	773
g	36.8	53.2	169.4	65.9	38.3	12.5	<0.1	78.0	7.66	67.3	54.9	15.9	16.0	39.0	56.6	8.4	42.3	90.3	106.8	61.2	118.4	25.0	201.8	170.0	85.8	21.5	51.5	119.0	26.3	0.9	21.9
As	<0.1	4.1	5.2	9.9	164.6	<0.1	49.5	45.6	22.7	10.3	<0.1	17.1	3.4	<0.1	<0.1	22.7	<0.1	<0.1	43.5	55.1	32.7	<0.1	71.0	12.8	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	5.1
Site	JVŞ	376	JV7	JV8	601	JV10	JV11	JV12	NOT	NO2	NO3	7ON	NOS	90N	LON	N08	60N	NOTO	NOLL	NO12	LYN	NY2	NY3	ΝΧγ	NY6	NY7	NY8	6XN	NYLO	ILYN	NY12

APPENDIX G: HEAVY METAL CONTENT (µg g<sup>-1</sup>) OF CYPERUS SPECIES

Heavy Metal Content (µg g-1) of Cyperus Samples

1 1	0	0	0	0	~	0	0	0	0	0	0	- <b>+</b>	10	ın	_	0	ın	10		0	0	10	10	10	_	_	0	0	
Hg	0.200	0.0500	0.020	0.050	0.166	0.1000	0.160	0.100	0.050	0.010	0.200	0.165	0.007	0.007	<0.005	0.010	0.017	0.012	0.008	0.010	<0.005	0.007	0.007	0.007	0.008	<0.005	0.015	0.010	
Zn	42.567	122.667	69.767	122.667	103.222	65.917	79.533	104.933	53.667	70.567	63.889	38.012	36.935	43.167	43.776	48.333	59.687	32.167	31.312	43.253	16.475	27.931	35.488	27.708	41.263	31.448	40.833	47.021	
Ņ	<0.300	0.433	<0.300	3.433	<1.000	<0.250	<0.600	<0.600	1.633	<0.300	<1.000	2.540	4.085	3.258	2.164	7.467	2.410	3.058	0.687	2.513	0.993	1.309	2.964	1.360	0.392	2.636	2.267	0.971	
Mn	74.967	62.267	68.367	93.667	66.222	205.167	226.333	139.933	244.167	77.067	222.889	114.829	87.085	88.542	65.205	84.583	47.315	103.792	76.484	56.404	132.063	72.578	72.186	95.685	133.637	58.851	101.083	75.385	
Pb	6.967	7.767	5.067	7.167	11.889	18.167	13.533	16.533	7.667	12.767	23.889	6.558	8.671	6.967	3.926	3.733	2.968	3.442	3.572	4.075	23.359	3.018	3.023	2.520	2.028	3.896	15.833	4.702	
ъ	126,400	117.400	103,400	146.400	120.667	256,000	108.800	156.800	136.400	91.400	146.333	87.820	110.655	112.850	106,366	125.700	105.653	86.350	85.173	89.780	76.763	100.400	99.800	101.252	97.519	138.989	135.200	134.237	(Continued)
Ç	5.500	6.500	5.200	9.100	<0.333	0.250	<0.200	5.000	12,000	4.100	6.667	3.634	7.447	4.150	4.837	8.000	8.604	5.150	5.111	6.212	9.652	6.328	7.966	6.309	5.953	9.234	8.850	6.526	_
r.	<0.100	0.800	0.100	1.400	<0.333	4.750	2.200	1, 800	3.900	<0.100	<0.333	<b>*</b> Q	Д	Д	Д	Д	Д	А	4.329	Д	17.200	Д	<0.025	0.451	2.002	2.778	2.400	1.054	
Cd	0.0767	0.2367	0.0367	0.3267	0.1889	1.6417	0.1733	0.1933	0.0267	0.1367	0.3889	0.1370	0.5895	0.4317	0.4553	0.2483	0.3819	0.1867	0.1099	0.4000	0.0133	0.2293	0.2672	0.2020	0.1109	0.0817	0.4233	0.4100	
As	0.100	<0.100	<0.100	<0.100	<0.333	<0.250	<0.200	<0.200	<0.100	<0.100	<0.333	<0.025	<0.025	<0.025	<0.025	<0.050	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.025	<0.050	<0.050	
Site	DEL A	DEL B	DEI C	DET D	DEI E	DE2 A	DE2 B	DE2 C							DE3 E					DE† E		. ~		DE5 D		DE6 A			

D = sample deleted.

As	Cd	Cr	Cu	Fe	<u>ا</u>	Mn	Ni	Zn	Hg
<0.025	0.1192	1.150	6.400	87.100	4.567	25.792	2.058	24.417	0.0075
<0.025	0.2392	2.300	7.200	162.350	4.242	88.792	2.208	38.667	0.0075
<0.025	0.2095	1.903	8.963	106.510	4.924	22.776	2.287	74.528	<0.0050
<0.025	0.2367	1.800	8.575	108.350	5.942	23.742	2.558	69.167	0.0050
<0.050	0.2001	3.389	8,675	104.572	4.651	24.197	2.165	78.420	<0.0100
<0.025	0.1968	1.376	4.752	105,403	2.443	50.317	2.460	21.177	0.0025
<0.050	0.3973	1.41	7.273	82,020	6.751	19.276	1.582	65.993	0.0253
<0.025	0.2625	<0.025	7.322	79.338	5.960	66.993	0.786	91.692	0.0050
<0.025	0.4252	1.278	6.391	110.877	5.856	98.538	0.610	130.994	<0.0050
<0.025	0.2505	<0.025	7.617	98.140	5.396	87.020	0.687	103.486	<0.0050
<0.050	0.2190	<0.050	4.514	92.477	3.594	80.324	<0.150	111.668	<0.0100
<0.050	0.2140	1.555	5.065	117.553	5.851	113.424	<0.150	127.716	<0.0100
<0.025	0.0367	1.127	8.438	99.750	4.098	103.447	0.885	99.065	<0.0050
<0.025	0.1268	<0.025	2.102	56.907	2.144	69.111	0.459	45.212	<0.0050
<0.025	<0.0025	0.776	6.294	88.232	4.173	116.967	0.759	84.293	0.0100
<0.025	0.4192	1.325	4.900	88.100	3.867	138.792	1.983	88.417	<0.0050
<0.025	0.2133	1.310	3.980	97.834	8.279	127.750	1.998	55.835	0.0025
<0.050	1.7733	0.750	6.900	107.200	2.533	37.933	4.717	94.833	<0.0100
<0.025	0.4344	<0.025	7.929	95.648	3.018	107.846	3.735	56.445	0.0075
<0.025	1.3612	2.579	4.331	77.967	4.023	57.876	3.238	84.794	0.0050
<0.025	1.7801	0.901	4.204	82.282	2.135	58.141	4.621	93.427	<0.0050
<0.025	0.8429	<0.025	7.286	100.250	5.850	57.878	3.514	68.019	0.0200
<0.050	0.6112	3.056	13.978	325.326	15.055	228.632	11.197	69.765	0.0075
<0.025	0.3605	2.278	7.361	158.325	7.524	204.144	8.650	36.117	0.0038
<0.025	0.2182	2.232	9.554	185.895	6.231	211.723	8.538	41.186	0.0088
<0.025	0.6209	1.152	9.815	157.824	6.122	94.479	4.710	59.652	0.0013
<0.050	0.2483	5.869	8,368	218.485	7.959	197.456	8.631	36.573	<0.0100
<0.025	0.3303	1.552	8.433	139.227	5.843	149.737	6.244	34.097	<0.0050
<0.025	0.2452	1.301	8.959	121.209	5.993	133.971	7.020	36.349	0.0038
<0.050	0.2510	<0.050	10.542	117.144	7.907	155.296	10.417	25.176	<0.0100
<0.025	0.1601	1.751	6.453	152.414	5.865	288.202	5.865	34.830	<0.0050
			_	Continued)					
								(Sheet	Sheet 2 of 6)

رعها ليجهدون يخطط دينجه

HE	0500.0	<0.0050	<0.0100	<0.0100	0.0125	0.0131	<0.0100	0.0075	<0.0200	<0.0100	0.0067	0.0076	0.0125	0.0325	0.0225	0.0081	0.0125	0.0038	<0.0100	0.0038	0.0025	<0.0100	<0.0100	0.0176	<0.0100	0.0013	<0.0100	<0.0025	0.0150	0.0175	0.0138	
Zn	38.145	58.650	137.901	51.279	93.813	82.575	26.711	54.734	57.365	47.673	41.754	77.713	27.680	36.662	73.625	47.879	176.125	69.952	79.705	73.636	91.189	22.784	116.125	77.357	100.827	181.150	120.245	93.514	108.333	86.296	110.729	
Ni	4.865	2.717	12.199	7.297	4.785	5.605	8.601	3.131	3.156	6.532	3.673	8.443	0.827	5.881	6.975	2.421	6.275	2.643	2.377	3.691	12.549	7.779	4.475	9.303	5.536	5.211	4.980	0.189	8.567	6.718	14.109	
Mn	217.763	91.725	100.376	103.485	77.330	98.697	210.807	108.392	136.623	153.829	145.435	396.222	160.496	120.295	106.175	132.977	60.675	90.018	39.364	101.689	649.69	115.987	52.675	35.030	41.308	74.321	417.593	189.358	361.583	495.535	405.446	
윤	4.242	4.394	4.785	8.400	2.630	6.425	3.786	5.235	6.263	7.382	4.090	4.336	4.083	3.979	8.075	4.435	2.875	3.394	2.027	43.556	2.525	3.600	5.175	2.633	3.081	4.190	6.031	3.589	5.833	7.378	6.773	
Fe	156.586	213.157	175.526	115.020	110.897	140.382	190.246	119.414	101.553	152.327	143.926	85.370	139.955	150.826	131.675	169.504	132.675	152.643	83.258	111.949	147.741	120.519	73.675	74.398	69.314	66.715	417.593	204.471	245.200	548.423	572.446	Continued)
Cn	7.906	9.815	10.621	10.331	12.525	11.264	6.419	8.216	9.319	9.710	9.381	8.012	8.317	7.758	9.150	9.545	16.250	11.799	12.763	10.010	13.205	14.099	9.700	7.272	9.820	6.847	20.621	10.202	19.350	11.467	15.423	
Ç	3.040	2.328	1.002	0.301	3.507	2,429	<0.050	1.904	2.605	2.052	2.043	1.166	2.555	0.901	1.950	3.565	2.650	3.82	454.4	2.628	3.737	1.158	2.950	0.953	1.503	3.578	3.504	4.030	3.150	4.882	4.732	
Cd	0.1447	0.1152	0.2405	0.2156	0.0802	0.3600	0.1254	0.1603	0.4008	0.0951	0.2994	0.1369	0.1353	0.1602	1.1850	0.0951	0.3850	0.4183	0.2402	0.5556	0.2708	0.3273	0.3600	0.2056	0.0501	0.4990	0.4254	0.1385	0.6783	0.3697	0.4982	
As	<0.025	<0.025	<0.050	<0.050	<0.050	<0.025	<0.050	<0.050	<0.100	<0.050	<0.025	<0.050	<0.050	<0.050	<0.050	<0.025	<0.050	<0.025	<0.050	<0.025	<0.025	<0.050	<0.050	<0.050	<0.050	<0.025	<0.050	<0.125	<0.050	0.025	0.125	
Site	MC2 E	MC3 A	MC3 B	MC3 C	MC3 D	_	MC4 A	MC4 B	MC4 C	_										MC6 D			<b>.</b> .	_	_	MC7 E				MC8 D		

Site	As	Cd	j	S	Fe	Pb	Mn	Ni	Zn	Hg
MC9 A	0.025	0.3034	4.037	13.691	112.926	3.573	35.945	1.367	86.071	0.0063
MC9 B	<0.025	0.3681	2.203	15.248	181.360	6.222	51.164	2.717	112.431	<0.0050
MC9 C	<0.025	0.4602	1.776	14.607	106.141	4.015	34.355	1.463	91.108	<0.0050
MC9 D	<0.050	0.4910	1.653	15.982	734.644	85.195	59.294	2,931	Д	<0.0100
MC9 E	0.025	0.2976	2.117	13.900	155.254	3.558	47.399	3.292	100.039	<0.0050
MC10 A	<0.025	0.0350	0.876	6.957	219.557	4.417	236.324	3.716	70.383	<0.0050
MCIO B	0.150	0.1168	2.653	10.961	217.317	3.846	127.669	6.014	59.977	0.0075
MCIO C	<0.025	0.0150	0.826	8.863	141.800	5.245	207.899	1.965	44.880	0.0088
MC10 D	<0.025	0.1101	0.025	12.262	157.495	4.042	116.704	4.867	33.846	0.0138
MCIO E	<0.025	0.1226	0.626	10.235	228.066	2.640	115.953	3.291	48.861	0.0088
MC11 A	0.325	0.1825	4.200	7.150	839.087	2.862	172.087	5.687	51.562	<0.0050
MC11 B	<0.142	0.6000	0.286	14.429	764.786	2.071	207.643	1.643	87.500	<0.0285
MC11 C	0.669	0.2285	6.522	9.197	694.649	<0.083	101.282	2.230	117.503	<0.0333
MC11 D	<0.111	0.3460	<0.111	18.973	1893.248	796.4	145.480	4.074	156.529	<0.0222
MC11 E	001.0	0.4600	<0.010	8.100	498.350	2.350	125.350	<0.300	169.250	<0.0200
MC12 A	<0.142	1.3150	7.370	23.988	55.419	10.188	29.63	<0.428	314.835	<0.0285
MC12 B	<0.100	1.5400	<0.100	13.700	69.350	10.450	38.150	7.850	177.250	<0.0200
MC12 C	0.250	1.2813	<0.050	11.211	66.742	2.528	18.243	1.527	105.731	<0.0100
MC12 D	<0 <b>.</b> 066	2.5844	5.133	11.800	76.267	6.578	28.578	5.022	237.111	<0.0133
MC12 E	<0.025	3.2749	2.828	16.917	118.719	2.044	37.829	3.011	206.123	<0.0050
MCl3 A	<0.111	0.0337	<0.111	16.854	74.551	13.539	136.348	0.955	141.854	<0.0222
MC13 B	<0.050	0.2302	<0.050	14.314	87.262	18.493	107.783	11.887	70.195	<0.0100
MC13 C	<0.133	0.3467	<0.133	18,800	81,800	3.800	175.133	1.667	100.333	0.1533
MC13 D	0.175	0.5594	1.726	9.880	73.387	3.118	54.069	3.010	82.208	0.0050
MC13 E	0.150	0.4532	1.252	11.242	157.574	4.745	184.114	5.066	71.920	<0.0050
ME1 A	<0.050	0.1035	<0.050	2.655	77.355	1.987	550.586	4.025		0.0100
	<0.100	0.2075	<0.100	4.217	119.880	6.191	319.444	5.644		0.0201
	0.351	0.0050	<0.050	6.012	189.554	4.434	471.618	0.777		0.0125
	0.250	<0.0050	4.104	900.9	124.299	3.829	357.032	<0.150		0.0125
ME1 E	<0.025	<0.1596	5.915	5.840	133.434	1.345	443.400	2.590	130.744	0.0175
	<0.100	<0.0100	9.018	6.313	95.541	2.555	596.142	<0.300		<0.0200
				)	Continued)					

Hg	<0.0100	0.0429	<0.0100	<0.0213	0.0302	0.0050	<0.0200	0.0100	0.0075	0.0038	<0.0133	0.0038	0.0025	0.0100	<0.0200	0.0113	<0.0100	<0.0100	0.0113	0.0163	0.0025	0.0351	<0.0050	<0.0200	<0.0100	0.0209	0.0051	0.0350	<0.0100
Zn	80.286																							269	917	045	233	84.867	584
Ni	3.783	2.381	3.070	6.709	11.301	2.750	8.333	5.267	0.626	1.139	3.622	1.768	2.025	1,004	1.570	1.614	2.175	2.021	0.513	2.689	1.008	3.975	2.636	5.756	2.269	7.864	3.455	5.1:95	<0.300
Mn	207.089	380.833	189.273	524.541	311.033	438.350	213.167	178.583	272.291	239.327	102.111	152.990	181.175	197.938	248.580	282.620	396.670	231.045	194.185	272.474	158.792	239.061	206.248	291.332	161.245	238.170	107.263	174.919	147.942
Pb	4.634	2.976	2.736	<0.053	2.985	2.950	4.967	1.033	1.779	2.290	4.311	3.716	2.725	4.325	2.199	3.766	2.092	1.737	3.764	4.765	3.367	6.947	5.948	1.071	3.887	1.427	2.713	3.560	20.331
ъ Б	48.272	131,000	210.911	203.814	84.909	175.350	155.400	77.700	145.967	119.207	66.933	125.963	102.675	123.848	119.940	114.452	103.614	170.541	103.389	132.404	106.100	205.605	118.969	83.735	249.950	49.7 <sup>1</sup>	260.566	185.014	51.556
Çn	5.311	6.143	5.255	18.144	26.761	8,500	8.900	3,800	5.461	4.179	3.533	2.853	4.900	3,307	2.906	4.304	2.108	3.357	7.054	5.928	4.225	13.727	6.982	<0.100	5.455	000.0	5.713	6.203	1.506
Cr	3.457	0.714	5.756	2.754	3.823	4.200	10,200	6.650	5.906	3.128	8.000	2.603	3.800	2.480	2.455	3.178	2.861	2.756	4.777	2.726	1.875	4.860	2.052	7.329	4.755	5.532	1.011	4.277	4.920
Cd	0.0351	<0.0071	0.3387	0.0706	0.1677	0.0500	0.2467	0.2833	0.2355	0.1702	0.1178	0.0776	<0.0050	0.2096	0.0785	0.1627	0.1339	0.1893	0.1826	0.1326	1.0817	1.5865	0.4796	0.2276	0.4538	0.0696	0.7162	0.2584	<0.0050
As	<0.050	<0.071	<0.050	<0.106	101.0	<0.100	<0.100	<0.050	<0.050	<0.025	<0.066	<0.025	<0.050	<0.025	<0.050	<0.025	0.050	<0.050	<0.025	0.100	<0.025	<0.050	<0.025	<0.160	<0.050	<0.100	<0.050	<0.050	<0.050
Site	ME2 B	ME2 D	ME2 E	ME3 A	ME3 B	ME3 C	ME3 D	ME3 E	ME4 A	ME4 B	ME4 C	ME4 D	ME4 E	MES A	ME5 B	ME5 C	ME5 D	ME5 E	ME6 A	ME6 B	ME6 C	ME6 D	ME6 E			IMI C		MW1 E	MW2 A

Site	As	Cd	r.	2	Fe	Pb	Mn	Ni	Zn	Hg
MW2 B	0.267	0.3449	0.067	0.935	332.043	6.854	263,129	1.090	15,131	0.0134
MW2 C	0.334	0.4292	<0.166	2.007	127.759	<0.083	96.767	<0.500	13, 103	7910 0
MW2 D	<0.100	0.3867	<0.100	<0.100	383,400	2.367	339,167	00000	20.067	0000
MW2 E	<0.050	0.1301	2.052	900.9	57.232	3.979	240.916	1.977	92.217	0.0075

In accordance with letter from DAEN-RDC, DAEN-ASI dated 22 July 1977, Subject: Facsimile Catalog Cards for Laboratory Technical Publications, a facsimile catalog card in Library of Congress MARC format is reproduced below.

Field survey of heavy metal uptake by naturally occurring saltwater and freshwater marsh plants: final report / by John W. Simmers ... [et al.] (Environmental Laboratory, U.S. Army Engineer Waterways Experiment Station.) -- Vicksburg, Miss.: The Station, 1981.
16lp. in various pagings: ill.; 27 cm. -- (Technical report / U.S. Army Engineer Waterways Experiment Station; EL-81-5.)
Cover title.
"June 1981."

"Prepared for Office, Chief of Engineers, U.S. Army, under Dredging Operations Technical Support Program."

"Available from National Technical Information Service, Springfield, Va. 22161.

Bibliography: p.63-64.

1. Great Lakes. 2. Marshes. 3. Plants. 4. Plants, Effect of heavy metals on. 5. Sedimentation and deposition. I. Simmers, John W. II. United States. Army. Corps of Engineers, Office of the Chief of

Field survey of heavy metal uptake by naturally occurring saltwater and freshwater marsh plants: ... 1981.

(Card 2)

Engineers. III. United States. Army Engineer Waterways Experiment Station. Environmental Laboratory. V. Title VI. Series: Technical report (United States. Army Engineer Waterways Experiment Station); EL-81-5. TA7.W34 no.EL-81-5

